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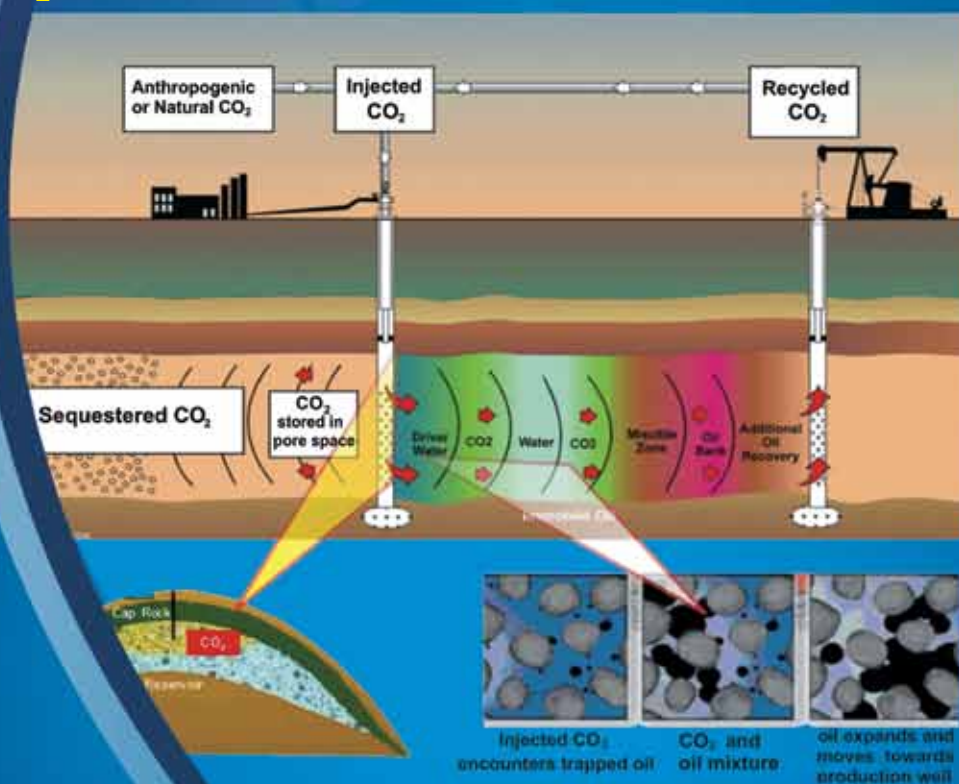
ISSN Number 0971 - 9709

The Journal of Indian Geophysical Union

JIGU Special Volume-1/ 2016

**SPECIAL ISSUE
on**

CO₂ Injection for EOR & Geological Sequestration



Guest Editors

Professor V.P. Dimri
Dr. Nimisha Vedanti



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Preface

We are pleased to present a special issue of the Journal of the Indian Geophysical Union entitled "CO₂ injection for EOR & Geological Sequestration". This issue is based on the papers presented at the Indo-Norwegian conference on "CO₂ Injection for EOR & Geological Sequestration" held during November 26-28, 2013 at CSIR-National Geophysical Research Institute, Hyderabad, India. The conference was jointly organized by the CSIR-NGRI, Hyderabad and the SINTEF Petroleum Research, Trondheim, Norway under the aegis of the Norwegian Embassy in India.

CO₂ injection for EOR has received a great attention due to multifold benefits such as enhancing oil recovery, possibility of CO₂ sequestration in the hydrocarbon reservoir and high tax savings in some countries by earning carbon credits. In this method, supercritical CO₂ is injected into a reservoir through injection wells ideally at a pressure equal to or above the minimum miscibility pressure (MMP), so that the CO₂ and oil mix together to form a liquid that flows easily to the production well. CO₂ can also be injected at a pressure below the MMP to swell the oil and make it mobile, this method is known as immiscible injection.

Global data shows that the CO₂-EOR techniques have a potential to increase the yield of depleting or heavy oil fields, but in India this method has not been applied to any oil field for commercial production because of several operational reasons. Laboratory studies carried out by the Oil & Natural Gas Corporation Limited (ONGC Ltd.), India, indicated that the recovery from depleting Ankleshwar oil field situated in Cambay basin of India could be increased by immiscible CO₂ injection as miscible injection required high MMP. Thus, a research project was formulated by CSIR-NGRI in collaboration with the SINTEF Petroleum Research, Trondheim, Norway to

estimate the CO₂-EOR potential of the Ankleshwar field with due support from the ONGC Ltd., India. The project was sponsored by the Norwegian Embassy, India and the Council of Scientific and Industrial Research (CSIR), India. Major outcomes of this project were two PhDs; one from Academy of Scientific and Industrial Research (AcSIR) (Ganguli, 2016) and other from Osmania University (Vadapalli, 2016 (a)), a text book (Dimri et al., 2012) published by Elsevier and many papers in scientific journals (Vadapalli et al., 2014; Ganguli et al., 2016; Pandey et al., 2016; Vadapalli, 2016 (b)).

The conference was organized by the project team to disseminate the knowledge gained in this research project. Total 32 papers from industry and academia were presented during the conference; however, 10 full manuscripts, mainly from academia were submitted for the publication. These 10 manuscripts were initially reviewed by the conference organizing committee and later by external experts, hence it took a little longer than expected to bring out the special issue. In this regard, we sincerely appreciate the authors for contributing their work to this special issue and for patiently revising the manuscripts to meet the suggestions of the referees. We also thank our learned reviewers for their valuable time.

Since the theme of the conference was based on CO₂ injection, most of the papers published in this volume are related to the technical aspect of CO₂ sequestration and CO₂ Enhanced Oil Recovery; however, one paper is based on the policy related aspects of Carbon Capture and Storage (CCS). The topics covered in this special issue include basic concepts of CO₂ injection, injection related micro-seismicity, CO₂ sequestration in Deccan volcanics, study of leakage risks associated with the CO₂ sequestration and estimation of uncertainty in the migration of the CO₂ plume in the reservoir. We

have made an attempt to bring together some of the insightful papers, covering most of the aspects of the CO₂ injection.

During the United Nations Climate Change Conference held in Paris in 2015, Government of India committed to reduce green-house gas emissions to 33-35% per cent of 2005 levels by 2030. This implies that in near future, there is ample scope of research in this field. We hope that this special issue of the Journal of Indian Geophysical Union entitled "CO₂ Injection for EOR & Geological Sequestration" will make a good reference material on this topic and be of great use for Geophysicists.

We acknowledge the Norwegian Embassy in India and our parent organization CSIR for financial support, ONGC Ltd. for data, our project team that includes Idar Akervoll, Per Bergmo, Szczepan Polak, Ravi Prakash Srivastava, Biswajit Mandal, Ranjana Ghosh, Pallavi Banerjee, Uma Vadapalli and Shib Ganguli, for their scientific contributions and our support staff P. Nagarjuna, Deepak Kumar and Ahmad for their assistance. Our special thanks are due to Dr. P.R. Reddy, who critically reviewed the final version of the manuscripts. We also thank Prof. Reddy (CEFL), Dr. MRKP Rao and Dr. O.P. Pandey for their suggestions. We are grateful to the then director Professor Mrinal K. Sen for his support.

**V.P. Dimri,
Nimisha Vedanti**

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Carbon Dioxide Sequestration During Enhanced Oil Recovery: Operational and Economical Aspects

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ABSTRACT

Rising oil prices coupled with continuously declining conventional oil production has led to significant use of Enhanced Oil Recovery (EOR) methods. Extensive laboratory studies, pilot projects and real field evaluations of the potential EOR methods have proved that the upcoming carbon dioxide flooding is one of the most promising recovery methods, which is being used on a large scale in many countries including the U.S., Canada and U.A.E. Carbon dioxide injection is being done in different modes: in miscible, near miscible or immiscible forms after water flooding as an EOR method, as a secondary displacement method, for producing well stimulation and for carbonated water flooding. During the CO₂ sequestration, CO₂ is retained in the reservoir by increasing its physical trapping or solubility in reservoir fluids, whereas during CO₂ - EOR process, oil recovery is maximized with a minimum amount of retention of CO₂ in the reservoir.

This paper presents a comprehensive review on the current status of CO₂ flooding processes with a critical analysis of latest CO₂ flooding projects featuring recent developments and operational problems and their solutions associated with those processes.

INTRODUCTION

The use of carbon dioxide as an oil-displacing agent has been investigated for many years and used for recovering oil from hydrocarbon reservoirs during 1920-1930 (Mangalsingh and Jagai, 1996). Extensive laboratory studies, pilot projects and field studies have further confirmed the efficacy of the technique in enhanced recovery of oil (Holm and Josendal, 1974). In 1945, Poettmann and Katz discussed the phase behavior of CO₂ and paraffin systems. Their studies indicated that for heavier crude, the increase in oil volume is normally between 10-22% and the viscosity reduction is up to less than 0.1 of the original value. The high solubility of CO₂ in oil and resultant reduction in viscosity attracted the investigators to study this process in laboratory. The investigations included fluid injection processes using continuous CO₂ or a single slug of CO₂ followed by water or carbonated water (Holm, 1987). The first successful process was carried out in 1958 at Oklahoma (Khatib et al., 1981). In the 1950's and 60's, the application of CO₂ was primarily as a miscible displacing fluid and certain limitations in the process were realized. These limitations led to concerted investigations on CO₂ for immiscible displacement. Field applications of immiscible CO₂ were carried out in the 1970s. Due to certain deficiencies in continuous CO₂ process, the water alternating gas (WAG) process was introduced in the 1980s. A comprehensive review of the immiscible process in heavy and light oil reservoirs was done by earlier researchers (Khatib et al., 1981). Recently, enhanced oil recovery using carbon dioxide has become one of the most popular investigations among the researchers and the oil producers.

MOST COMMON TYPES OF CO₂ FLOODING

Elaborated studies of phase behavior and CO₂ flood performance in laboratories and fields have brought into light certain mechanisms by which CO₂ displaces oil. These mechanisms in general are related to the phase behavior of CO₂-crude oil mixes.

These are: (1) immiscible CO₂ drive, (2) hydrocarbon-CO₂ miscible drive, (3) hydrocarbon vaporization, (4) direct miscible CO₂ drive and, (5) multiple contact dynamic miscible drive.

HOT CO₂ FLOODING

A new CO₂ - EOR technique "Hot CO₂ Flooding" is being worked out these days in Bati Raman heavy oil field in Turkey. This technique (Picha, 2007; Issever et al., 1993) is a combination of thermal and solvent techniques, wherein the hot CO₂ heats the reservoir and reduces oil viscosity. This enables partial miscibility with the oil, thus leading to improvement in oil mobility and recovery factor. The operating temperature of hot CO₂ is decided according to the reservoir temperature and oil property and is kept above the critical point in CO₂ phase diagram. Hot CO₂ can similarly be injected in different ways such as continuous injection, alternating with water injection or followed by steam injection. According to the theoretical calculations, the ultimate recovery increased from 10% in case of conventional CO₂ flooding to 13% in case of hot CO₂ flooding, due to marginal reduction in oil viscosity and increase in swelling factor.

WATER-ALTERNATING-GAS (WAG) PROCESS

Water-alternating-gas (WAG) process is a combination of two traditional techniques of improved hydrocarbon recovery: water flooding and gas injection, which can be defined as a short-cycle alteration of definite ratios of water and CO₂. The first field application of WAG is attributed to the North Pembina field in Alberta, Canada, by Mobil in 1957. Laboratory models (Rogers and Grigg, 2001) developed early in the history of flooding showed that simultaneous water/gas injection had sweep efficiency as high as 90% compared to 60% for gas alone. The displacement mechanism caused by the WAG process occurs in a three phase regime; the cyclic nature of the process creates a combination of imbibition and drainage (Christensen et al., 2001). Optimum conditions of oil displacement are achieved if the gas and water have equal velocity in the reservoir. Important technical factors affecting WAG performance that have been identified are heterogeneity, wettability, fluid properties, miscibility conditions, injection techniques, WAG parameters, physical dispersion, and flow geometry (Surguchev et al., 1992).

OTHER METHODS

Another method was introduced where mobility control could be achieved by adding one or more of certain surface active chemicals to carbon dioxide, which is either in a liquid or supercritical state, before it is injected into wells (Heller et al., 1983). The surface active chemicals have sufficient solubility in dense carbon dioxide to cause water that is already present in the oil reservoir to emulsify with the carbon dioxide. The emulsion decreases the mobility of the injected carbon dioxide. Upon continued injection of carbon dioxide, flow is diverted to other less permeable zones of the reservoir or away from the top of the reservoir thereby improving the vertical conformance of the carbon dioxide flood and substantially increasing the oil recovered by the flooding process. In some cases carbon dioxide flooding is carried out with a premixed transition zone of carbon dioxide and crude oil components. This method consists of injecting into the formation a premixed transition zone slug comprising carbon dioxide and said crude oil components at said critical concentrations, formation temperature and said selected pressure; and injecting a drive fluid into the formation after the premixed transition zone slug (Heller et al., 1982).

In order to identify a potential reservoir for CO₂ flooding, a screening criteria based on technical as well as economic factors may be helpful. The reservoirs are screened for economic feasibility according to capital costs, operation expenses, availability of CO₂ and then ranked based on the economics, namely ratio of derived revenue based on day to day international market rates

and production costs. Prior to proceeding with the pilot production operations, appropriate coring, core analyses, and logging operations were carried out to evaluate certain parameters, which were compared with the performance of a pilot field test.

LATEST CO₂ FLOODING PROJECTS

It has been reported that the total number of CO₂ projects in operation so far has increased in the U. S. and worldwide (Moritis, 1996). Recently, the promising results of two gas injection pilot initiatives of a full field non-miscible gas injection schemes, implemented in a carbonate reservoir in Abu Dhabi, U. A. E., have been reported (Shedid et al., 2007). The study stressed the importance of a detailed reservoir characterization and maximum flexibility in the development design and close monitoring of the reservoir response. Of the four immiscible CO₂ pilot floods conducted in Trinidad from 1973 to 2003 (Mohammed-Singh and Singhal, 2004), one project with a large throughput of CO₂, indicates incremental oil recovery between 2-8% of the OOIP with predicted ultimate recoveries of 4-9% of the original-oil-in-place (OOIP). The projects were conducted in a gravity stable mode after primary, secondary and tertiary production (after natural gas and water injection). Cumulative CO₂ utilizations improved with efficient production practices and ranged from 3-11 MCF/bbl till 2003. In this field, oil recovery improved as more production wells were added downstream of the injection wells. In some of the projects, production increase was observed even after discontinuation of CO₂ injection (supply interruptions) aided by water influx and gravity stabilization. It was concluded that immiscible gas injection projects should be implemented with special attention to GOR control at the producers, and also the gas injection rates should be selected such that oil contributions from CO₂ injection as well as water influx are optimized. Evaluation of the application of CO₂ injection into a Norwegian oil reservoir showed that this application produced 63 % of initial-oil-in-place (IOIP) while water injection recovered 43 % IOIP (Lindeberg and Holt, 1994). The increment in oil recovery by miscible flooding was attributed to improvement in displacement efficiency and reduction in oil viscosity. A review of twenty-five CO₂ projects as a maturing EOR process indicated that the application of CO₂ floods provided good increments in oil recovery (Grigg and Schechter, 1997). The study presented some concerns about high mobility of CO₂ and its early breakthrough. The study also indicated that most of the newer floods and some older floods increased CO₂ slug size, injected CO₂ continuously till CO₂ breakthrough, and then converted to water-alternating-gas (WAG) process. It has been observed that miscible CO₂ flooding might recover up to 8 % to 16 % of the OOIP depending on the

reservoir depth and other peculiarities of the reservoir and the crude oil characteristics. The study also indicated that maximum CO₂ flood recovery could be plus 10 % of OOIP after matured water flooding recovery (Jeschke et al., 2000). A case study of the Dulang field in Malaysia was carried out to evaluate the CO₂ injection. The study included phase behavior studies, vaporization test and core displacement tests. The results indicated that 15 % of the initial stock oil in place was produced using CO₂ flooding at the initial reservoir pressure of 1800 psig. The study also showed that the application was capable of extracting components heavier than heptanes (Zain et al., 2001). Using recombined fluid in a slim tube to determine the Minimum Miscibility Pressure (MMP) under different gas-oil ratio (GOR) and CO₂ concentrations proved that the main factors to enhance the oil recovery in CO₂ flooding were the oil swelling and the viscosity reduction (Yongmao et al., 2004). In the Neuquén Basin in west-central Argentina, with no prior CO₂ injection, viability and prospects of immiscible CO₂ flooding process in the Avile reservoir were found to be favorable with the results of core flood evaluation, compositional simulation and evaluation of facility requirements (Brush et al., 2000). This project was innovative in the sense that it assessed the feasibility of extracting previously vented CO₂ and injecting it again to recover additional oil and at the same time reduce green house gas emissions. The estimated increase in oil recovery was 4% of OOIP and the potential greenhouse gas emission reduction ranges from 185,000 to 714,000 carbon equivalent metric tons. Asghari et al., (2006) designed a model to predict the performance of CO₂ flooding in this field based on the data obtained from past water flooding performance and CO₂ rate injection by developing a correlation. It is estimated that implementation of CO₂ flood would extend the economical life of the field more than 25 years, with an incremental recovery prediction of 13% to 19%. The limitation in applying this correlation is that it can be applied only to existing CO₂ flood area, since the two correlations developed require some post-CO₂ flood data to develop constants. These correlations provide a very fast and practical method for determining the expansion area of an existing CO₂ flood operation. A recent example of the significance of this process can be seen in the Spraberry Trend area of West Texas (Montgomery et al., 2000), once known as the "largest uneconomical field in the world," which contains as much as 10 billion bbl of original oil in place. Despite five decades of production, including several large-scale water flood projects, recovery from the Spraberry rarely exceeded 8-12%. A significant new effort has been launched to correct this situation and to evaluate the efficiency and economics of using CO₂ flooding to enhance recovery from Spraberry reservoirs. Laboratory study of

CO₂ gravity drainage in Spraberry core suggests that significant additional oil can be recovered in this manner. Recently, a new approach (Liu et al., 2004) of studying the progress of carbon dioxide in reservoirs using ultrasound has been introduced. This may lead to unique monitoring the position of carbon dioxide, as well as visualizing the reality of the oil producing area. The problem of identifying the position and location of the immiscible or miscible layer is a critical issue in the method of carbon dioxide injection for oil producing bodies. Accurate knowledge of the position could lead to an ability to control the position of the carbon dioxide front affecting production, efficiency, economy, and in the end profitability. Some measures of digital signal processing are introduced to obtain position information of interfaces in porous medium and improve the resolution and accuracy of the location. Ultrasound has many advantages as a tool in underground probing, such as small size, low power consumption and safety.

CO₂ - EOR has been used by the oil and gas industry for over 40 years, but only recently has its potential as a carbon sequestration method been realized and investigated.

Currently, CO₂ - EOR comprises approximately more than 37 percent of all EOR being performed in the United States and it has been a leader in developing and using technologies for CO₂ - EOR by performing approximately 96 percent of worldwide CO₂ - EOR.

OPERATIONAL PROBLEMS IN THE FIELD

CORROSION

Carbon dioxide in the presence of water forms carbonic acid, which produces corrosive environment. The corrosion can be minimized by having a separate injection facility for both these fluids, by using corrosion inhibitors and by using stainless steel wellheads and downhole equipments. The carbon dioxide should also be dehydrated at the source before it is compressed and transported (IPCC, 2005).

ASPHALTENE DEPOSITION IN CO₂ FLOODS

Asphaltenes are believed to exist as colloidal suspension in oil, stabilized by resins present and are in equilibrium with oil at reservoir condition. This equilibrium gets disturbed during production mainly due to pressure and temperature reductions and introduction of miscible gases and solvents, thus leading to loss in production and failures of Electrical Submersible Pumps (ESP), safety valves and other downhole equipment. In general, as more gas dissolves into the crude oil, the asphaltenes problem increases (Tin and Yen, 2000). In order to curb this problem several chemical inhibitors were introduced.

HANDLING OF THE PRODUCED CARBON DIOXIDE

The best solution to this is re-injection of CO₂ in the reservoirs to reduce the volume of CO₂ to be procured (Necmettin, 1991). It can be injected without processing or after separating it from associated gases. Finally all such proposals may be considered only after the economics of the processes are studied thoroughly.

SCIENCE AND ENGINEERING REQUIREMENTS FOR GEOLOGICAL CO₂ SEQUESTRATION

CO₂ sequestration in geological formations requires proper site selection, effective monitoring, and remediation options should a CO₂ release occur. Much information about subsurface reservoirs has been obtained from oil and gas fields, natural CO₂ reservoirs, and subsurface storage facilities for natural gas and other fluids. The knowledge from these geologic storage examples is extremely valuable (Friedmann, 2007). However, they need to be complemented with additional information, specific to CO₂ properties, to ensure the selection and monitoring of a safe storage site.

Wells for CO₂ sequestration need more demanding specifications

1. Temperature and pressure may be higher under the bottom hole conditions
2. Routine maintenance and repair or replacement of the well equipment may be less acceptable, so, higher integrity in designs expected
3. Well service lives may be longer
4. Leak tolerance during well servicing should be minimum;
5. Integrity should be maintained in every stage of well design and construction, operation, monitoring, work over, suspension and abandonment.

Some areas for additional studies are

1. Geochemical and Petrophysical Studies of the CO₂-Fluid-Rock System
2. Geomechanical Aspects of Injections
3. Basin-Scale Modeling of CO₂ Distribution
4. Isotropic Tagging of Injected CO₂
5. Geophysical methods will play a key role in the monitoring. Developing sensitive and cost effective methods for long-term monitoring is an important part of sequestration.

In Salah Project in Algeria and the Sleipner Project in Norway, the CO₂ that was removed from a natural gas sales stream was being injected into saline aquifers, rather than being vented to the atmosphere. CO₂ enhanced oil recovery in Weyburn project in Canada captured CO₂

from a gasification plant. The West Texas Permian Basin projects relied upon naturally occurring CO₂ from reservoirs (Herzog, 2010; NETL, 2014).

ECONOMIC DECISIONS DEPEND PRIMARILY ON

- i. Price of oil
- ii. Cost of capital (interest rates) and capital infrastructure construction (drilling, gas processing, pipelines)
- iii. Cost of carbon emission taxes, or conversely, the value of carbon sequestration credits
- iv. Cost of carbon dioxide capture from anthropogenic sources
- v. Pilot project results and
- vi. Speed of technology advancement and dissemination

PROPOSED PETROLEUM R&D PROGRAM

- i. To Evaluate and enhance carbon dioxide flooding through sweep improvement.
- ii. Improve CO₂ flooding sweep using CO₂ gels.
- iii. Conduct CO₂ injection tests to improve the reliability of computer simulations of oil fields from CO₂-EOR and calculations of sequestration capacity.
- iv. Determine the economic and technical feasibility of using CO₂ miscible flooding to recover oil in a selected oilfield.
- v. Employ molecular modeling and experiments to design inexpensive, environmentally benign, CO₂-soluble compounds that can decrease the mobility of CO₂ at reservoir conditions.
- vi. Develop a neural network model for CO₂ - EOR.
- vii. Develop a novel, low cost method to install geophones for CO₂ monitoring.

MARKET AND PRICE FOR CO₂

CO₂ - EOR production is linked to the price of oil; and rising oil prices have increased the demand for CO₂. The price of CO₂, strongly influenced by regional constraints in supplying CO₂, also increased with rising demand during this period. In the United States alone, the oil and gas industry operates over 13,000 CO₂ - EOR wells, over 3,500 miles of high pressure CO₂ pipelines, has injected over 600 million tons of CO₂ (11 trillion standard cubic feet) and produces about 245,000 barrels of oil per day from CO₂ - EOR projects (Hargrove, 2006).

According to The Energy Information Administration (EIA) (2012) and International Energy Agency (IEA) (2011a), as oil prices will continue to increase over the next decade there will be increasing demand for CO₂ leading to increased CO₂ supplies. In the US the supply of CO₂ is expected to increase by 50 per cent by 2015 relative to

2010 production levels, and could potentially double by 2020 (EIA 2012). More than half this growth will come from A-CO₂, which will become increasingly important during the following decade.

The average rate of use of CO₂ in the US is estimated to be 0.5 tonnes of CO₂/barrel of oil in 2011. This is an increase from 0.3–0.4 t CO₂/barrel of oil for some projects as described by earlier studies (Global CCS Institute, 2012). With increasing pipeline investments to relieve supply constraints, together with additional A-CO₂ supply sources being developed, it is expected that over the medium term CO₂ prices will be set by these low-cost anthropogenic sources.

ECONOMIC SUMMARY

Capex: 8% of oil price per barrel;

CO₂ cost: 2.5% of oil price per MCF;

Pay out: 5 years;

IRR: 20%

Cost of CO₂ injection, in general ranging from \$60 per ton to higher values based on 2008 data (Jahangiri and Zhang, 2011), depending upon capture process applied, volume of CO₂, distance from source to sink and some other site-specific characteristics. "Next generation" CO₂ - EOR technologies, primarily focused on increasing oil production, could create between 165 and 366 Gt of CO₂ storage capacity, while producing 705 to 1,576 billion barrels of incremental oil. Assuming emissions of 6.2 million metric tons/year over 40 years of operation per plant could result in storing the emissions associated with 2,200 to 4,900 one-GW size coal-fired power. This capacity is sufficient to store 18% to 40% of global energy-related CO₂ emissions from 2010 to 2035. "Next generation" CO₂ - EOR technology stores 14% to 18% more CO₂, and produces 47% to 50% more oil than "state-of-the-art" technology. Recent developments in the Permian Basin indicate that there may be vast, previously unrecognized opportunities for additional oil production from the application of CO₂-EOR, besides providing additional capacity for storing CO₂ (US DOE/NETL, 2010).

This potential exists in residual oil zones (ROZs) below the oil/water contact in traditional oil reservoirs that are widespread and rich in oil saturation. In addition to the traditional main pay portion of depleted oil fields, they represent a second potentially much larger CO₂ storage option. Field pilots have shown that applying CO₂ - EOR in ROZs appears to be commercially viable. This may result in a two-to-three fold increase in the potential storage capacity associated with the application of CO₂ - EOR.

Other approaches to increase CO₂ storage in conjunction with CO₂ - EOR may further increase storage capacities associated with such applications.

CCS ACTIVITIES IN VARIOUS COUNTRIES

Algeria

A CCS project is in operation (Wright, 2007).

Botswana

Initial assessments on role and opportunity for CCS, undertaking preliminary geologic assessment (Seiphemo, 2014).

Brazil

A Centre of Excellence in CCS R&D has been established. Completed a Geographic Information System (GIS)-based database of CO₂ sources and sinks. Pilot CO₂ injection program underway, reviewing and refining Brazilian Carbon Geological Sequestration Map (CARBMAP) program (Brazil, 2014).

China

CCS has been adopted as a key GHG mitigation technology in National Climate Change Program. Numerous domestic R&D initiatives and efforts are underway to assess and characterize CO₂ storage capacity by Chinese Geological Survey. Several pilot projects for CO₂ capture and CO₂ - EOR, 11 large-scale integrated demonstration projects in the planning stages. (Seligsohn et al., 2010).

Egypt

A study is underway assessing potential for CCS in gas processing and power industry, identifying barriers and environmental impacts (Egypt, 2013).

India

In India, Institute of Reservoir Studies, Oil & Natural Gas Corporation, Ahmedabad, has been carrying out laboratory simulation and pilot scale studies. Western region onland light oilfields have been chosen to be first candidate for the carbon dioxide flooding programme. Under Indo Norwegian collaboration program CSIR-NGRI and SINTEF Petroleum Limited, Norway has carried out reservoir modeling & simulation study for the CO₂ - EOR potential of declining Ankleswar Oil Field, with the support of ONGC. (Dimri et al., 2012).

Indonesia

A study is being finalized on potential for CCS as part of South East Asia CCS Scoping Study, including opportunities for deployment, regulatory and economic analysis. After the assessment of current CCS R&D activities and technical capacity of the domestic industry to provide support throughout the CCS chain, developing a CCS Technology Roadmap for preliminary studies on CCS and EOR (Takahashi, 2015).

Jordan

A study is underway to assess the potential for CCS in oil shale development strategy and to identify and address legal, regulatory, and financial barriers (World Bank. 2012).

Kenya

There is an investigating possibility for high-level storage study. Workshops and training on CCS technology are being conducted (Kenya, 2012).

Kosovo

A study is completed for preliminary geologic potential and capacity-building assessment including legal and regulatory requirements (Kulichenko, N. and Ereira, E., 2011).

Mexico

CCS identified in Special Program on Climate Change and National Energy Strategy 2012–26. Country-level preliminary assessment of CO₂ storage potential completed, pilot projects being considered, including for CO₂ capture with a focus on EOR (McCoy, 2014).

Maghreb

A study is underway to assess the potential for carbon capture on projected and existing power plants in Tunisia, Algeria and Morocco for CO₂ geologic storage and transportation at a regional scale (Kulichenko, N. and Ereira, E., 2011).

Malaysia

Several CCS workshops were conducted for raising awareness and discussing key issues. A study was completed on the long-term role for CCS, opportunities for near-term deployment, technical and financial feasibility, and next steps for further investigation. Capacity-building program developed and activities being implemented (Kulichenko, N. and Ereira, E., 2011).

Norway

Sleipner (Norway) is the world's first successful demonstration of CCS technology into a deep saline reservoir and remains the only development where the CO₂ is both captured and injected offshore. The captured CO₂ is compressed and injected (via one injector well on the Sleipner A platform) into the Utsira Formation, a sandstone reservoir 250 metres / 820 feet thick. The reservoir unit is at a depth of 800-1,100 metres / 2,625-3,610 feet below the seabed. The seal to the reservoir is provided by a 700 metre / 2,430 feet thick gas-tight caprock above the Utsira Formation (Kulichenko, N. and Ereira, E., 2011).

Philippines

A Study is being finalized on potential for CCS as part of South East Asia CCS Scoping Study, including opportunities

for deployment, regulatory and economic analysis (Asian Development Bank, 2013).

Saudi Arabia

Has identified CCS as an appropriate low emission technology, Workshops and roundtable discussions were held and sponsored on CCS, including on monitoring and storage specifically, challenges and opportunities. Working towards a EOR-CCS project (Lui et al., 2012).

South Africa

CCS identified as a priority in national White Paper on National Climate Change Response, South African Centre for CCS established, Storage Atlas completed, further basin-specific storage studies underway, scoping study for test injection project being developed. Legal and regulatory review undertaken and further work commenced (Beck et al., 2013).

Thailand

A study being finalized on potential for CCS as part of South East Asia CCS Scoping Study, including opportunities for deployment, regulatory and economic analysis (Asian Development Bank, 2013).

Trinidad and Tobago

CCS Scoping Study, including Legal and Regulatory Review (Kulichenko, N. and Ereira, E., 2011).

United Arab Emirates

Three industrial CCS projects in the planning stages (in the hydrogen, steel, and aluminum industries) (Havercroft, I. et al., 2011).

Vietnam

Study being finalized on potential for CCS as part of South East Asia CCS Scoping Study, including opportunities for deployment and regulatory and economic analysis (Asian Development Bank, 2013).

CONCLUSIONS

1. Carbon dioxide flooding is one of the most attractive processes for enhancing the recovery of oil, which can also capture part of the carbon dioxide and help in reducing greenhouse gas emission. India should apply this EOR process for its oilfields.
2. Besides laboratory injection studies, investigation of rock and fluid properties of the candidate reservoir with specific studies on wettability, dissolution, precipitation, particle invasion/migration, residual oil saturation on trapped gas saturation, contact time, mass transfer are to be carried out before taking up a pilot project.

3. Finally, the economic viability of any such project should be studied thoroughly before its field application.

ACKNOWLEDGEMENTS

Author expresses his thanks to anonymous reviewer for useful inputs in refining the manuscript.

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Monitoring Microseismicity in Oil Boreholes

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ABSTRACT

Micro seismic events or ultra-micro earthquakes ($M -3$ to 1.0) are generated from normal production and also by injection activities for early oil recovery (EOR) in oil boreholes. The recorded micro seismicity is associated with the stress and pore pressure changes in and around the reservoir and can be used to continuously monitor the reservoir stress field and related fluid flow dynamics in real time. These ultra-micro earthquakes, when detected with proper instrumentation, allow us to measure fluid movement away from the well locations. Field wide monitoring between the wells is important for optimum reservoir management, and it is achieved by mapping the hydrocarbon fluid pathways in a producing reservoir. Recording and analyzing the passive micro seismic events have the potential for defining the fluid flow patterns and or the reservoir flood fronts. Maps of fluid pathways aid in reservoir management by optimizing development plans and improving the ultimate oil recovery.

INTRODUCTION

Optimization of reservoir management requires the monitoring of fluid dynamics or drainage patterns of reservoir production and injection processes. Modelling studies in a large oil field of Saudi Arabia suggest that conventional monitoring technology like time lapse or 4D seismic technique may not be applicable every where. This is because the sensitivity of change in acoustic impedance or seismic signature in carbonate reservoirs is extremely low, below the detectability of 4D seismic measurements (Dasgupta, 2005). Passive monitoring of micro seismic tremors, has the potential for defining the fluid drainage patterns or the reservoir flood fronts that are invisible to active seismic techniques.

Reservoir rocks react to changes in stress and strain associated with pressure changes in the reservoir and generate microtremors ($M -3$ to 1.0). These tiny tremors or ultra-micro earthquakes are caused by slippage or tensile deformation on pre-existing fractures/faults. The ambient stress field is perturbed by fluid or CO_2 injection and oil extraction. The stress change induces shear slippage along the zones of weakness like fractures and faults. This shear slippage generates micro seismic events. Several successful experiments are made towards monitoring micro seismicity in geothermal boreholes (e.g. Dorbath et al., 2009), and also for EOR in oil boreholes (e.g. Dasgupta and Jervis, 2008; Rutledge et al., 2008); some results are presented here.

FIELD EXPERIMENTS

A network of sensors is made spatially on the surface and at different levels in the oil borehole. The sensors record the arrival times for compressional waves (P-waves) and shear waves (S-waves) of the micro seismic events (Fig.1). These data may be used to determine the hypocenter parameters

by multiple regression analysis (Kayal, 2008), and the epicentral and 3D hypocentral maps could be produced to understand the fluid flow at depth.

The experiment provides an opportunity for a field wide continuous monitoring in real-time as the fluids are produced or extracted from, and injected into the reservoir. Anisotropic fluid flow or directionally uneven flow rates are sometimes associated with reservoir production and injection operations. This information could also be applied in inferring the distribution of reservoir properties such as permeability related to fractures and faults and reservoir connectivity for numerical modelling.

FIELD EXAMPLES

Examples of an epicentral map and a 3D hypocentral map are illustrated here to understand the efficacy of the microearthquake technique in oil producing boreholes.

Rutledge et al., (2008) reported a successful experiment of micro seismicity monitoring in boreholes due to CO_2 enhanced oil recovery in the Utah oil field (Fig 2). The seismicity revealed two NE-SW striking fracture zones. No seismically active structure was found by the surface 3D seismic survey. Some 3800 micro earthquakes with moment magnitude $M_w -1.2$ to 0.8 were recorded within ~ 2 km of the treatment well during the first year of the monitoring; the seismicity was induced during the continuous CO_2 injection. Focal mechanism solutions of the events indicated strike slip faulting that indicates shear failure on micro fractures due to fluid flow dynamics. The frequency magnitude relation showed the b-value of the order of 2.0 , which is almost twice the normal value in a tectonically active zone. This higher b-value is characteristic of induced seismicity due to injection. There was no main shock to associate this large sequence of micro earthquakes as aftershocks. The micro seismicity

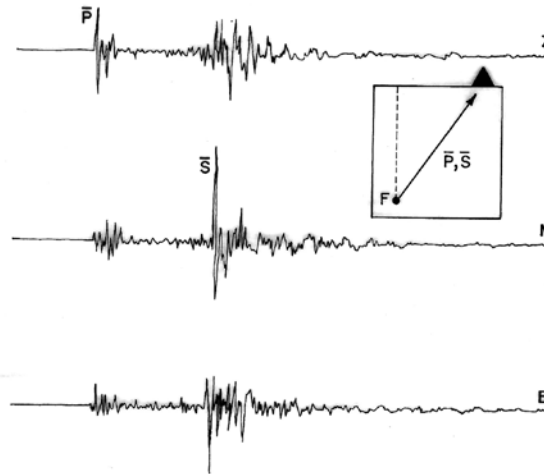


Figure 1. A record (seismogram) of a micro seismic event; Z represents the vertical component, N and E are the two horizontal components. Arrival of P wave is well identified in Z component, and the S-wave in horizontal components (after Kayal, 2008).

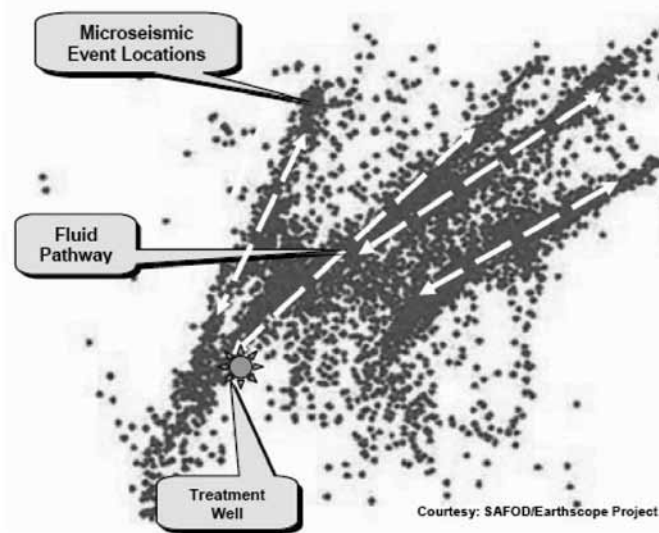


Figure 2. Epicentral map of ultra-microearthquakes ($M -1$ to 0) in an experiment of CO_2 injection in Utah oil well that were recorded within about 2 km (after Rutledge et al., 2008, AGU meeting).

trends indicate the fluid flow dynamics in real time. The increase in pore pressure causes micro fractures to generate the micro seismic events.

Dorbath et al., (2009), on the other hand, examined cumulative seismic moment with fluid injection in geothermal boreholes. The temporal evolution of the cumulative seismic moment showed that once the faults/fractures are activated, the seismicity becomes largely independent of the injection parameters. This means that, once a large fault is activated, their behaviour becomes independent of the injection. Their identification and mapping may be challenging for future prospecting.

Dasgupta and Jervis (2008) reported microearthquake

monitoring in a Saudi Arabian oil field (Arab-D) for mapping reservoir drainage pattern. The experiment was unique because of the large array of permanent multi-component seismic sensors that were deployed at various levels in the borehole and over a surface area surrounding the borehole. The microearthquakes down to moment magnitude $M_w -3.0$ were recorded simultaneously by the surface and borehole sensors. The microearthquake data could provide the location, relative fracture density and reservoir-flow for optimizing reservoir production and ultimate recovery. A 3D view of the microearthquake activity in an oil reservoir in Arab-D is illustrated in Fig.3.

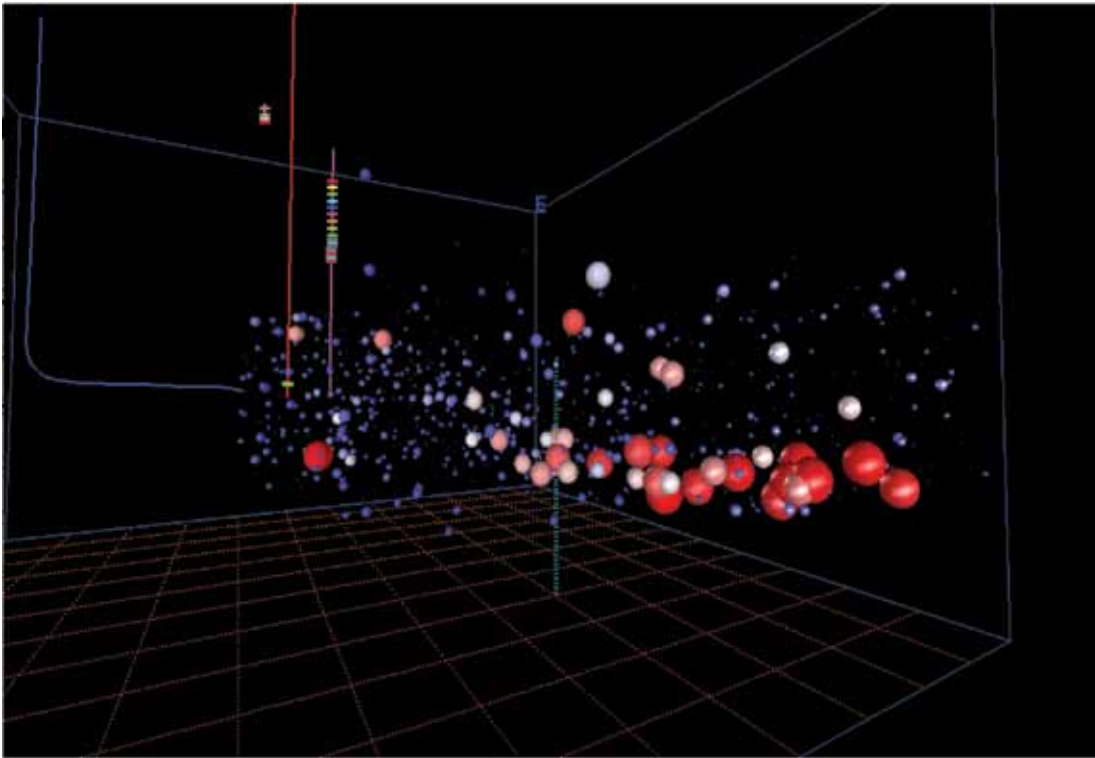


Figure 3. A 3D view of microearthquake hypocenters following liquid injection at the Arab-D reservoir level, the magnitudes (Mw -3.0 to 1.0) of the events are indicated by size of the balls (after Dasgupta and Jervis, 2008).

CONCLUSIONS

Micro earthquake monitoring techniques provide a method for continuously detecting micro seismic events for monitoring fluid pathways in a hydrocarbon reservoir. By combining surface and down hole sensors the mapping of fluid flow over a reservoir volume between the wells is possible. This information may be applied in inferring the distribution of reservoir properties such as permeability related to fractures and faults and reservoir connectivity for numerical simulation of fluid flow. The results increase the precision in reservoir model descriptions and potentially improved recovery.

ACKNOWLEDGEMENT

This paper is based on a presentation at the Indo-Norwegian Conference on CO₂ injection for EOR and geological sequestration, NGRI, Hyderabad, India, November, 26-28, 2012. I am sincerely thankful to Mr S N Dasgupta for his

kind help and cooperation for presenting this paper at the conference, and to the organisers for inviting this paper.

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Possibility of CO₂ Sequestration in Basalt and Sub-Basalt Sediments in and Around Peninsular India

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ABSTRACT

The present study brings out the recent geological findings concerning Cretaceous volcanism. These findings are very significant, not only for hydrocarbon exploration, but also in CO₂ sequestration in vast areas in and around peninsular India. The peninsular India is transected by several rift and grabens extending to offshore areas, which started subsiding since the beginning of Proterozoic times. During Proterozoic period sedimentary rocks were deposited within rift and grabens. This phase was followed by deposition of coal bearing sediments during Gondwana period, when the central parts were down faulted. The Cretaceous volcanism has taken place in pulses along these rift and grabens. These phased depositional episodes are more pronounced in their intersectional areas. The volcanic units are inter-layered with thick marine sediments in offshore regions and inter-trappean beds inland areas. The entire package of volcano-sedimentary rocks is sandwiched between the older Mesozoic and younger Tertiary sediments. This package laterally extends from land to offshore and beyond to the oceanic regions, covering the entire Bay of Bengal and the eastern half of the Arabian Sea.

INTRODUCTION

The Cretaceous volcanism comprises several felsic and mafic units, which are interlayered with sedimentary successions and represent entire ninety million years, from the beginning to the end of Cretaceous. These volcano-sedimentary sequences are invariably present in both producing as well as potential petroliferous basins of peninsular India. The producing basins include Barmer-Cambay, Bombay off-shore, Kaveri, Krishna-Godavari and Mahanadi. The potential basins include Kutch-Saurashtra and their off-shore components, Deccan Volcanic Province (DVP) of Maharashtra and Madhya Pradesh, Kerala, Mannar, Bengal oceanic basins with oceanic ridge components.

Earlier, it was not possible to ascertain the lithological characteristics of volcanic units and underlying sediments. Seismic studies for hydrocarbon exploration followed by drilling contributed significantly to our understanding of lithological characteristics and stratigraphy. Pre-Stacking and Depth Migration (PSDM) and Pre-Stacking and Time Migration (PSTM) processing techniques have facilitated better understanding of sub-basalt stratigraphy. The basaltic units form technical basement, over which uninterrupted Tertiary sequences are deposited on land, offshore and oceanic regions have been noticed in different parts of the globe, including India. The heat coming up along fractures provided environment for hydrocarbon generation. These fractures were later sealed by the flowing lava to provide an ideal trapping mechanism. The primary porosity of vesicles and cavities in basalts and secondary porosity developed due to cooling and fracturing, holds a fairly good amount of

hydrocarbons in them. The space created by extraction of hydrocarbons in the volcano-sedimentary package can best be utilized for CO₂ sequestration and enhanced recovery. In the present study all the basins and areas containing Cretaceous volcanic sequences, whether occurring on land, subsurface or oceanic regions, have been taken into account. Attempts have been made to map the lateral continuity along with the underlying and overlying sequences. Drill hole logs available from hydrocarbon exploration, have been utilized to know the thickness and nature of such volcanic units.

CRETACEOUS VOLCANICS AND CO₂ SEQUESTRATION

Earlier the volcanic flows exposed in the Deccan region of western India, were described as the Deccan Traps. Field mapping assisted by aerial photographs and satellite imagery (Misra, 1999, 2002, 2007) has brought out that the Deccan Traps form only a very small portion of Cretaceous volcanism. They cannot be stratigraphically separated from the other units, as they have similar petrological characteristics and wedge into each other. Recent studies (Misra and Misra, 2010; Misra and Joshi, 2013; Misra and Misra, 2013) have shown that in all the cases, entire package of stratigraphic successions laterally continues from land to offshore and later to oceanic regions. They cover almost entire Bay of Bengal in the east and up to Lakshadweep ridge in the west (Fig. 1), and are overlain by an uninterrupted Tertiary sequence. It has emerged that tectonic activity has largely affected rocks up to the Cretaceous period.

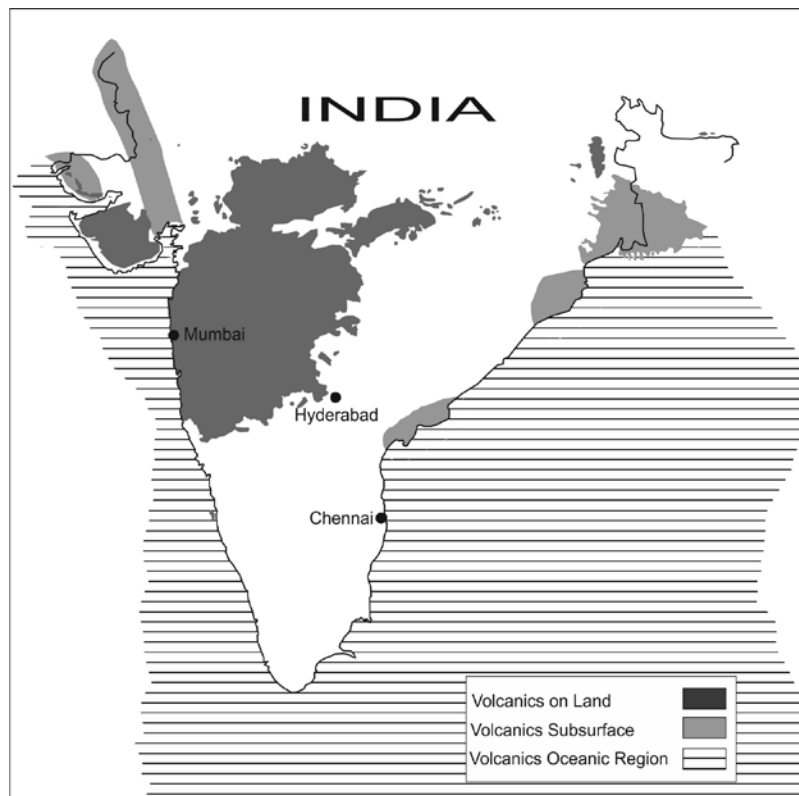


Figure 1. Outline map of Peninsular India, showing the distribution of Cretaceous volcanic units on land, in offshore and oceanic regions. (After Misra and Misra, 2013)

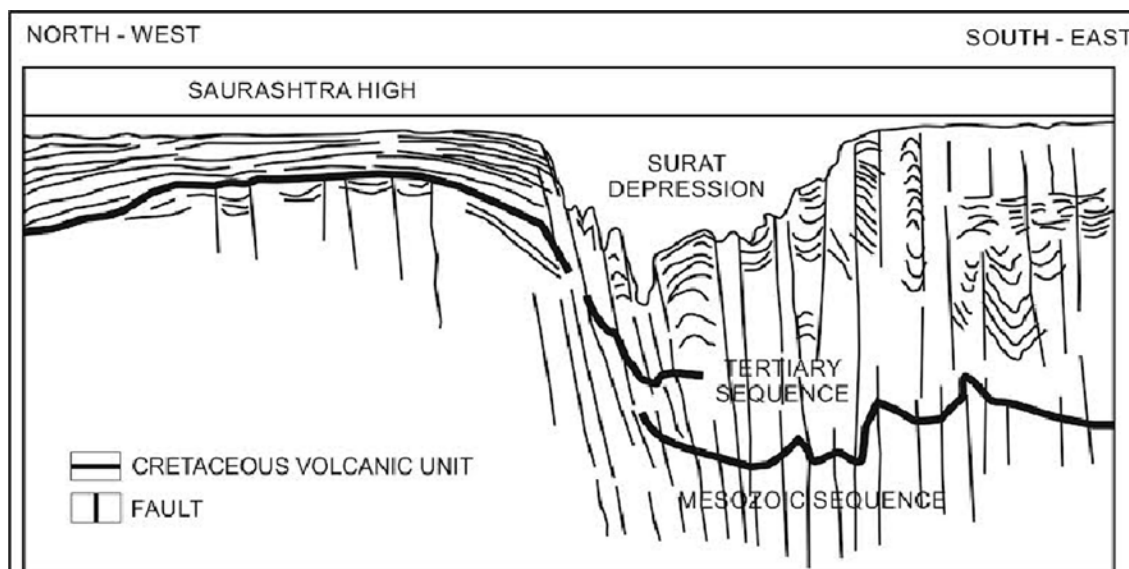


Figure 2. Interpretation of seismic profile across Narmada-Tapti Tectonic Zone in offshore region. Subsidence of blocks along vertical faults has increased the thickness of Tertiary sequence. (Modified after Misra and Misra, 2010)

A series of technological developments have significantly improved our understanding of structural disposition of Cretaceous volcanics. These developments include, improved resolution of multispectral and radar satellite imagery; processing of seismic data by PSDM and PSTM

techniques. Pre-stack partial migrations are required when there are conflicting dips with varying velocities or a large lateral velocity gradient (Gadallah and Fisher, 2009). Interpretation of these seismic profiles (Fig. 2) has helped to not only estimate the thickness of volcano-sedimentary

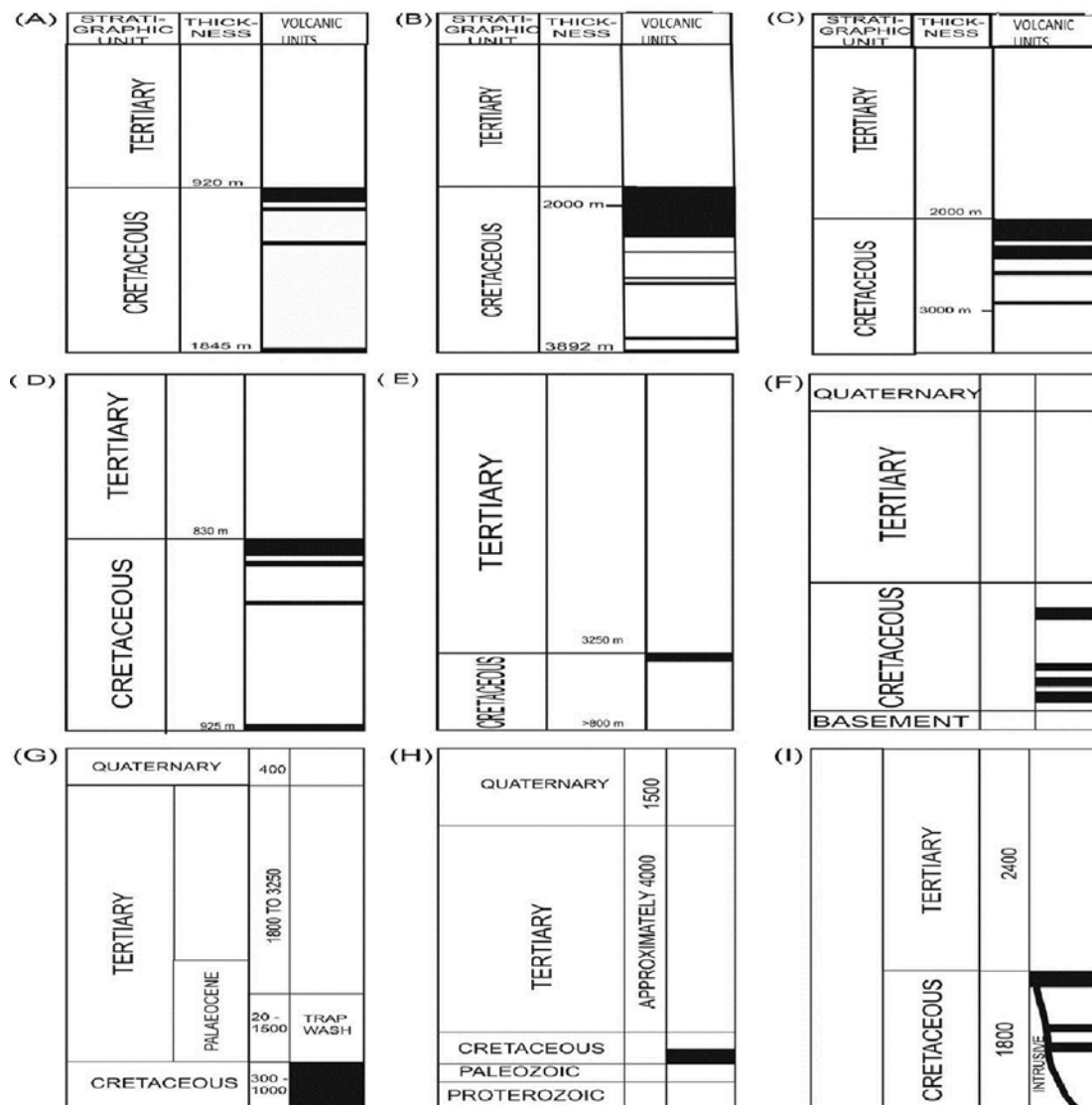


Figure 3. Generalized well logs of different basins, added and modified after Berger, et al., (1983), Fig. 3A Off-shore Kutch, Fig. 3B On-land Kutch, Fig. 3C. Off-shore Saurashtra, Fig. 3D. Cambay basin, Fig. 3E Krishna-Godavari basin, Fig. 3F Off-shore Mahanadi basin, Fig. 3G Lakshadweep basin, Fig. 3H Bengal basin, Fig. 3I Andaman basin.

sequence but also the details of underlying older sediments. Furthermore, innumerable borehole logs obtained during hydrocarbon exploration, have also established that the Cretaceous volcano-sedimentary sequence is integral part of stratigraphy in all the basins (Fig. 3). Evidences suggest that eruption of volcanic units have taken place along the major rifts and grabens, transecting the peninsular India (Misra, 2008). Both Mesozoic and Tertiary sequences along with the volcanic units are thicker in intersectional areas, due to compounding effects of subsidence along them. The temperatures of volcanics are estimated to have been around 750° to 800° C. These lava units spread over the earlier Mesozoic successions have sealed the earlier cracks like epoxy. It is because of this reason; higher pressures are noticed in hydrocarbon pools below the volcanic units. Such

sealed sedimentary basins are believed by the authors, to form ideal reservoirs for CO₂ sequestration, following the hydrocarbon extraction. These findings could be applicable to any of the peninsular basins.

Several units of Cretaceous volcanics are prolific producers of hydrocarbons. For example, Raageshwari field in Barmer, Ingoli and Padhra in Cambay are producing natural gas, while the down faulted Razol formation, which represents Rajahmundry volcanics in the K-G basin, is producing both oil and gas. The original source rock for hydrocarbons could be from both older Mesozoic and older sediments, or the sedimentary horizons interlayered with the volcanic units. In most of the cases, bore holes in Cambay basin are drilled only up to top volcanic unit. It may be rewarding to re-enter suitable wells, to

ascertain the hydrocarbon potential of older volcanic units and interlayered sediments for possible use in CO₂ sequestration.

CONCLUSIONS

The study has led to the following conclusions.

1. Cretaceous volcanic units are invariably associated with petroliferous basins wherein the magma was generated due to decompression melting and erupted along the subsiding rift and grabens. The volcanism was more pronounced in the intersectional areas. Furthermore, due to the compounding effects in these areas, the sedimentary sequences of older Mesozoic and younger Tertiary are also thicker along with volcanic units.
2. Nearly vertical basinal faults were not only responsible for formation of basins but provided heat for distillation of sediments in early stage while at later stages of volcanism, the lava has sealed the fracture system to form unique trapping mechanism.
3. Data processed by PSDM and PSTM techniques have helped to map structures where hydrocarbons are trapped. Such structures can be utilized for CO₂ sequestration and subsequently for hydrocarbon extraction.
4. Sediments below Cretaceous volcano-sedimentary sequence could be favorable for CO₂ sequestration. The most ideal will be those, from where the extractions of hydrocarbons are extracted and enhanced recovery is contemplated.

ACKNOWLEDGEMENTS

The authors are indebted to Dr. V.P. Dimri and Professor Mrinal Sen Former Directors, National Geophysical Research Institute (NGRI) for inviting us to present the work in Indo-Norwegian Conference on CO₂ Sequestration. We wish to thank Ms. Sujata Venkatraman, Program

Director, GX Technology, Houston, Texas, USA for providing the seismic profiles. Professor S.J. Chopra, Chancellor, University of Petroleum & Energy Studies has kindly provided a research environment where this work was carried out. We also thank anonymous reviewer and Prof. P.R. Reddy chief editor of the journal for very constructive suggestions, which helped to improve the manuscript.

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Some Insights into Possible CO₂ Sequestration in Subsurface Formations beneath Deccan Volcanic Province of India

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ABSTRACT

The mid-western part of on-land India and adjoining offshore region are covered with a thick pile of 65 Ma old Deccan volcanics, which are known to be the largest flood basaltic eruptions on the earth's surface. Extremely rapid eruption and fast cooling of these lavas have severely affected their physical properties like density, porosity and permeability, which have relevance to geologic sequestration of CO₂. Since this region is now being considered primarily for such an endeavour, the knowledge of the thickness details along with physical properties and structural disposition, besides what lies above and below these volcanics becomes essential. Based on some previous and our present multi parametric geological and petrophysical studies in a 617 m deep Killari bore hole (KLR-1), drilled in the eastern part of the Deccan Volcanic Province (DVP), we infer that these volcanics appear to be composed of thick columns of alternating layers of both massive and vesicular/amygdaloidal type basalt. Vesicular basalt has low density, low velocity and high porosity. In some areas (including offshore), Deccan volcanics are underlain by thick Mesozoic sediments, followed further by either Jurassic volcanics (like Saurashtra region) or high density, high velocity CO₂ rich amphibolite to granulite facies transitional mid-crustal rocks. Based on our findings, we suggest that in comparison to Deccan volcanics, the subsurface thick Mesozoic sediments as well as immediately underlying Mesozoic volcanics, can be seriously considered as a leading option for geologic CO₂ sequestration. At the same time, since the entire DVP is pervasively fractured, faulted and highly deformed, apart from being seismically active since historical times, CO₂ sequestration in on-land exposed volcanics should be given least priority.

INTRODUCTION

In the last 300 Ma, there have been several volcanic episodes on the earth's surface like, Columbia River, Ethiopian, Deccan Traps, Serra Geral, Karoo and Siberian basalts. The 65 Ma old Deccan volcanics, which cover almost one-sixth of the Indian landmass and extends for over half a million square kilometres on the onshore and offshore near western margin, is considered the largest among them all. These volcanics are composed of a large number of gentle dipping tholeiitic basalt flows, having an estimated thickness of more than 2 km at places, resting directly either over the late Archaean crystalline basement or thick Mesozoic sedimentary cover. The main phase of this historic volcanism had an extremely short duration of only about 10,000 years (Courtillot, 1990).

Currently, this large igneous province (Fig. 1) is viewed as a prominent site for the storage and injection of CO₂ either directly into Deccan volcanics or underlying sediments to reduce greenhouse warming. Present study aims at discussing various possible options including Deccan volcanic formations for CO₂ storage, based on recently acquired geological and geophysical informations.

NEED TO CO₂ CAPTURE AND STORAGE

It is no secret today that as much as 22 billion tons of CO₂ is getting pumped into atmosphere from man-made or anthropogenic sources (Benson and Surles, 2006). It is

feared that it may cause considerable stress to the climate and thus to the existence of mankind itself. On world wide scale, CO₂ injection for enhancing oil recovery, has been considered a major option since 1950's (Karmarkar, 2014). However, as a remedial measure to global climate change, CO₂ capture and sequestration (CCS) in deep geological formations, has emerged as an important option over the past two decades (Torp and Gale, 2002; Benson and Cole, 2008). It started in a big way in 1996 with a Sleipner saline aquifer storage project in the North sea (Torp and Gale, 2002), followed by many such projects like In-Salah gas field project in Algeria in 2004 (which stores CO₂ in the flank of a depleting gas field) and Weyburn project in Saskatchewan (Canada), where enhanced oil production and storage of CO₂ in the Weyburn oil field happens simultaneously (Moberg et al., 2002; Riddiford et al., 2002). Currently, some of the large scale CCS operations involve storing of around one million tons a year, and encouragingly, several such projects are planned by many other countries like Algeria, Mexico, Saudi Arabia and United Arab Emirates. To have a significant impact on climate, many billion tons of CO₂ need to be sequestered annually and that would require selection of proper depository sites.

BASIC REQUIREMENTS FOR SITE SELECTION

Basic requirement for CO₂ sequestration is the site selection and thereafter monitoring. Selecting an appropriate

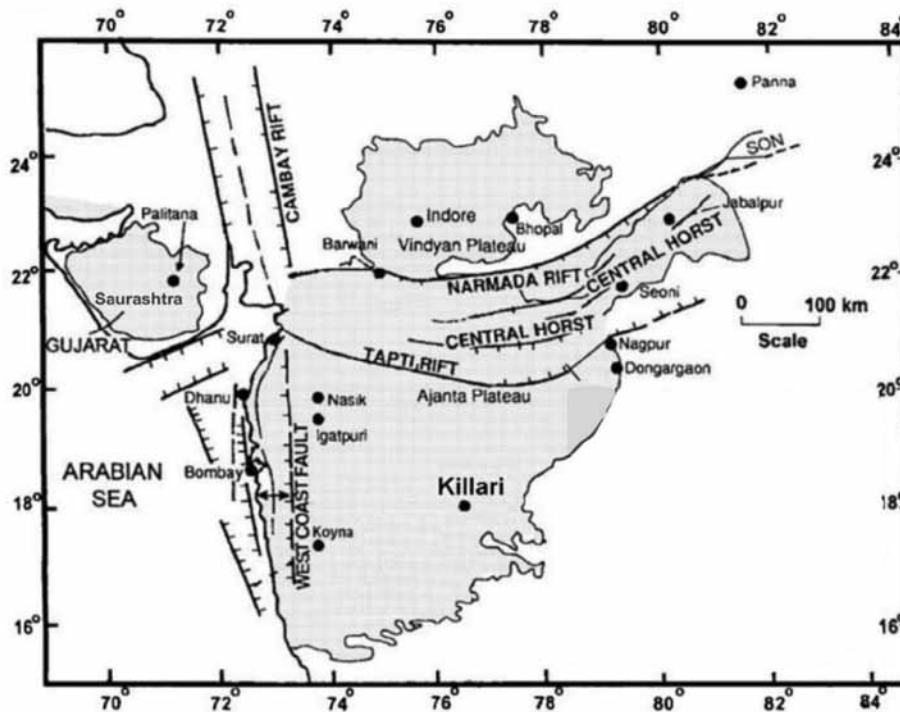


Figure 1. Major geotectonic elements and known extent of on-land Deccan flood basalts in western India.

storage site with minimal leakage risk is not an easy task. Current scenario indicates that there is no serious problem in sequestering CO₂ directly into (i) thick permeable sediments of depleted oil and gas reservoirs, (ii) brine-filled porous reservoir rocks saturated with brine (or saline formations), and (iii) deep thick coal seams (Benson, 2005; Metz et al., 2006), if they have impermeable and low porous cap rocks (viz., fine grained shale, mudstone or carbonate rocks like limestone and dolomite etc.) above and below the sequestered layer. Apart from thick sediments, fluid filled deep volcanic sequences are also now being considered a safe option, if its physical properties are favourable, with proper capping of rocks. Such regions preferably need to be devoid of faults and aseismic in nature. Thus, before embarking on CO₂ sequestration in volcanic formations, knowledge of their thickness, physical properties, structural disposition as well as what lies above and below of it, becomes essential.

CONSTRAINTS ON POSSIBLE CO₂ SEQUESTRATION IN DECCAN VOLCANICS

Deccan volcanic magmas (tholeiitic basalts) apparently fractionated in the lower crust, before their fast extrusion on the earth's surface (Pandey and Negi, 1987). They are not derived directly from the mantle and as such, their shallow origin and rapid extrusion as well as cooling, considerably affected their physical properties.

Conforming to this, the estimated Mg Ratios for the Mahad - Mahabaleswar section, which is one of the thickest Deccan Trap sequence, were found quite low (0.34 - 0.50). Unusual nature of these basalts has been further confirmed by laboratory measurements on core samples derived from KLR-1 borehole drilled in the epicentral region of 1993 Killari earthquake region (Gupta et al., 2003).

This 617 m deep borehole penetrated 338 m thick Deccan basalts, followed further by 8 m of intratrappean sediments and 270 m of late Archaean crystalline basement (Gupta and Dwivedi, 1996). 53% of the basaltic column above the basement, comprised massive basalts with a high density of 2.90 g/cm³ (Fig. 2), while the rest 47% of the basaltic column consisted mainly of non-massive vesicular variety, characterized by dry and wet densities of 2.36 g/cm³ and 2.56 g/cm³, respectively (Gupta et al., 2003). Non-massive variety is found to contain lots of void spaces with estimated porosity of 20% and more. Average density of Deccan basalt is estimated to be around 2.65 g/cm³ only. Our own measurements too have indicated a wide range in dry density from 2.24 g/cm³ to 3.00 g/cm³ for the Killari borehole basalts, conforming to their both massive and non-massive nature. The average density of these volcanics is lower than the expected density. This is also reflected through much lower P-wave velocities of 4.3 to 5.2 km/s (Kaila et al., 1990; Sain et al., 2002; Sain and Zelt, 2008; Dixit et al., 2000, 2010; Murthy et al., 2010, 2011; Pandey et al., 2011).

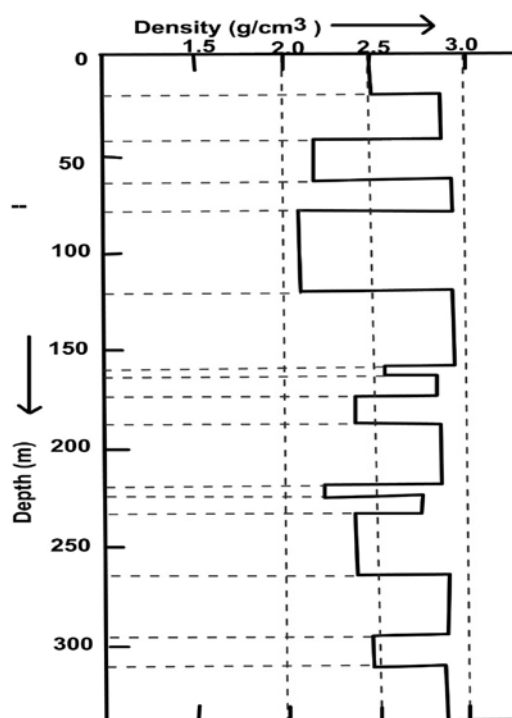


Figure 2. Measured mean density variation in Deccan Trap sequences penetrated by KLR-1 borehole in killari region of Maharashtra (Gupta et al., 2003). Massive and vesicular and/or amygdaloidal basalts are characterised by high and low densities respectively.

The underlying crystalline basement below the Killari earthquake region has been diagnosed as high density (2.82 g/cm^3), high velocity late Archaean amphibolite to granulite facies transitional mid crustal rocks, containing more than 2 wt% of CO_2 (Pandey et al., 2009, 2014; Tripathi et al., 2012 a, b). Similar grade high velocity basement rocks are expected in other Deccan volcanic covered areas also (Rathore et al., 2000; Krishna, 2006; Pandey, 2009; Rao et al., 2013).

It is indicated from the recent geophysical and deep drilling results that apart from exposed areas, considerable thickness of such volcanics are concealed below thick sedimentary column at several places that would include offshore areas too, for example, Cambay graben, Kachchh-Saurashtra region and adjoining onshore and offshore areas (Biswas, 2005; Sain and Zelt, 2008; Carmichael et al., 2009; Dixit et al., 2010; Pandey et al., 2011; Murty et al., 2011). On-land, low density porous volcanic formations are as thick as 45m, which are bounded on either side by massive non-porous high density basalts (Fig. 2). Similar massive/porous Deccan volcanic formations are also present at subsurface depths below the thick sediments, as they too are characterised by a $V_p \sim 4.75\text{--}5.2 \text{ km/s}$ (Dixit et al., 2010), similar to the on-land exposed volcanics ($V_p \sim 4.6\text{--}5.0 \text{ km/s}$; Murty et al., 2010). These subsurface volcanic layers are sometimes quite thick. For example, near Ankleshwar, more than 3 km thick volcanics are expected to be present below $\sim 2 \text{ km}$ thick sedimentary

column (Dixit et al., 2010). The low density/low velocity, sub surface highly porous non-massive volcanic zones, which are sandwiched between massive (high density and almost zero porosity) volcanic layers on top and bottom, could turn out to be a suitable on-land/offshore site for CO_2 sequestration. It would fulfil major geological criteria, like leakage, impervious cover and large storage capacity. Interestingly such volcanic layers have thick overburden and multiple impervious horizons.

SEQUESTRATION IN MESOZOIC SEDIMENTS

Recent studies (Kaila et al., 1990; Dixit et al., 2000; Sain and Zelt, 2008; Carmichael et al., 2009; Murty et al., 2010, 2011; Pandey et al., 2011; Rajendra Prasad et al., 2010) indicate presence of thick Mesozoic sediments (up to $\sim 3 \text{ km}$ thick) below Deccan volcanic flows. They mainly contain sandstone and claystone rocks (Dixit et al., 2000; Pandey et al. 2011). Fig. 3 illustrates the nature of one such sequence. Deep-seated column of porous sandstone, a few thousand meters below the surface and having suitable cap rocks on top and bottom, may provide an ideal situation for CO_2 injection and storage. The pore spaces in sedimentary columns are usually filled with salt water (saline formations), where oil/gas reservoirs are often located. Importantly, multiple sealing units, like claystones will protect against fluid migration, a prerequisite condition for CO_2 sequestration.

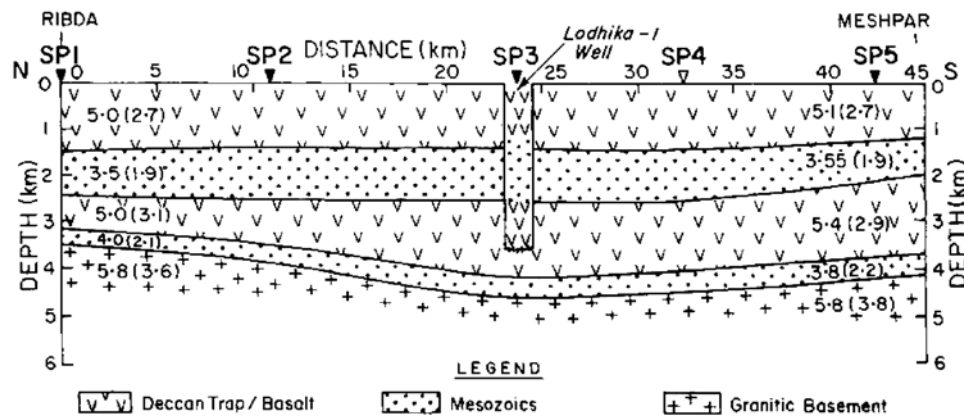


Figure 3. Occurrence of subsurface Mesozoic sediments and older Mesozoic volcanics along the Ribda-Meshpar profile, in the Deccan volcanic region of Saurashtra, Gujarat (after Dixit et al., 2000). P- and S-velocities of each layers, together with the location of the Lodhika-1 borehole, is also shown.

CO₂ SEQUESTRATION IN JURASSIC VOLCANICS - A STILL BETTER OPTION

In certain areas like Saurashtra peninsula, Deccan basalt covered Mesozoic sediments are further underlain by 0.75 to 1.5 km thick older volcanic sequences of early Jurassic age (~184 Ma), having a relatively higher velocity of 5.0 to 5.4 km/s (Dixit et al., 2000). They are concealed below a total overburden of about 2 to 3 km (Dixit et al., 2000; Sain et al., 2002) and formed consequent to thermal perturbations caused by initial rifting between Africa and India. These volcanics may prove to be a better option for CO₂ sequestration than Deccan volcanics. Having a thick and multiple impervious overburdens, they would be free from leakage problem and fluid migration. Only disadvantage could be limited pore volume than the Mesozoic sediments. Their porosity, however, would probably be similar to the Deccan volcanics, in view of the velocity estimates (Fig. 3). It will also fulfil the requirement of high in-situ temperature and pressure.

DISCUSSION AND CONCLUSIONS

Different types of subsurface rock formations can be considered as large carbon reservoir. They include coal seams, organic rich shales, hydrocarbons and carbonate-rich rocks. In connection with the enhancement of oil recovery, CO₂ sequestration was initiated as early as 1950's (Khatib et al., 1981; Grigg and Schechter, 1997; Rogers and Grigg, 2001; Karmakar, 2014). However, for a large quantity of geologic sequestration of CO₂, the depleted oil/gas reservoirs may not always be readily available.

It is feared that injection of million tons of super critical liquid CO₂ directly into deep geological formations, may induce earthquakes. In this context, Deccan volcanic

region, unlike countries like Norway, where CO₂ injection has been successfully attempted, may not be ideal as it is highly fractured, faulted and seismically unstable (Ravi Shanker, 1995; Agrawal and Pandey, 1999). This region has witnessed mild but significant seismic activity in the past in the form of many damaging earthquakes, like, 1967 Koyna (M 6.3), 1993 Killari (M 6.2), 1997 Jabalpur (M 5.8) and 2001 Bhuj (M 7.7) (Pandey et al. 2009).

Interestingly, there are no such reports of significant increase in induced seismicity with any of the storage sites so far. In any case, it can be managed with suitable monitoring. Taking the above facts and the latest geological and geophysical findings into consideration, we suggest that subsurface rock formations like (i) Deccan volcanics, concealed below thick Tertiary sediment cover, (ii) thick Mesozoic sediments underlying these subsurface volcanics, and (iii) the older Mesozoic volcanics which form the base of Mesozoic sediments (like that found in Saurashtra region, Fig. 3), may form imminently suitable sites for geologic sequestration of CO₂. Out of the three options outlined above, second and third options are preferable. However, earlier detailed factors must be taken into consideration before embarking on such an endeavor.

ACKNOWLEDGEMENTS

The authors are extremely thankful to Satyajit Dutta for helping in the organization of the manuscript. O.P. Pandey is thankful to CSIR (New Delhi) for granting Emeritus Scientist position at CSIR-NGRI, Hyderabad. Nimisha Vedanti and SSG wish to thank Royal Norwegian Embassy and SHORE project of CSIR-NGRI for supporting this study. Permission accorded by the Director CSIR-National Geophysical Research Institute, Hyderabad-500007 to publish this work is gratefully acknowledged.

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Investigation of Geomechanical and Rock Physics Aspects Related to Underground Storage and Monitoring of CO₂

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ABSTRACT

This paper presents an overview of activities related to identifying and quantifying leakage risks as well as monitoring methods as applied to large scale, long term underground sequestration of CO₂. Here, the focus is on geo mechanical and rock physics related issues that are identified as major show stoppers for successful operations as well as potential monitoring methods in addition to surface seismic techniques. Laboratory experiments are made to investigate possible fracturing mechanisms around injection wells and stress change effects on sonic velocities in sealing overburden cap-rock associated with injection in an underlying reservoir. Numerical simulations investigate both the near-well region and large reservoir-scale, looking for relevant scenarios where stress changes can trigger fracturing, in particular, the sealing cap-rock.

INTRODUCTION

Mitigating climate change implies reducing dramatically the amounts of greenhouse gases in the Earth's atmosphere. In particular, many research projects target reducing the amount of CO₂, by targeting efficiency improvements leading to reduced CO₂ emissions and by attempting to capture the gas (e.g. from coal-fired power plants) before it is released to the atmosphere. An inevitable corollary to CO₂ capture is its disposal and hence, the research efforts are to be put in to identify suitable storage sites. In particular, the idea to enable large scale underground sequestration is gaining momentum, either in aquifers or depleted oil and gas reservoirs.

In order to successfully store CO₂ underground, it is imperative to be able to identify all risks that could lead to leakage and loss of containment. In addition, if a leak occurs, monitoring methods must be elaborated to provide early warning and enable efficient intervention. Reliable monitoring methods should also provide the means to quantify and locate stored CO₂ volumes. Two areas are identified as most problematic for storage, namely injection wells that contain sealing caprock. Loss of injectivity and failure of well barriers together with fracturing in the overburden are the most serious factors for loss of reservoir containment.

4D MONITORING TECHNIQUES AND CHALLENGES

Various monitoring techniques are envisaged to be used in the long term surveillance of a CO₂ injection storage site, such as repeated seismic surveys and frequent common source electro-magnetic (CSEM) surveys. The Combination of these two techniques will improve imaging resolution as will added sonic monitoring.

In order to improve 4D monitoring, fundamental understanding of the impact of fluid injection on the sonic velocity is required (e.g due to stress changes occurring in the reservoir and the overburden during injection of CO₂). These variations add to velocity changes caused by fluid substitution. The acoustic signature of both reservoir and overburden, at different locations, must be interpretable in terms of stress state, in particular in the overburden. This can help in determining whether the overburden strength is being exceeded and leakage becoming imminent. Both seismic frequency and sonic log interpretation for repeat surveys rely on knowledge of the elastic properties of the rock layers above, inside, below and around the storage reservoir. These will change with stress change and fluid saturation. It is thus expected that both the cap-rock and reservoir stress states will change with increase in the reservoir pore fluid pressure, itself accompanying CO₂ injection campaigns. Due to the limited extent and shape of the reservoir, these stress changes are complex and location-dependent.

The purpose of acoustic monitoring under injection of CO₂ in a reservoir is thus to map the acoustic signature of both reservoir and overburden rocks. The velocities measured must be interpreted in terms of corresponding stress state. Once the stress map is obtained, local values can be compared to the formation strength. If there is excess stress, a fracture may start developing from this location. This is particularly important for the overburden rock, right above the reservoir, since opening a fracture there implies loss of sealing of the reservoir with potential upwards migration of the stored CO₂ towards overlying aquifers and their contamination, let alone a worst case scenario with the CO₂ migrating all the way to surface.

Laboratory experiments have been performed with soft synthetic sandstone cemented under stress. Passive listening to sandstone rock plugs was achieved by using the

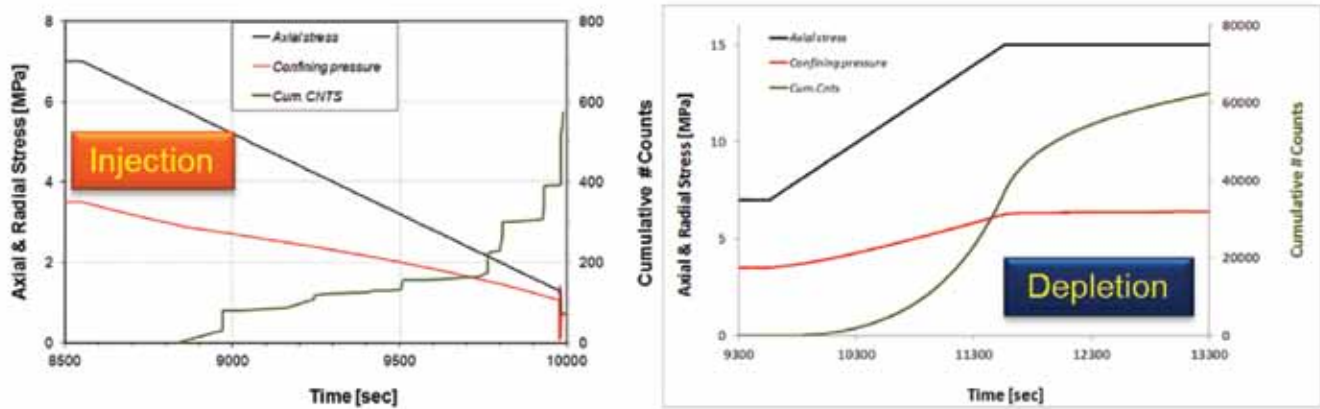


Figure 1. Acoustic emission counts for stress paths simulating injection of a pore fluid (left) and depletion associated with pore pressure decrease (right).

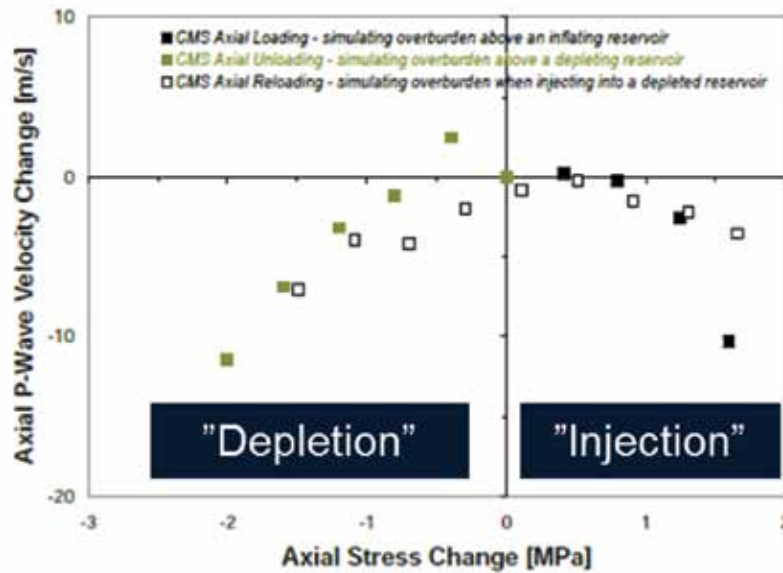


Figure 2. Experiments on compacted brine-saturated clay specimens to simulate low permeability cap-rock formations: transmitted sound velocity measurements for initial stress state and stress changes representative of those occurring in-situ during depletion or injection into an underlying reservoir.

acoustic emission (AE) technique to monitor the stress state in an analogue to a reservoir rock as pore pressure changes (Gettemy and Holcomb, 2006). This technique captures the sonic bursts accompanying micro-fracturing events in the rock. Hysteresis behaviour was observed, where depletion-related compaction or even shear failure produced more AE events than injection-induced extension of the rock matrix (Fig.1). This is expected, since bulk rock behaviour during unloading is largely elastic and hence does not generate significant amount of AE. However, one will be able to pick up micro seismic events associated with fracturing or fault displacements induced by the injection process (Zoback, 2011), but most likely not their precursors.

The data demonstrate that non-elastic, grain-scale

processes contribute to large acoustic emission (AE) activity during depletion-induced compaction and to significantly less but pronounced AE activity associated with injection-induced extension. Grain scale deformation mechanisms contribute to our understanding of rock failure processes at this scale. Experiments with transducer arrays to locate the fracturing events in space and waveform analysis to address focal mechanisms will be helpful in further studies along this line. Although failure mechanisms are largely scale invariant, the findings from this type of laboratory experiments cannot be directly transferred to complex field situations; laboratory data are useful for generating constitutive models for prediction of fracture initiation and growth under the right field conditions. Wave velocities

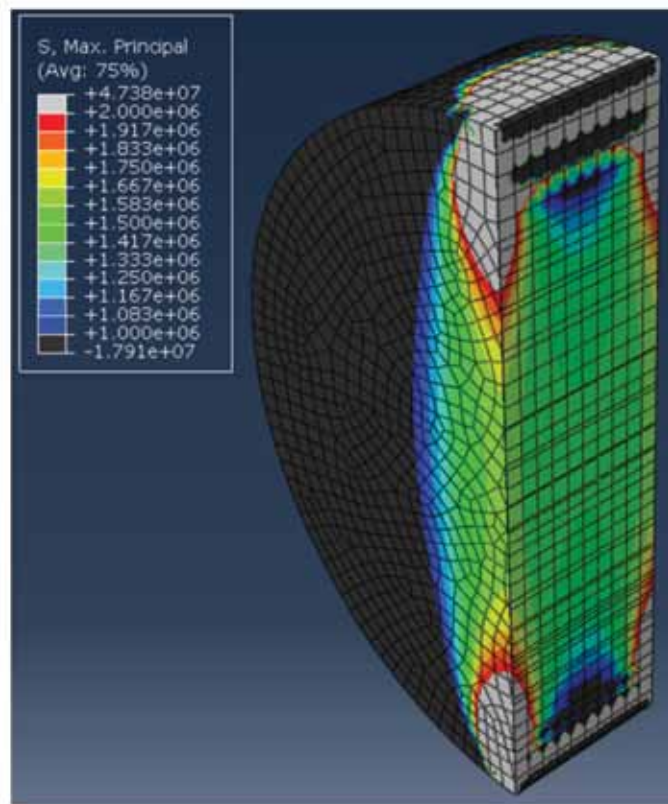


Figure 3. Tensile principal stress distribution in a steady-state simulation of a disk specimen placed between fixed pistons and heated from 0°C to 80°C.

in the synthetic sandstone are strongly stress dependent, in particular, implying a strong seismic velocity decrease during simulated injection (Holt and Stenebråten, 2013). Other experiments have been conducted on compacted brine-saturated clay specimens. These experiments simulated an initial stress state and stress changes representative of those occurring in-situ during depletion of or injection into an underlying reservoir. The results (Fig. 2) show that under injection conditions, the slow-down in sonic velocities serves as a signal when the cap rock is on the verge of failure, thus giving an early warning for a potential leakage.

THERMAL STRESS RISK ASSESSMENT

Injecting cold CO₂ into a warmer reservoir may create thermal contrasts all along the well trajectory (Krogh et al., 2012); in the worst case, the thermal stresses may induce tensile fracturing of the overburden in the immediate vicinity of the well, creating a leakage path. Another danger is the delamination or interface fracturing of the different layers around the injection well (metal casing, cement sheath and surrounding formation).

Numerical simulations confirmed that radial tensile fractures can be expected in the near-well area in shale

layers adjacent to sandstone layers if the cooling is sufficient, typically by more than 50°C, for given, typical, rock properties (Hettrema et al., 1992). The simulations indicate that tensile hoop stresses are likely to develop in the shale layers and be sufficiently high to generate radial tensile fractures in shale (Lavrov and Cerasi, 2013). It should be noted that 'shale' here means a rock with a relatively high thermal expansion coefficient, compared to 'sandstone' (Gilliam and Morgan, 1987).

A laboratory study of thermal fracturing in shale is initiated to investigate if and how such fractures can develop in shale. Since it is impractical to perform uniaxial test with cooling (instead of producing thermal cracks, the contracting sample will most likely separate from the loading pistons), a simple so-called Brazilian test is set-up, where the specimen is subjected to heating. In this test, also called indirect tensile test, a thin cylindrical sample is compressed axially inducing tensile fracturing in the disk's center. In the thermal version, tensile thermal stresses are generated in the central part of the disk, thus providing a laboratory model of the thermal cracking expected under in-situ cooling.

Finite element numerical simulations were performed to help plan the laboratory tests and showed that (Fig. 3), if the thermal state is allowed to equilibrate, heating from 0°C to 80°C is likely to produce tensile failure just as an

ordinary (purely mechanical) Brazilian test would. The remaining question is whether such fracturing in shale would propagate far enough to be of concern in terms of initiating a leakage out of a storage site. These simulations show that thermal stresses are of concern for liquid CO₂ injection, but probably not relevant for injection of a gas, although this conclusion depends on tensile strength and thermal property variations.

CONCLUSION

The importance of sound geo mechanical analysis is getting more and more recognized. A certified CO₂ storage site in terms of sufficient capacity may still not be satisfactory due to concerns of fracturing risk and leakage. SINTEF is involved in research on several fronts concerning monitoring and qualification of storage sites. This includes laboratory work addressing fracturing and 4D rock physics aspects. Ongoing research activities include simulation work supplementing laboratory findings, and upscaling to field conditions. Near-well integrity issues and reservoir scale fracture risks are at the centre of both laboratory and simulation efforts.

ACKNOWLEDGEMENTS

This publication has been produced with support from the BIGCCS Centre, performed under the Norwegian research program Centres for Environment-friendly Energy Research (FME). The authors acknowledge the following partners for their contributions: Aker Solutions, ConocoPhillips,

Gassco, Shell, Statoil, TOTAL, GDF SUEZ and the Research Council of Norway (193816/S60).

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Visualizing Uncertainty in CO₂ Plume Migration During Sequestration

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ABSTRACT

During the operation of a geological carbon storage project, verifying that the CO₂ plume remains within the permitted zone will be of particular interest both to regulators and operators. A model selection algorithm was developed, which refines an initial suite of subsurface models representing the prior uncertainty to create a posterior set of subsurface models that reflect injection performance consistent with that observed. Such posterior models can be used to represent uncertainty in the future migration of the CO₂ plume. The method provides a very inexpensive alternative to map the migration of the plume and the associated uncertainty in migration paths due to the fact that only injection data is required. An essential aspect of the model selection algorithm is to group prior models on the basis of their connectivity. The base algorithm assesses that connectivity using a physical proxy such as a random walker. An alternate approach would be to use statistical tools for assessing connectivity of models. In this paper, we present an approach to compute the shortest connected path between well locations in an aquifer model and to define a measure of similarity of shape based on the concept of a discrete Fréchet distance.

INTRODUCTION

A key difficulty in accurately predicting the CO₂ plume migration path in the subsurface is the uncertainty in the underlying geology. To resolve that uncertainty, the observed flow response of the subsurface system can be used to optimally constrain the subsurface models. The key underlying technology is thus parameter estimation (commonly referred to as history matching). As opposed to traditional permeability or probability perturbation methods (e.g. Pro-HMS - (Kim, 2008)) that work on updating permeability values at a pixel-level, the proposal is to work directly in the realization space of the random function (RF) model. In traditional approaches, $P(A|B)$ assumes that A is a simulation event at a pixel level (e.g. permeability at a location u) and B is the available data. In this new proposal, A will be a model realization of the RF (i.e. it will be all the pixels taken jointly). Thus $P(A|B)$ will be the conditional probability of a model given the data. Similarly, $P(A|C)$ is the probability of a model given the injection data C . At the end of the merging process, we get $P(A|B,C)$ i.e. the probability that a model A that exhibits conformance to the available static and injection data. If the distribution $P(A|B,C)$ exhibits a peaked structure, this implies that the available data cause some of the models to be preferred over others. The history matching (injection data integration) approach therefore becomes an exercise in selecting the most plausible model(s) guided by the available data.

Here we propose a new perspective: rather than estimating the properties of the storage formation, we seek

to infer the location of the plume of injected CO₂ within that formation. In particular, we wish to be able to identify deviations of the plume from its anticipated migration path, as these deviations can have substantial regulatory ramifications. A primary cause of such deviations will be heterogeneities in the storage formation (baffles, sealing faults, high permeability streaks, and the like), which are not known at the beginning of the injection. A unique model-selection algorithm will be implemented that uses the injection data to select a subset of initial aquifer models that best adhere to observed injection characteristics. The simulation of plume displacement in this reduced subset of models yields the uncertainty in the plume location.

MODEL SELECTION

In this new framework, we start with a whole suite of plausible models for the particular subsurface system being studied. This set of prior models could be rather large, reflecting uncertainty in depositional environments and aquifer architecture. This is likely to be the situation at the outset of many geologic sequestration projects. This suite of prior models will be processed through a fast transfer function model such as a streamline simulator or a random walker or alternatively, a fast algorithm for assessing reservoir connectivity as in this paper. The fast transfer function model is a proxy for the aquifer flow model and is only used to gauge rapidly the comparative performance of the prior models. Denoting the response of the proxy function on the prior model $z^l(u)$, $\forall u \in \text{Domain}$ as

$f^i(\mathbf{u})$ and assuming that there are L prior models, we can calculate a distance matrix:

$$D = \begin{bmatrix} d_{11} & \dots & d_{1L} \\ \vdots & \ddots & \vdots \\ d_{L1} & \dots & d_{LL} \end{bmatrix}$$

the distances d_{ij} are defined as $d_{ij} = ||f^i(\mathbf{u}) - f^j(\mathbf{u})||$. Multivariate classification techniques such as Principal Component Analysis, Independent Component Analysis or k-mean cluster analysis would yield a grouping of aquifer models in terms of similarity of the proxy function response. We emphasize that this grouping procedure might result in geological models representing different environments/architecture getting grouped together. This would indicate that despite the overt differences in geology, those models exhibit similar connectivity characteristics that have a predominant influence on the observed injection data. This feature will also be exploited below to provide a quantitative estimate of uncertainty in the plume location.

Based on the distances between the models that make up a cluster, an average realization representative of the cluster can be computed by performing distance-weighted averaging of the realizations making up the cluster.

Alternatively, any of the models in a cluster could be retrieved as a representative model for that cluster. We denote these representative models as $z^m(\mathbf{u}), m=1, \dots, M$, where M is the number of clusters identified. Since each of the M models is equally conditioned to the available well information (such as log/core data etc.), each one is equally plausible. In that case, the prior probability $p(z^m(\mathbf{u})) = 1/M$.

Flow simulations with the full physics of the flow process of interest can be performed on the M reduced models. Denoting the corresponding responses as $RF^m(\mathbf{u}), m=1, \dots, M$ and given the observed injection history RF_{ref} , we can calculate the deviation $\sigma_m^2 = ||RF_{ref} - RF^m(\mathbf{u})||$. If $RF^m(\mathbf{u})$ is a vector in time i.e. $RF^m(\mathbf{u})$, then the deviation is calculated as $\sigma_m^2 = [RF_{ref} - RF^m(\mathbf{u})]^T [RF_{ref} - RF^m(\mathbf{u})]$. Assuming a Gaussian probability distribution for the mismatch with the mean as RF_{ref} and the variance as σ_m^2 , we can calculate the probability that any model in the cluster m would match the observed data i.e. $p(RF_{ref} | z^m(\mathbf{u}))$. The prior probability of the response RF_{ref} within the pool of M responses for the reduced models i.e. $p(RF_{ref})$ can also be calculated by pooling together the responses for the M reduced models and finding the probability corresponding to RF_{ref} .

The objective is to derive the posterior probability $p(z^m(\mathbf{u}) | RF_{ref})$ i.e. the updated probability for the M models given the observed injection response. This can be computed by a straight forward application of Bayes' rule:

$$p(z^m(\mathbf{u}) | RF_{ref}) = \frac{p(RF_{ref} | z^m(\mathbf{u}))}{p(RF_{ref})} \times p(z^m(\mathbf{u}))$$

All the quantities on the RHS are available. Hence, the updated probability can be calculated. A random draw can be made from the updated probability to retrieve the particular model $z_i^m(\mathbf{u})$, where the subscript i denotes the iteration for the application of this workflow.

After an application of the above process, a group m that exhibits the flow response that is similar to the observed injection data is identified. It is quite likely that a one-time application of this process would not yield a satisfactory match to the observed injection history. Consequently, the process is repeated using the member models that make up the group m . Multivariate classification (cluster analysis or PCA) is performed in order to further subdivide the member of group m . Application of the Bayes' rule now will yield the subgroup $z_i^{m'}(\mathbf{u})$ for iteration $i=2$. Repeated application of the process therefore refines further the selection of the model that is closest to the observed injection history. The entire process is summarized in the schematic in Figure 1. Flow chart of the model selection process using injection data..

A DEMONSTRATION EXAMPLE

This approach is applied to a synthetic case, which is modeled after the In Salah field. The preferential orientation of the high permeability features in the In Salah field is in the SE-NW direction - so we assume that the major orientation of the channel reservoir is in that direction. Fig. 2 shows the preferential orientation, a training image and two realizations generated using the multiple point stochastic simulation algorithm SNESIM (Strebelle, 2002). 400 realizations are generated using SNESIM and then they are clustered based on the similarity in connectivity characteristics of the models.

We need a direct measure of similarity of spatial connectivity in order to cluster reservoir models by spatial connectivity. We introduce the notion of a connected path to describe spatial connectivity of reservoir models. Since the injected fluid flows along permeable zone, the connected path is defined as the most permeable path between an injection point and a target point (which could be a hypothetical monitoring location) so as to delineate reservoir connectivity. Details of the procedure to compute the connectivity can be found in Srinivasan and Jeong (2012). Broadly, the computation procedure consists of identifying the shortest connected path between an injector and a hypothetical producer and then computing the distance between the shortest path for two different reservoir models using the notion of the Discrete Fréchet Distance (DFD – Eiter and Mannila, 1994). This permits the computation of the distance matrix. The realizations are projected onto a metric space or a kernel space using multidimensional scaling (MDS – Romney, 1972; Kruskal, 1978) or kernel principal component analysis (KPCA -

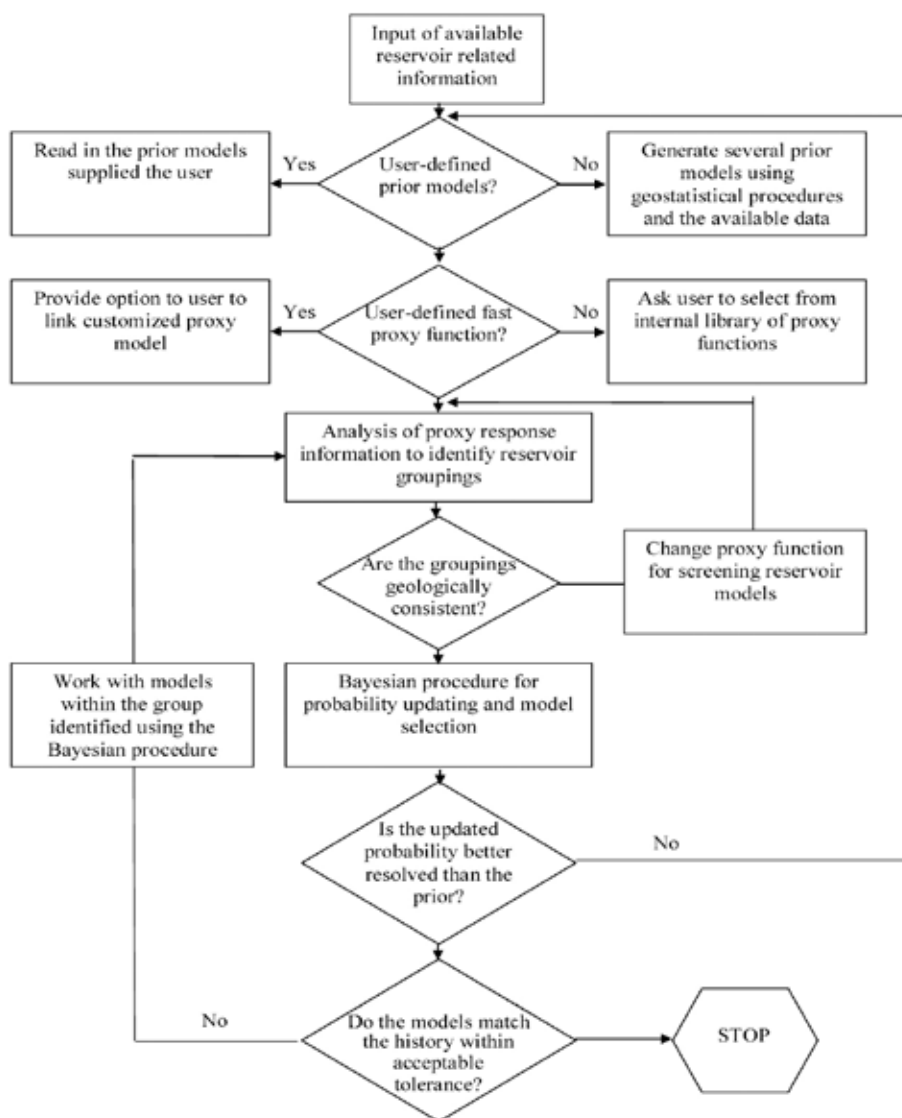


Figure 1. Flow chart of the model selection process using injection data.

Scholkopf, 1999) of the distance matrix. In the projected space, the realizations are clustered using K-means clustering.

In this synthetic example, there is one producer and three injectors. The region of interest is in the vicinity of the three injectors: KB-501, KB-502, and KB-503. We calculate the connected paths between the producer and the three injectors over 400 realizations as shown in Figure 3. Connected paths between a producer and three injectors. The distance between the models is calculated in terms of the Fréchet distances. The distance matrix of five realizations for the path from KB-501 to PROD is given in Table 1. The discrete Fréchet distances between five models for the path from KB-501 to PROD. Models #1 and #5 look similar and #2, #3, and #4 also look similar. However, the two groups have quite different shapes.

Thus, the DFDs between the same group members are small, but the DFDs between the different group members are large. After we calculate all of the DFDs for KB-501 to PROD, KB-502 to PROD, and KB-503 to PROD, the combined distances are calculated as shown in Figure 4. An example for computation of the combined distance. Then the realizations are clustered according to the combined distances between the realizations.

Multi-Dimensional Scaling (MDS) requires high dimension as well as high computational cost for transforming the distances to the metric space. For these reasons, MDS is not appropriate for projection of the 400 models. However, MSD is beneficial for the visualization of the relation between models. Therefore we projected the 20 medoids into 2D space using MDS so that their relation is visualized as shown in Figure 5. Two-dimensional

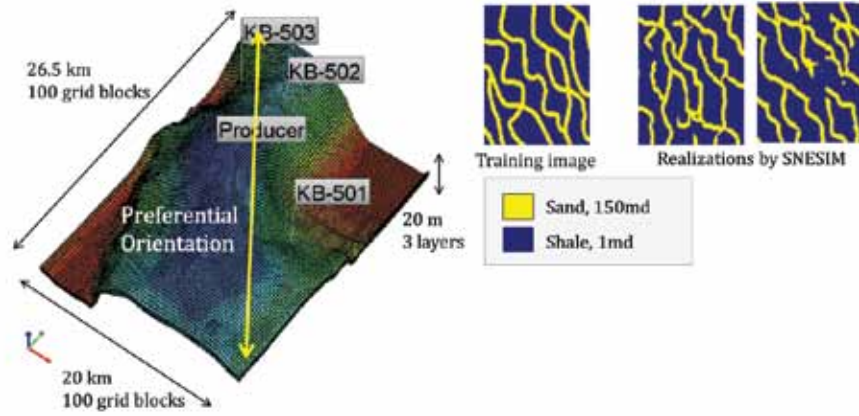


Figure 2. Preferential orientation of reservoir connectivity and a training image of the In Salah field. The color scale shown in the Fig. corresponds to the reservoir top.

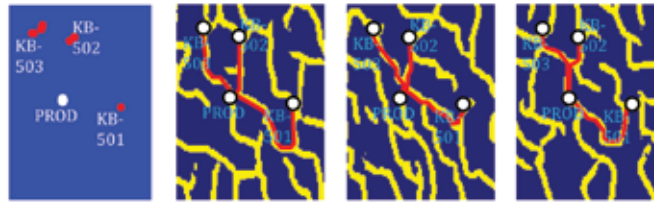
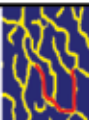
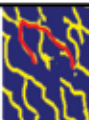
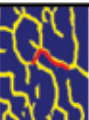
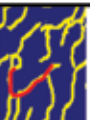
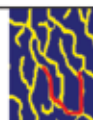
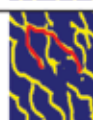
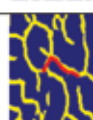
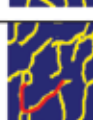


Figure 3. Connected paths between a producer and three injectors.

Table 1. The discrete Fréchet distances between five models for the path from KB-501 to PROD.

Discrete Fréchet Distance						
	#1	#2	#3	#4	#5	
	#1	0.00	39.00	37.34	49.93	15.81
	#2	39.00	0.00	33.94	25.18	39.00
	#3	37.34	33.94	0.00	34.21	23.02
	#4	49.93	25.18	34.21	0.00	49.93
	#5	15.81	39.00	23.02	49.93	0.00

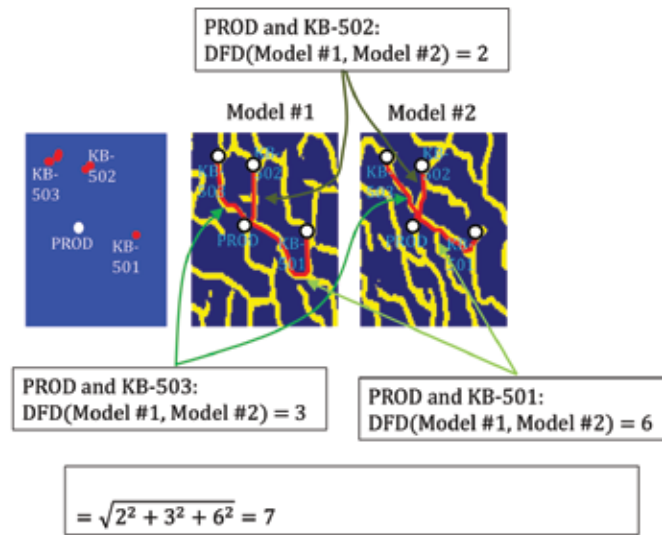


Figure 4. An example for computation of the combined distance.

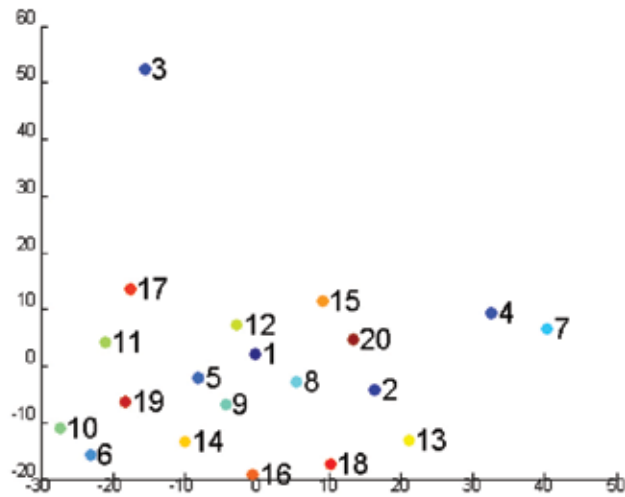


Figure 5. Two-dimensional space of 20 medoids projected by MDS.

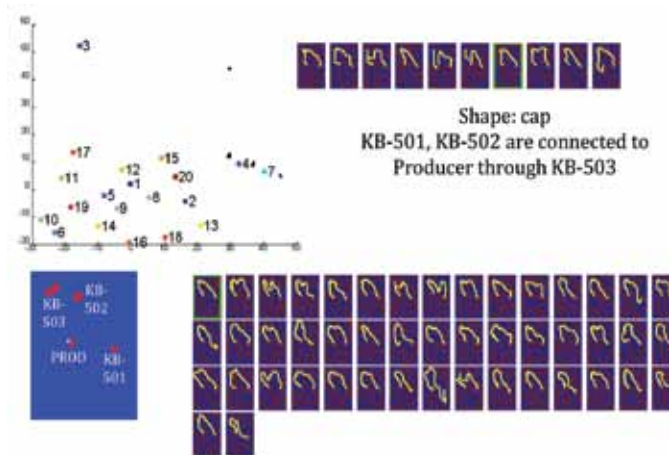


Figure 6. Comparison of cluster #4 and cluster #7.

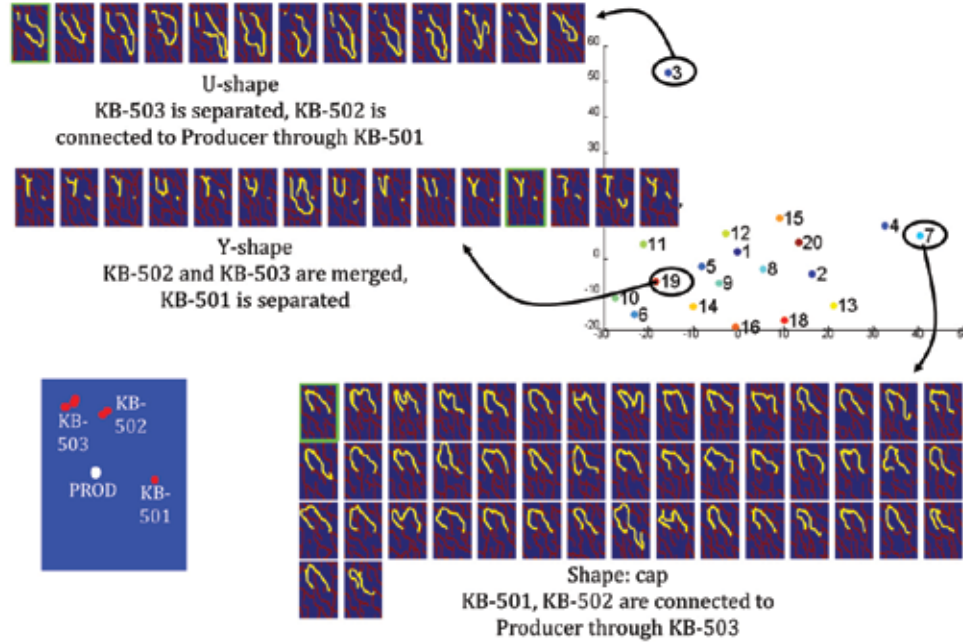


Figure 7. Comparison of cluster #3, cluster #7, and cluster #19

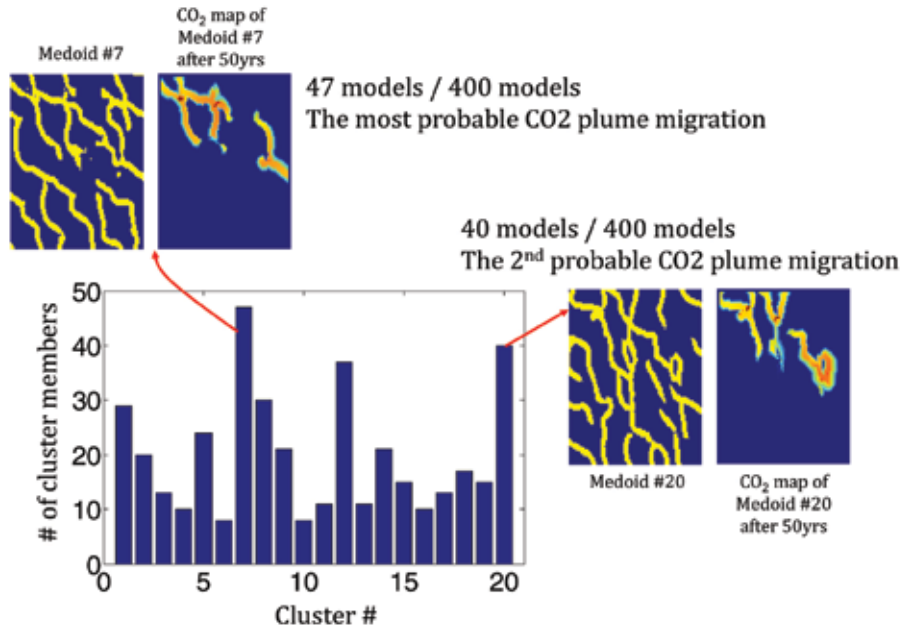


Figure 8. The extracted features of CO₂ plume migration.

space of 20 medoids projected by MDS.. This is only for visualization of the 20 medoids' locations and MDS is not used to cluster the 400 models.

Figure 6. Comparison of cluster #4 and cluster #7. shows the connected paths and the facies map of cluster #4 and cluster #7. The yellow lines in Figure 6. Comparison of cluster #4 and cluster #7. are the connected paths. Since cluster #4 and cluster #7 are very close to each other, shape

of their connected paths should be similar. The shape of their connected paths looks like a cap. KB-501 and KB-502 are connected to the producer through KB-503.

Figure 7. Comparison of cluster #3, cluster #7, and cluster #19 shows the connected paths and the facies map of cluster #3, #7, and #19. They have long distances to each other, so their shape of the connected paths is quite different.

Since we assume the connected paths are similar to CO₂ plume migration paths, we do not need to run a simulator to quantify uncertainty of the CO₂ plume migration. We cluster the models based on the shape of the connected paths instead of the simulation results of CO₂ plume migration to quantify the uncertainty of CO₂ plume migration. Once we cluster the models, we are able to extract the features CO₂ plume migration by simulating the representative models instead of all the models. Figure 8. The extracted features of CO₂ plume migration. shows the two extracted features of CO₂ plume migration. In Figure 8. The extracted features of CO₂ plume migration., the histogram shows the number of cluster members for each cluster. Since cluster #7 has the most number of cluster members, the CO₂ map of cluster #7 can be considered as the most probable CO₂ plume migration. Likewise, the CO₂ map of cluster #20 can be considered as the second probable CO₂ plume migration.

CONCLUSIONS

The paper presents a novel approach to classify prior geologic models on the basis of connectivity of flow paths within the reservoir. Dissimilarity between prior models is computed on the basis of the differences in the characteristics of the shortest connected path between well pairs. The concept of discrete Frechet distance is used for this computation. The dissimilarity measures are grouped using k-medoid clustering. Flow simulation on models belonging to any cluster reveals similarity in CO₂ swept regions in all models belonging to a cluster. However, the current approach has two problems in quantifying the uncertainty of CO₂ plume migration. The first problem is that CO₂ may migrate along disconnected paths, in other words local flow and transport gradients may cause the CO₂ plume to bridge permeability discontinuities. The current

approach compares only connected paths. We therefore should measure path similarity regardless of connection between points of interest. The second problem is that the current approach can be applied to only convection-dominant models such as channel, fractured reservoir models. Thus, development of a similarity measure applicable to general geologic models is necessary.

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Coal Bed Methane Exploration and Possibility for CO₂ Sequestration in Jharia Coalfield, India

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ABSTRACT

Coal proximate parameters, petrography and permeability are key parameters for evaluating the economic production of natural gas from coal seams of Jharia coalfield. Coal quality parameters, including maceral composition, grade and rank, have a significant impact on the source rock and reservoir characteristics of coal. Coal proximate parameters such as ash content, volatile matter and moisture content of these coal seams vary between 17.45% to 33.75%, 15.45% to 25.84% and 0.68% to 1.21% respectively. Coal petrographic analysis, carried out for a selected 16 coal core samples shows that vitrinite reflectance of coal samples varies from 1.21% to 1.75%. Gas content values, calculated from available empirical equation vary from 11.11 cc/gm to 11.91 cc/gm. Considering the actually available experimental data, coals rich in vitrinite macerals is promising for the exploitation of CBM. The injection of CO₂ into coal beds and simultaneous production of CBM combines gas production (enhanced CBM recovery) with CO₂ sequestration results in cleaner energy production. The present work would be useful for CO₂ sequestration while producing CBM from the Jharia coalfield in near future.

INTRODUCTION

Exploration in coal bed methane (CBM) in Jharia coalfield has been going on vigorously for some 15 years. Methane from coal represents an emerging energy resource that has considerable long-term potential for discovery and development. During the formation and maturation of coal, considerable amount of hydrocarbons, mainly methane, is generated by both biogenic and thermogenic processes which in turn are controlled by burial history, maceral composition, and basin hydrodynamics (Scott and Kaiser, 1996). Coal quality parameters, including maceral composition, grade, and rank, have a significant impact on the source rock and reservoir characteristics of coal (Scott and Kaiser, 1996).

There are several case studies which focus on the relationship of coal quality to gas sorption and to quantify the CO₂ sequestration capacity of CBM reservoirs in USA and Canada (White et al., 2005; Gentzis, 2000). Gas content of the recovered coal samples were determined by carrying out canister desorption tests at reservoir temperature at well sites by using the modified direct method of the United States of Bureau of Mines (USBM) (Diamond and Levine, 1981). The other available techniques of estimating gas content from well log mainly uses the density log for determination of ash content and other proximate parameters for the estimation of gas content.

Coal reservoir parameters such as gas content, coal seam permeability, coal seam thickness, critical desorption pressure, reservoir pressure generally affect the CBM production, among which permeability is the main factor

to the migration and flow of gas in CBM reservoir and thus controls CBM production (Paul and Chatterjee, 2011a; Paul and Chatterjee, 2011b). Recent studies of cleat orientation patterns and fracture style in Jharia coalfield yielded insight into coal permeability and CBM production (Paul and Chatterjee, 2011b). Prior information about opening mode fracture/cleat orientation pattern in a coal field is another requirement to plan well locations or horizontal drilling for any CBM project development including CO₂ sequestration.

To assess the coal seam for its CBM potentiality and for future enhanced CBM recovery, coal properties are to be known. The objectives of this paper are (a) to obtain coal chemical properties from proximate analysis of core samples, (b) estimation of coal bed permeability from well log and (c) to assess the possibility for CO₂ sequestration. Although the focus of this work has been on potentiality of coal beds as CBM reservoir, it will also serve as a basis for future assessments of the coal beds for other use such as sites for sequestration of CO₂ with simultaneous recovery of CBM.

STUDY AREA

The Jharia coal field located in Jharkhand state, India is roughly sickle shaped, its longer axis running northwest-southeast. The dip of the Formation in general is southerly (10°). The stratigraphic units of the Jharia coalfield are marked by the presence of two coal-bearing horizons: the Barakar Formation and the Raniganj Formation. Present study area, consisting of Singra, Kapuria, Barki, Dumarda

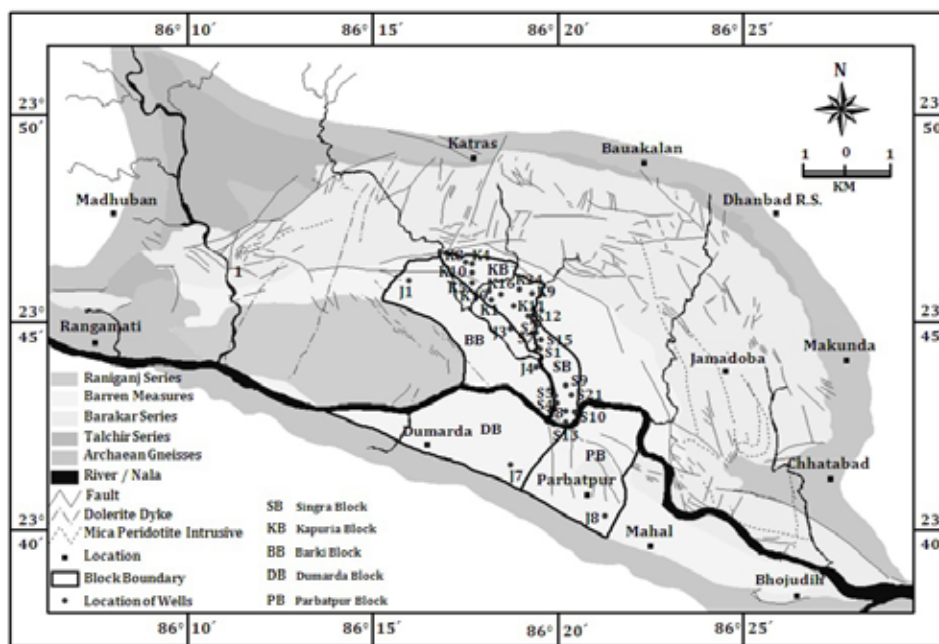


Figure 1. 27 numbers of wells are distributed in Singra, Kapuria, Barki, Dumarda and Parbatpur blocks, Jharia coalfield, India.

and Parbatpur blocks, is located in the central part of Jharia coalfield (Fig. 1). There are 18 major regional coal seams of Barakar Formation of Permian age, which are designated as: A (Bottom seam), B, C, D, E, F, G, H, I, J, K, L, M, N, O, P, Q and R (Top seam). Well logs, litho logs and available coal analysis data of 27 exploratory wells have been used for coal seam correlation (Fig. 1). For most of these wells, it was easy to identify the litho-units such as coal, shaly coal, carbonaceous shale, shale, shaly sand and sandstone from the density versus gamma ray cross-plots.

COAL PETROGRAPHY AND PROXIMATE PARAMETERS

Coal petrographic analysis for 16 samples from 5 wells, namely, S15, K10, K16, K19, and K24 were carried out at the Central Institute for Mining and Fuel Research, Dhanbad following the Bureau of Indian Standard (BIS) procedures. From petrographic study, it is found that the volume of vitrinite ranges from 16.10% to 65.50%, semi-vitrinite ranges from 1.10 % to 6.40%, inertinite ranges from 24.40% to 69.70%, mineral matter ranges from 2.30% to 13.10% and vitrinite reflectance (VRo) (check superscript or subscript) ranges from 1.21% to 1.75% in this study area (Table 1).

Seam-wise proximate analysis for about 80 coal samples collected from 10 wells, namely, S4, S5, S9, S15, S21, K9, K10, K16, K19 and K24, were carried out at the Dept. of Fuel and Mineral Engineering, ISM. The moisture content is representative of the inherent moisture of the coal which ranges from 0.68% to 1.21%. All core samples

have medium to high volatile matter content ranging from 15.45% to 24.84%. The ash content and fixed carbon content of the core samples of the Singra and Kapuria blocks of Jharia coalfield ranges from 17.45% to 33.75% and from 49.48% to 59.76% respectively (Table 2). It can be inferred from proximate analysis that the moisture and volatile matter contents of coals gradually decrease with the increase of depth, whereas ash content relatively increases with depth (Table 2). The volatile matter on dry ash free basis estimated by Rudra and Hazra (2009) indicates that the Barakar coals of Jharia are high volatile 'A' bituminous to low volatile bituminous in rank (Rudra and Hazra, 2009).

Peters (2000) showed that measured gas content of Barakar coal seams around the study area varies from 7 cc/gm to 17 cc/gm (Peters, 2000) whereas other authors like Sahay (2009) showed that gas content of a few coal seams at the Moonidih block ranges 5 cc/gm to over 10 cc/gm (Sahay, 2009). Bhanja and Srivastava (2008) showed that gas content of a few coal seams at Parbatpur block ranges 12 cc/gm to 15 cc/gm (Bhanja and Srivastava, 2008).

ESTIMATION OF COAL PARAMETERS FROM WELL LOGS

Shallow resistivity logs of the 20 wells namely: S1, S2, S4, S5, S7, S8, S10, S13, K1, K4, K5, K8, K10, K11, K12, J1, J3, J4, J7 and J8 have been considered for computation of permeability of 14 major coal seams namely: C, D, E/F/G, H, I, J, K, L, M, N, O, P, Q and R. Coal seams are typically characterized by high electrical resistivities (725 ohm-m to 1750 ohm-m). It had been observed that the resistivity

Table 1: Showing the results of coal petrography analysis from 5 wells of Jharia coalfield, India

Seam Name	Well Name	Coal Core Recovery Depth Interval (m)	(Volume % as received basis)(why colour)					
			Vitrinite (%)	Semi-vitrinite (%)	Liptinite (%)	Inertinite (%)	Mineral Matter (%)	Mean VRo (%)
R	S15	512.05 - 512.20	60.20	3.40	1.80	25.40	9.20	1.27
Q	K19	477.45 - 477.55	65.50	2.90	1.80	25.10	4.70	1.21
P	S15	834.57 - 834.76	50.50	2.30	1.10	37.60	8.50	1.56
O	K16	729.65 - 729.55	52.30	2.70	0.00	39.70	5.30	1.46
O	K19	696.04 - 696.14	63.40	2.60	0.90	29.60	3.50	1.41
N	S15	964.25 - 964.35	44.40	1.23	1.00	40.60	12.77	1.75
M	K10	624.80 - 624.95	61.60	2.40	1.20	29.80	5.00	1.40
M	K24	663.25 - 663.65	58.80	6.40	1.90	24.40	8.50	1.41
L	K24	737.75 - 737.85	60.70	1.90	0.30	30.70	6.40	1.48
K	K10	712.65 - 712.75	58.90	2.10	0.80	33.80	4.40	1.48
J	K24	790.30 - 790.40	46.30	5.60	0.40	39.50	8.20	1.49
I	K10	747.45 - 747.55	57.50	1.90	0.60	35.80	4.20	1.51
I	K16	886.75 - 886.85	46.30	2.10	0.30	48.20	3.10	1.61
H	K24	822.95 - 823.10	42.10	5.10	0.20	45.40	7.20	1.54
C	K16	1074.55 - 1074.70	44.60	1.10	0.20	51.80	2.30	1.67
B	K10	895.85 - 896.00	16.10	1.10	0.00	69.70	13.10	1.71
VRo = Vitrinite Reflectance (why colour)								

Table 2: Seam-wise proximate analysis results for 5 wells of Jharia coalfield, India

Seam Name	Proximate Analysis Results (weight %)(colour)			
	Moisture (%)	Ash (%)	Volatile Matter (%)	Fixed Carbon (%)
R	1.05 - 1.21	17.86 - 26.38	19.50 - 24.84	50.41 - 55.09
Q	1.03 - 1.13	22.31 - 27.39	18.92 - 24.43	50.51 - 54.40
P	0.80 - 1.18	17.45 - 29.73	17.48 - 23.52	50.77 - 58.91
O	0.78 - 1.15	20.83 - 30.26	17.27 - 23.28	49.52 - 54.82
N	0.75 - 1.13	21.31 - 30.87	17.15 - 22.67	49.96 - 54.95
M	0.82 - 1.09	18.52 - 31.35	16.38 - 22.30	50.01 - 58.31
L	0.72 - 1.01	21.70 - 33.75	15.89 - 21.42	49.48 - 55.93
K	0.75 - 0.98	24.85 - 30.90	15.45 - 20.91	50.42 - 54.10
J	0.84 - 0.94	19.20 - 26.63	18.55 - 20.34	53.30 - 59.56
I	0.79 - 0.90	24.85 - 30.82	18.39 - 20.13	53.24 - 59.76
H	0.81 - 0.86	25.43 - 26.63	18.38 - 19.78	53.45 - 54.62
E/F/G	0.72 - 0.93	25.71 - 27.85	17.87 - 18.74	52.52 - 54.73
D	0.68 - 0.89	25.79 - 29.92	17.18 - 18.69	52.02 - 54.73
C	0.70 - 0.87	26.68 - 31.72	16.77 - 17.61	50.65 - 53.84
B	0.85	27.83	17.28	54.04

measured by shallow resistivity logging tool across the coal seam decreases substantially in wellbores filled with high salinity fluids compared to wells with low salinity fluids (Yang et al., 2006). This indicated that the borehole fluid had replaced pore fluid in the cleats (invasion zone). The cleat volume / porosity of cleated coal is given by the previous authors (Paul and Chatterjee, 2011a) as,

$$\text{Cleat volume or Porosity } (\Phi) = 100 \times (0.65 / \text{Resistivity}) \quad 0.6 \dots (1)$$

Using the matchstick model of cleating, initial porosity and initial permeability of coal in the Jharia Coalfield area can be expressed as a function of cleat spacing and aperture (Harpalani and Chen, 1995).

$$\text{Porosity } (\Phi) = 2b/s \text{ and Permeability } (K) = b^3/12s \quad \dots (2)$$

where, b is cleat aperture and s is cleat spacing.

Cleat spacing of 20 mm as observed previously at the underground mines and opencast mines of Jharia coalfield (Paul and Chatterjee, 2011a; Paul and Chatterjee, 2011b) has been used for computation of permeability. Coal seam permeability value ranges from 0.60 md to 1.25 md. There is a significant variation of permeability within the same seam.

POSSIBILITY FOR CO₂ SEQUESTRATION

The recent technology of CO₂ sequestration offers an approach to redirect CO₂ emissions into sinks (e.g., depleted oil/gas fields with enhanced oil recovery, deep saline aquifers, gas rich shales, salt caverns, oceans, and other geological formations) and stabilize future atmospheric CO₂ levels (Gentzis, 2000; White et al., 2005). Coal seams have the ability to store significant amount of gases by adsorbing them. The CO₂ storing capacity within coal seams is much greater than CH₄ storing capacity. The net ratio of injected CO₂ to recovered CH₄ varies from 2:1 to as high as 10:1 (Nelson, 2003; Carroll and Pashin, 2003). This storage mechanism occurs naturally and is well documented, suggesting that a gas, particularly CO₂, can be reliably stored in coal seams. Unmineable coal seams are among the types of geological formations that are being considered by industry and government agencies for use as sites for geological CO₂ sequestration (White et al., 2005). Unmineable coal seams at a depth of around 700 m to 1500 m are suitable for CO₂ sequestration. Many of these coal deposits occur at depths greater than 700 m, which renders them uneconomical for recovery by surface mining or underground mining, and, thus, potentially viable targets for CO₂ sequestration. CO₂ sequestration in unmineable coal seams would most likely only be conducted in areas where natural gas recovery would provide a value-added

revenue stream to partially offset the cost of the CO₂ capture and sequestration processes (White et al., 2005).

It is not anticipated that all coalbed gas reservoirs are suitable candidates for CO₂ (subscript) sequestration or application of Enhanced Coal Bed Methane (ECBM) technology. Carbon capture, CO₂ sequestration and CO₂ related ECBM have been conducted in several regions of the United States, Canada, Europe, Japan, Australia, Poland and China (e.g. Nelson, 2000; Carroll and Pashin, 2003; Damen et al., 2005; Yu et al., 2007; Gurba et al., 2009 and Pinetown, 2013). These laboratory research and field projects have been focused on bituminous or sub-bituminous coals, and their results, although relevant, are not entirely applicable to the lignite coals of the Williston Basin. Permeability and gas sorption capacity are two key characteristics of coal that are known to vary significantly according to coal rank. The results of the Colorado, Alberta, and Alabama projects, therefore, have limited applicability to lignite coals. With this in mind, permeability and sorption isotherm tests are needed to be conducted in order to determine if CO₂ injection into a coalbed would be feasible and would enhance CH₄ recovery from a coalbed reservoir.

Considering the actually available experimental data, coals rich in vitrinite macerals (> 50%) are promising for the exploitation of CBM. Production of CBM by injecting CO₂ in coal beds is one of the popular methods of enhanced recovery of CBM (ECBM). The coal seams occurring at a depth of more than 700 m, thickness of more than 1 m and permeability of about 1 md with gas content more than 10 cc/gm may be considered for planning of any ECBM project. The present work may be useful for CO₂ sequestration while producing CBM from the Jharia coalfield in near future. Though it is beyond the scope of the present work, it is suggested to conduct adsorption isotherm study in future before going for CO₂ sequestration.

CONCLUSIONS

The Barakar coal seams under the study area, as revealed from the proximate and petrographic analysis, are of medium volatile to high volatile bituminous "A" in rank, ash content ranging from 17.4% to 33.7%, VRo% ranging from 1.21% to 1.75%, reflecting the maturation of coal and prospecting for CBM exploration/exploitation. Permeability of the 14 major coal seams of the study area has been estimated by using well logs. As significant variation of permeability is observed within the same seam, it may be used as a guide to locate potential CBM areas for exploration. Predicted coal seam permeability values ranging 0.65 md to 1.25 md are in good match with the well test permeability data (Peters, 2000; Sahay, 2009) adjacent to the study area. The findings summarized here, are useful for CO₂ sequestration and ECBM studies.

ACKNOWLEDGEMENT

The authors express their sincere gratitude to Coal India Limited (CIL) and Central Mine Planning and Design Institute Limited (CMPDIL), Ranchi for the financial assistance under its R&D scheme.

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Geochemical Modeling and Experimental Studies on Mineral Carbonation of Primary Silicates for Long-term Immobilization of CO₂ in Basalt from the Eastern Deccan Volcanic Province

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ABSTRACT

Deccan flood basalt records immense accumulation (1.5×10^6 km² area) of tholeiitic magma in a relatively short time span. For mineral carbonation study and long-term storage of CO₂, massive tholeiitic basalt from the Mandla lobe of the eastern Deccan volcanic province was considered as it contains a high amount of reactant minerals such as Ca, Mg and Fe rich silicates. Main objective of the present study is to understand effects of CO₂ concentration on conversion of Ca, Mg and Fe bearing silicate minerals in to stable carbonate minerals such as calcite, dolomite, magnesite and siderite. Computer based computation details of chemistry of basalt-water-CO₂ interaction, under laboratory induced hydrothermal-like conditions, form the basis of the treatment of basalt specimens. Grain surface area value = 23,000 cm²/g was maintained throughout the experiments. A series of experiments were performed at accelerated conditions of 5 and 10 bars *p*CO₂, while vessel pressures were maintained at 10 and 20 bars at 100 and 200 °C temperatures for 50, 60, 70 and 80 hours, respectively. XRD and SEM-EDS results show presence of calcite (Ca_{0.29} C_{0.79} O₆), aragonite (Ca_{0.43} C_{1.19} O₉), siderite (Fe_{0.45} C_{0.79} O₆) and magnesite (Mg_{0.11} C_{0.79} O₆) in the treated samples. It is also observed that mineral carbonation accompanied by dissolution reactions led to rise in the pH (7.42) of the solution, when treated for 80 hours at 100 °C temperature and 5 bar *p*CO₂ pressure. A sudden decrease in the pH (6.85) is also noticed in case of a specimen treated similarly, but at elevated (10 bar) *p*CO₂. Bulk of the neo-formed secondary product is mainly composed of clay minerals. Present experimental results correspond largely to those obtained from kinetic modelling of basalt-water-CO₂ interaction. Results indicate presence of similar carbonate minerals such as calcite, aragonite, siderite and magnesite. Details pertaining to mineral carbonation related phase transformations are discussed in the paper.

INTRODUCTION

Carbon dioxide is one of the most important greenhouse gases in Earth's atmosphere. Changes in *p*CO₂ concentration have significant effects on the surface temperature of the Earth (Houghton and Wood, 1987), which leads to major deviation in the climate, surface processes and biota. Marini (2007) had stated that soon after the onset of the industrial revolution, atmospheric CO₂ concentration increased considerably with time (for the past 45-50 a.) attaining anomalous values (e. g. Oelkers and Cole, 2008 - CO₂ concentration in Earth's atmosphere has increased from ~ 325 ppm in 1970 to 380 ppm at the beginning of this century). Rochelle et al., (2004) have discussed a variety of CO₂-water-rock chemical reactions to safely trap CO₂, suitable for geological time-scales. Geologic storage (Bachu and Adams, 2003; IPCC, 2005) and other options of CO₂ sequestration were exercised in the past (Brennan and Burruss, 2003; Xu et al., 2005) to reduce CO₂ emissions (Metz et al., 2005). Massive basalt flows contain abundant Ca-Mg-Fe rich silicate minerals (occur all around the world). They play a significant role in the global carbon cycle (Brady and Gislason, 1997; Retallack, 2002). Large igneous provinces such as continental flood

basalts, volcanic passive margins and oceanic plateaus (Best, 2001) represent immense accumulation of mafic (Fe and Mg rich) magmas. Presence of abundant quantities of olivine, pyroxene and plagioclase (that are rich in Mg or Fe or Ca contents, respectively) were reported (Ahmad, 2002) from Mandla lavas. They provide a suitable matrix for CO₂ sequestration. Charan et al., (2010) have suggested suitability of the Igatpuri picritic basalt flows for CO₂ sequestration. Holloway et al., (2008) have stated that India has potential sites for large scale geological CO₂ storage. Conversion of CO₂ gas into stable carbonate minerals such as calcite, dolomite, magnesite and siderite is considered as one of the best options for long-term storage of CO₂ (Oelkers et al., 2008). Studies by Broecker (2008), on capturing and storing of CO₂, have revealed that the layered basalt provinces such as Deccan Traps in India (Fig. 1) offer large area storage depot for captured CO₂. However, such possibilities are yet to be tested in detail.

To predict or model long-term CO₂ sequestration in Deccan basalts, mineral carbonation mechanism requires to be quantified by resolving key parameters, such as *p*CO₂ (in ppmv), temperature and time. The main objective of the present study is aimed at in better understanding of silicate mineral carbonation reactions under natural and laboratory

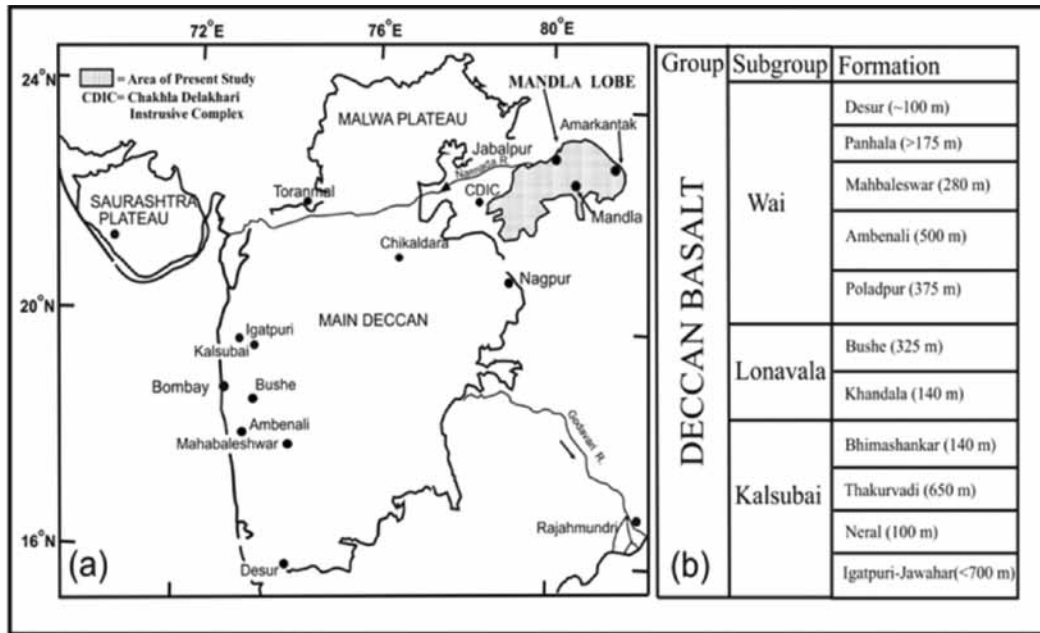


Figure 1. Map showing outcrop of the Deccan volcanic province and location of the Mandla lobe (with grey shade) of the eastern Deccan volcanic province, and (b) stratigraphic succession of the western Deccan province.

induced accelerated pressure and temperature conditions. These reactions between CO₂ and primary silicate minerals such as olivine, pyroxene and plagioclase (as reactants) and secondary carbonates help to understand the rate and extent of mineral carbonation. The resultant pathways will be utilized in understanding the reactive phases suitable for mineral trapping of supercritical CO₂ reactions under variable P, T, S/V, porosity and permeability conditions.

MANDLA LOBE: A LABORATORY FOR CO₂ SEQUESTRATION

The Mandla lobe, an isolated, large (29,351 km²) north-eastern outlier of the Deccan Traps, extends E-W for 344 km and 156 km N-S around the towns of Seoni, Jabalpur, Mandla, Dindori and Amarkantak areas (Fig. 2). The landscape is covered with flat-topped plateaus (commonly known as *macula plateau*) and ridges that often contain small mesas and buttes. Major topographic breaks occur at elevations of about 450, 600 and 900 m a. s. l. (Fig. 2), where duricrusts of laterite often cap the plateau above these breaks and provide several stratigraphically important lava flow exposures of the lava pile. Tholeiitic lava flows from the Mandla lobe of the eastern Deccan basalt province are considered in the present study because of their large and continuous areal extent, large combined thickness of the flows (>1 km in some localities), inter and intra-flow features, abundance of reactive silicate minerals and Mesozoic sediments containing Fe-Mg-Ca silicate minerals. On the basis of regional mapping, physical characters, petrography and lateral tracing of the lava

flows, Pattanayak and Shrivastava (1999) have recorded the presence of thirty-seven lava flows in the ~ 900 m thick volcano-sedimentary sequences around Mandla. These were characterized further by their major element chemistry (Shrivastava and Pattanayak, 2002). Major-oxide chemistry was used for further grouping them in to eight chemical types (Shrivastava and Ahmad, 2005a). Modal data for each of the mineral phases and glass show vast variability in the modal percentages of pyroxene (65.31-18.27 %) and plagioclase (47.00-19.00 volume %), though, these flows contain < 8.29 % olivine and the subordinate amount of Fe oxide and glass. Distribution of cations on M₁ and M₂ sites, together with Mg: Fe: Ca ratios for pyroxenes show presence of Ca-rich and Ca-poor pyroxenes; however the later type of the pyroxene present in this area is either Fe-rich or Fe-poor in composition. The Ca-poor, but Fe-rich pyroxenes (Ca: Mg: Fe = 24: 22: 54 and 22: 14: 64, respectively) are associated with the commonly found high diopside and olivine normative tholeiitic lava flows. Most of the Ca rich microphenocrysts are mainly ferro-augite. Sub-calcic augite and pigeonite are also common in this area. The Fe rich mineral phases in lava flows are mainly ilmenite-magnetite, olivine and pyroxenes. The Fe/Fe+Mg for pyroxene (both phenocrysts and microphenocrysts), olivine and whole rock data show wide variation. Distribution of cations in W and Z sites of the plagioclase structure and Na (as Ab): Ca (as An): and K (as Or) ratio when calculated for the Mandla lavas show poor An content in the plagioclase reflecting a more sodic composition. The co-existing magnetite-ilmenite in 37 lava flows represent a wide variation in their FeO (47.14-

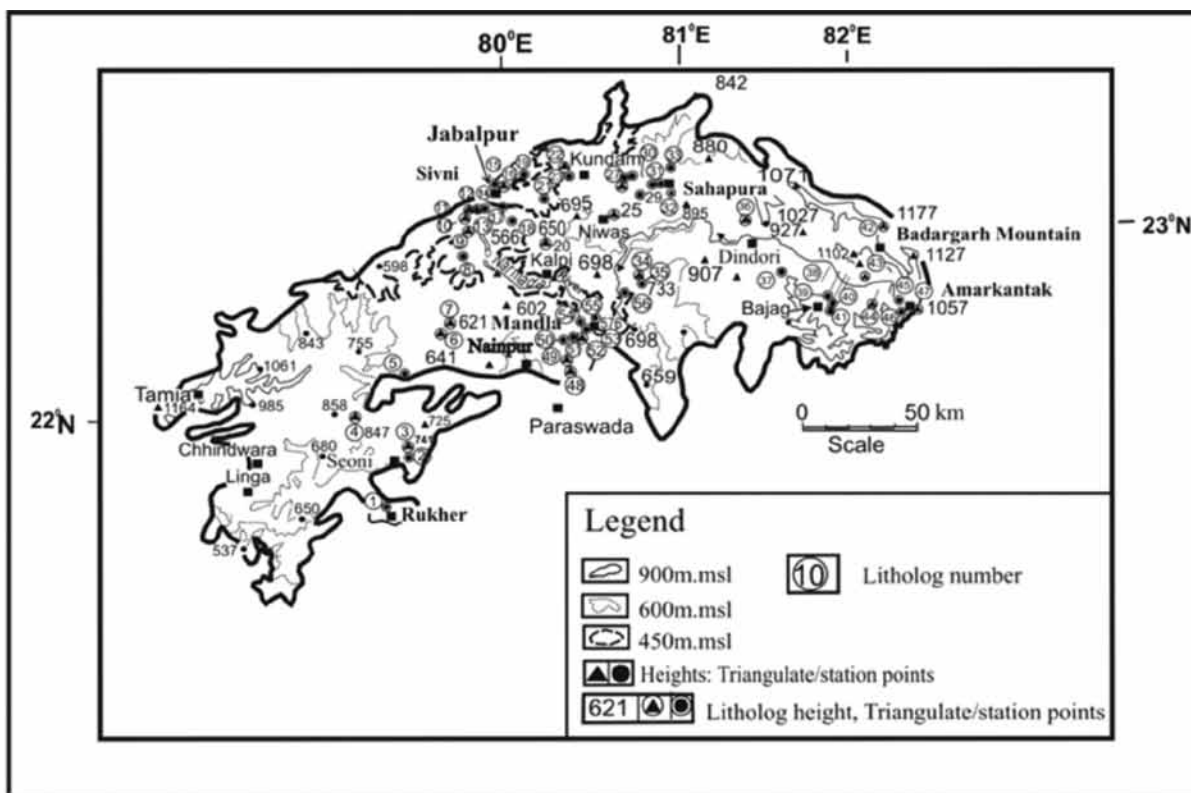


Figure 2. Map showing topographic details of the area of study - Mandla lobe of the eastern Deccan volcanic province

56.59), Fe_2O_3 (14.41-38.44) and TiO_2 contents (Ahmad and Shrivastava, 2004). Combined trace elemental and isotopic compositions for basaltic lava flows of the Mandla lobe show chemical affinities with Poladpur, Mahabaleshwar and Ambenali formation basalts of the distant south-western Deccan volcanic province (Shrivastava et al., 2014).

CARBONATE FACIES: EVIDENCE OF GEOLOGICAL SEQUESTRATION OF CO_2

Sufficiently thick, sedimentary carbonate units (sandwiched between lava flows), developed in the hiatuses between two successive eruptive events suggest relatively less violent eruptive activity in Mandla area as compared to those that occurred in its western counterpart. In this area, significantly thick limestone and inter-calcareous lithofacies constitute major part of the infra/inter-traps (Fig. 3). It contains a variety of carbonate lithologies - limestone, brecciated calcrete, calcareous siltstone with calcareous nodules, nodular and pisolitic calcrete, calcrete nodules, marl and calcified sandstone. Presence of secondary minerals such as siderite, calcite, aragonite, quartz, chalcedony, smectite, chlorite, zeolite, Mg-Fe-Al hydroxides and clay minerals associated with basalt alteration are largely an effect of CO_2 concentration. Fine to medium grained clay, carbonate and quartz-rich matrix occurs at several places. Carbonaceous laminations at several places

are also noticed. White to buff coloured intertrappean limestone beds vary in thickness from few cm. to ~ 8 m, though their lateral continuity is inconsistent, forming lenses and discontinuous beds. At places, limestone circular micritic pellets are also observed. The limestone contains microcrystalline and chalcedonic chert, formed by precipitation of silica saturated hydrothermal solutions. The clay mineral phases in the infra-/inter-trappean beds and weathered Deccan basalt show dominance of montmorillonite, minor amounts of chlorite and illite (Salil, et al., 1994). Association of smectite with the carbonate facies rich in Fe, Mg and Al is observed (Salil, et al., 1996). These rocks show Mg, Fe substitution for Al in smectite (Salil et al., 1997; Shrivastava and Ahmad, 2005). But, clay minerals present in the intra-volcanic bole horizons formed during hiatuses between two volcanic episodes, contain Fe-rich smectite and most of these clays are dioctahedral type, showing a balance between net layer and interlayer charges (Ahmad and Shrivastava, 2007). Kaolinite contains silica in which Fe^{3+} substitutes for Al^{3+} at the octahedral site (Shrivastava, et al., 2012). The secondary minerals formed during low temperature (<150 °C) basalt alteration, occur between the different Deccan volcanic episodes. The mineral assemblages developed depend upon the extent of alteration and temperature that existed at the initial stage of basalt alteration. Primarily, opal/chalcedony/quartz, allophone, kaolinite/halloysite are



Figure 3. Development of carbonate facies between two lava flows - CO₂ sequestration in the past represented by thick limestone and inter-calcarious lithofacies, calcrete, calcified siltstone and marl.

formed, but with increased degree of alteration, formation of more complex smectite, zeolites and carbonates takes place at the expense of SiO₂ rich minerals (Steffansson and Gislason, 2001; Steffansson, 2010). In an environment where high concentration of CO₂ is available, zeolites and clays undergo replacement by the carbonate solid solutions (siderite, dolomite, and magnesite) and quartz (Rogers, et al., 2006). Nature of intercalated weathering and sedimentary horizons are common within continental basalt successions (Widdowson et al., 1997; Jerram, 2002), where, high atmospheric *p*CO₂ (1110-1850; mean value of 1480 ppmv) often existed during the Deccan volcanic activity [from 70 m. y. (Late Cretaceous period) to 63 m.y. with a major event at 65 ± 0.3 m.y. (Salil, et al., 1997; Chenet, et., 2007; Shrivastava, et al., 2012)] thus, making the possibility of replacement of clays by carbonates highly likely leading to the development of carbonate facies, though these assumptions need to be verified experimentally.

MINERAL CARBONATION REACTIONS AND SEQUENCE OF MINERAL FORMATION

Experimental and natural studies (McGrail et al., 2006; Matter et al., 2007; Oelkers et al. 2008 and Prasad et al., 2009) on basalt-water-CO₂ chemical reactions have concluded that water charged with abundant CO₂ reacts

with basalt and releases the Mg bond of pyroxene and olivine. This in turn combines with the carbonates to form highly stable MgCO₃ (magnesite), whereas water with lower quantity of CO₂ reacts with plagioclase, releases Ca and forms CaCO₃. However, several important issues related to - (a) what fraction of CO₂ is to be injected into the basalts, which would react before it finds its way back to the surface, (b) do the secondary carbonate minerals so formed clog the plumbing and (c) will alteration by-products such as silica, clays, zeolite coat the surfaces and slow down the reaction rate between the CO₂ charged water and the rock (Table 1) remained unanswered.

Aqueous mineral carbonation process enhances the extent of reaction as well as the rate of conversion-activation of the mineral reactants. Mineral activation follows two paths - (a) size reduction to increase relative surface area and (b) destruction or disordering the crystal lattice to form an amorphous material. For carbon sequestration or mineral carbonation, initial studies of O'Connor et al. (2002) consisted of experiments to substantiate carbonation reaction $[(\text{Mg}_2\text{SiO}_4 \text{ (Olivine)} + 2\text{CO}_2 \text{ (Carbon dioxide)}) = 2\text{MgCO}_3 \text{ (Magnesite)} + \text{SiO}_2 \text{ (Silica)} + 54\text{kJ}]$ of natural olivine. The aqueous mineral carbonation reaction sequence includes early-stage precipitation of Mg carbonate, which is co-eval with silicate dissolution. Mani et al. (2008) and Kelemen and Matter (2008) have discussed reactions, wherein CO₂ is stabilized

Table 1. Summary of the experimental data pertaining to effects of CO₂ on the rate of dissolution of silicate minerals (modified after, Golubev et al., 2005)

Minerals	pH	CO ₂ effects	References
Forsterite	4-11	Absent	Golubev et al. (2005)
Olivine (Fo92)	4-6	Absent	Pokrovsky and Schott (2000b)
Olivine (Fo92)	>9	Inhibition	Wogelius and Walther (1991) Pokrovsky and Schott (2000b)
Olivine (Fo92)	10.5-11	Absent	Golubev et al. (2005)
Augite	4	Absent	Brady and Carroll (1994)
Labradorite	3.2 ^a	Absent	Carroll & Knauss (2005)
Anorthite	4	Absent	Brady and Carroll (1994)
Anorthite	5.5-8.5	Acceleration	Berg and Banwart (2000)
Basalt glass	8	Absent	Brady and Gislason (1997)

Table 2. Thermodynamic equilibrium constants at specified temperatures and total pressures of the chemical reactions used to construct the log-log diagram of Ca/Mg activity ratio vs. fugacity.

Reactions	Log K values	
	100°C 1.013 bar	150°C 500 bar
Calcite + Mg ²⁺ + H ₂ O = Brucite + Ca ²⁺ + CO _{2(g)}	-3.3197	-1.6714
Dolomite + Mg ²⁺ + 2H ₂ O = 2Brucite + Ca ²⁺ + 2CO _{2(g)}	-8.0936	-4.9137
Magnesite + H ₂ O = Brucite + CO _{2(g)}	-3.5065	-2.1415
2Calcite + Mg ²⁺ = Dolomite + Ca ²⁺	1.4542	1.5709
Dolomite + Mg ²⁺ = 2Magnesite + Ca ²⁺	-1.0806	-0.6307
2Dolomite + Mg ²⁺ = Huntite + Ca ²⁺	-3.5007	-2.6995
Huntite + Mg ²⁺ = 4Magnesite + Ca ²⁺	1.3395	1.4381
2Calcite + Mg ²⁺ = dis-Dolomite + Ca ²⁺	0.3416	0.6615
Disordered-Dolomite + Mg ²⁺ = 2Magnesite + Ca ²⁺	0.032	0.2787
2Disordered-Dolomite + Mg ²⁺ = Huntite + Ca ²⁺	-1.2755	0.8807

by reacting with olivine [(Mg₂SiO₄ Forsterite/Olivine + 2CO₂ → 2MgCO₃ Magnesite + 2SiO₂ Silica)], pyroxenes [(Fe₂Si₂O₆ Ferrosilite/Pyroxene + 2CO₂ → 2FeCO₃ Siderite + 2SiO₂ Silica)] and plagioclase [(CaAl₂Si₂O₈ Anorthite / Plagioclase + CO₂ + 2H₂O → CaCO₃ Calcium Carbonate + Al₂Si₂O₅(OH)₄ Kaolinite]] and forms the stable carbonate minerals - magnesite, siderite and calcite. Addition of water during mineral carbonation facilitates formation of calcite and magnesite [(Mg₂SiO₄ Mg-olivine + CaMgSi₂O₆ Ca,Mg-Pyroxene + 2CO₂ + 2H₂O → Mg₃Si₂O₅(OH)₄ Serpentine + CaCO₃ Calcite + MgCO₃ Magnesite)], as CO₂ combines with water and forms HCO₃⁻. However, limited work has been carried out on these reactions using P, T, and time variables.

Detailed field and laboratory studies (including modelling under varied, but controlled physical and chemical parameters of near natural conditions) are required to be undertaken on basaltic lava flows in the eastern Deccan volcanic province. Sequestration by mineral carbonation leads formation of solid carbonates, involving dissolution of primary silicates and precipitation of neo-formed phases, mainly carbonates, opaline silica and

clay minerals. Thermodynamic equilibrium constants at particular temperatures and pressures of chemical reactions are summarized in Table 2.

Mineralogical studies of Crovisier et al., (2003) indicate that an alteration layer, primarily of palagonite forms over the surface of basaltic glass in response to chemical attack by water. Under certain conditions, such as prolonged reaction time and high temperature, the alteration layer formed consists of an amorphous gel-like material, leading to the suspicion of hydration of glass due to water permeation and alkali inter-diffusion. In some cases, the alteration layer is crystallized to some extent and contains clay minerals (smectite). Such layers are formed mainly by the process of co-precipitation of the elements dissolved from the basalt. In addition to these hydrated aluminosilicates, hydrated residual glass is also present, formed as a consequence of inter-diffusion processes. In this scenario, an integrated study on carbonation mechanism, mineral paragenesis and application of geochemical codes for modeling is required.

GEOCHEMICAL CODES AND THEIR APPLICATION

Quantification of basalt carbonation and analysis of macro-chemical processes require application of geochemical reaction codes that help in simulation of the mineral carbonation products formed during CO₂ sequestration reaction with basalt. A geochemical code is defined as the integration of mathematical expressions, describing theoretical concepts and the thermodynamic relations on which aqueous speciation, oxidation-reduction, precipitation-dissolution, and adsorption-desorption calculations are based. These codes are considered efficient when they encompass all the requisite sub-models and the imperative aqueous complexes, solids and gases for significant elements of preference to interpret a provided data-set adequately. Extensive databases of thermodynamic property values and kinetic rate constants are required for these codes.

Geochemical codes (Cited Table 5.1; Allison et al., 1991) used for speciation and solubility calculations are more pertinent to such work and include- MINTEQA2 (Allison et al., 1991), EQ3/6 (Wolery, 1992), PHREEQC (Parkhurst and Appelo, 1995), REACT (Bethke and Torgersen, 1998) and KINDIS. These chemical reaction codes are divided into two main groups such as (a) aqueous speciation solubility codes and (b) reaction path codes. The former group of codes such as WATEQ, REDEQL, GEOCHEM, MINEQL, MINTEQ, and their later versions have used in the calculation of aqueous speciation/complexation and the degree of saturation of the speciated composition of the aqueous solution with respect to the solids in the code's thermodynamic database. They have also capabilities to calculate mass transfer between a single initial and final state that results from mineral precipitation/dissolution and adsorption/desorption reactions. Besides the above mentioned features, the reaction path codes, such as PHREEQE, PATHCALC, and the EQ3/EQ6 series of codes also permit simulation of mass transfer due to mineral precipitation as a function of reaction progress. Typical applications include modeling of chemical changes associated with the interaction of minerals and groundwater (Delany et al., 1985) as a function of time. At each step of reaction progress, the code calculates the changes or path of mineral and gaseous solubility equilibrium, the masses of minerals precipitated or dissolved to attain equilibrium and the resulting composition of the aqueous solution. The application of geochemical code for modeling of silicate mineral carbonation mechanism is summarized (Fig. 4). The EQ3/6 release 3230B software package is a set of computer programmes and supporting data files / databases - used in the modeling of complex geochemical processes that take place when aqueous solution reacts with basalt silicates and CO₂, due to which new carbonate

minerals are formed. The code PHREEQC (version 2) written in C programming language has been designed to perform a wide variety of low-temperature aqueous geochemical calculations. PHREEQC is based on an ion-association aqueous model. The MINTEQA2 code is used in combination with a thermodynamic database to calculate complex chemical equilibrium among aqueous species, gases & solids and between dissolved and adsorbed states. Theoretically, this code has four sub-models - (a) aqueous speciation, (b) solubility, (c) precipitation/dissolution, and (d) adsorption. It also incorporates a Newton-Raphson iteration scheme to solve the set of mass-action and mass-balance expressions. The geochemical code REACT is a set of software for manipulating chemical reactions, calculating stability diagrams and the equilibrium states of natural waters, tracing reaction processes, modeling reactive transport in one and two dimensions and plotting the results of these calculations. It can integrate kinetic rate, kinetic rate law and simulate the fractionation of stable isotope in a reacting system. The reactive transport model is a groundwater flow transport model coupled to a chemical reaction model. The KINDIS (version 1) programme is used to calculate speciation of the aqueous species and to simulate the irreversible dissolution of minerals and reversible precipitation of secondary phases.

To understand thermodynamics, kinetics and reaction path of minerals formed, parameters such as solubility constant, saturation index, log activity and thermodynamic properties of minerals (Gibbs free energy, enthalpy and entropy) are required to be considered. Geochemical reaction models require thermodynamic data to calculate geochemical reactions associated with CO₂ sequestration. This is an important data requirement for successful application of geochemical reaction modeling for CO₂ sequestration (Marini 2007). The availability of thermodynamic data for CO₂ and carbonate aqueous species and minerals is in the database files used in geochemical reaction models. It is focused on issues related to computer modeling of geochemical reactions associated with CO₂ sequestration in basic igneous rocks such as basalt and peridotite. These codes differ in their ease of use, but can accurately solve the equilibrium assemblage of mineral solubility, within the limits of thermodynamic databases. The quality of the result of each of these codes is directly related to the quality of these databases. These databases facilitate creation of phase diagrams, describing stability of mineral phases as a function of temperature and pressure. The utility of thermodynamic database leads to their incorporation into "user friendly" chemical speciation, reactive path, and reactive transport using computer codes, such as EQ3/6(Wolery 1983), PHREEQC (Parkhurst and Appelo 1999), and CHESS (Vanderlee et al., 2002). These codes allow rapid calculations of mineral solubility and solute speciation in a variety of geochemical systems. The

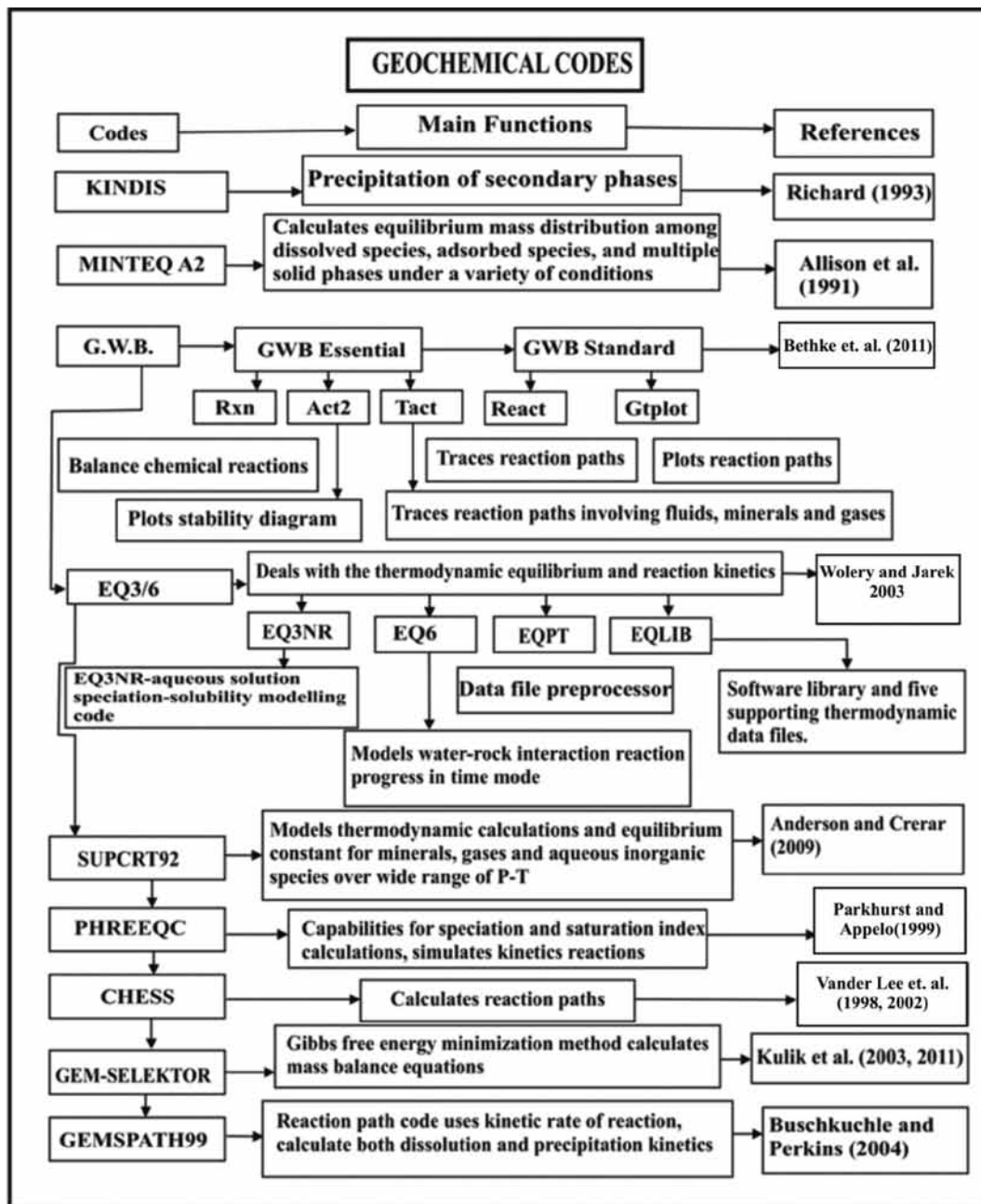


Figure 4. The application of geochemical codes for modeling of silicate mineral carbonation reaction modelling

computer algorithms allow calculation or prediction of the equilibrium state and or the evolution of geochemical systems as a function of reaction progress.

CARBONATION REACTION MODELLING

To predict the course of chemical reaction that takes place during CO₂ sequestration, building of thermodynamic and kinetic model is necessary. Owing to high MgO, CaO contents, the Deccan flood basalt possesses high potential

for CO₂ sequestration, which is further enhanced by the presence of significant amounts of Na₂O and FeO contents. Out of 37 physically and chemically distinct lava flows that occur in the Mandla lobe of eastern Deccan volcanic province, whole rock chemical data for the 4th lava flow (well exposed in the Barasimla and other sections around Jabalpur area) is considered (Table 3) as a starting phase of chemical analysis (cited Fig. 1, 2 and Table 5 from Pattanayak and Shrivastava, 1999 for lava flow location and chemical data, respectively).

Table 3 Composition of basalt (in wt. %)

		Elements	
SiO₂	48.64	Si	17.26
TiO₂	2.83	Ti	0.59
Al₂O₃	14.88	Al	5.51
CaO	10.02	Ca	2.51
Fe₂O₃	14.85	Fe	2.65
K₂O	0.34	K	0.08
MgO	5.77	Mg	2.37
Na₂O	2.77	Na	1.20
P₂O₅	0.19	P	-

- Not considered.

Olivine [chlorophaeite pseudomorphs after olivine; Stokes (1968)], pyroxene, plagioclase, magnetite mineral phases and glass are present in this lava flow. The modal data (for 4th lava flow) determined (on volume percent basis) for pyroxene (Py.), plagioclase (Pl.), magnetite (Ma.), chlorophaeite (Ch.), and glass (Gl.) is 44.75, 40.74, 9.55, 4.75 and 0.25 percent, respectively. These values were multiplied by their respective density values (Py.-3.2, Pl.-2.6, Ma.- 5.2, Ch.-2.6 and Gl.-2.6) to calculate weight percent of pyroxene, plagioclase, magnetite, chlorophaeite and glass, respectively. Mineral phases such as ilmenite and apatite were neglected from subsequent data elaboration because of their low amount and chemical durability. Whole rock composition was recalculated so that the sum of the molar fractions of pyroxene, plagioclase, magnetite, chlorophaeite, and glass is equal to 1. By mass balance calculations, the values obtained for molar fractions (in %) of all the phases of pyroxene, plagioclase, magnetite, chlorophaeite and glass are 0.4600, 0.3387, 0.1575, 0.03972 and 0.00209, respectively. Rock composition was then recomputed assuming that the sum of the molar fractions of minerals is equal to 1. Using major oxide data (Table 2), mineral formulae were calculated for pyroxene [(Ca0.23Mg0.19Fe0.24)Si1.09O3], plagioclase (Ca0.62Na0.25Al1.02Si2.85O8), magnetite (Fe0.28O4), olivine [(Mg0.29Fe0.38)2Si1.66O4], chlorophaeite [(Ca0.47Mg0.38Fe0.49)Fe0.49Si2.15O10(OH)8] and glass (Si0.042Al0.05Fe0.003Mg0.019Ca0.032Na0.013K0.008O3.3) present in this lava flow. Based on mineral formula, cation values (in %) were calculated (Table 4).

The total surface area in contact with water with 1000 ml of water was set to 250cm²/g in the case of Mandla basalt. It was distributed in proportion to the volume percentage of pyroxene, plagioclase, magnetite, chlorophaeite, and glass (Table 4). Their initial amounts in the considered system were calculated through geometric calculations assuming an initial porosity of 0.3. Computer

simulation was performed through EQ3/6 software package of Wolery (1992), maintaining 4936.6 x10⁵ (5 bar pCO₂ pressure) and 9873.2x10⁵ (10 bar pCO₂ pressure) ppmv concentrations of CO₂, specifying the initial surface areas of primary minerals (which are changed by the code in proportion to their remaining masses) and describing their dissolution kinetics (100 - 800 hours for glass, 100-900 hours for plagioclase and 100-1000 hours for pyroxene) at 100 and 200 °C temperatures, respectively. The reaction progress (log Q/K values) obtained for plagioclase, pyroxene and glass (Tables 5a-c) when plotted (Fig. 5a-d) show the formation of carbonate minerals. In the case of plagioclase, the rate of formation of calcite and aragonite is slow at the beginning, but, after 100 hours, reaction rate increases and attains a steady state beyond 1000 hours. In the case of pyroxene (Fig. 5b) and glass (Fig. 5c), the carbonation reaction starts initially at a high rate and there is a slight increase in the rate of formation of calcite and aragonite. However, with progression of time, it lowers down considerably in the case of magnesite, siderite, dolomite and huntite. Standard molar thermodynamic properties for these mineral species were compared with the published values of Gibbs free energy, enthalpy and entropy of formation (Table 6). Values obtained through simulation are significant and considered for the present experimental work.

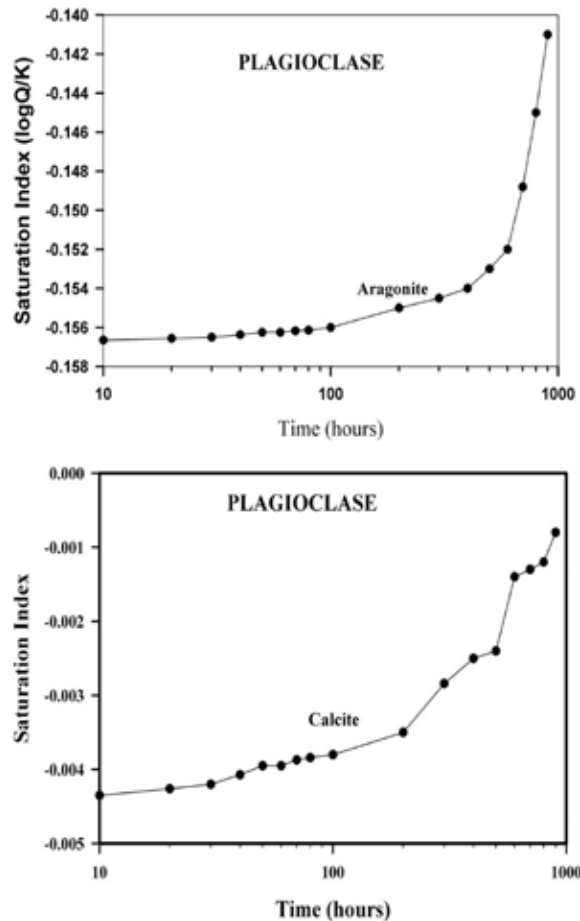
EXPERIMENTAL WORK

The experiments involved basalt samples in which major oxides (SiO₂= 48.64, TiO₂= 2.83, Al₂O₃=14.88, CaO= 10.02, F₂O₃= 14.85, K₂O= 14.85, MgO=5.77, Na₂O=2.77 and P₂O₅= 10.19) in weight percent were analyzed using XRF technique. To achieve laboratory induced carbonation of silicate minerals under hydrothermal-like conditions and to comprehend the carbonation reaction mechanism, the samples were powdered and passed through 140-170 ASTM sieves and cleaned with deionized water and acetone in an

Table: 4 Mineral formula and cations calculated from major oxides.

Deccan Basalt	Mineral phases and glass				
	Chlorophaeite	Magnetite	Clinopyroxene	Ca-Plagioclase	Glass
Modal % (vol. basis)	4.75	9.55	44.75	40.74	0.25
Densities (g/l)	2.6	5.2	3.2	2.6	2.4
Oxides (wt %)	12.35	49.66	143.2	105.9	0.65
Cations					
Ca	0.00087	-	0.02227	0.016484	0.00002
Mg	0.00069	-	0.0184	-	0.00001
Fe	0.00089	0.0102	0.02324	-	0.0000018
Al	-	-	-	0.02712	0.00004
K	-	-	-	-	0.00001
Na	-	-	-	0.00665	0.0000007
Si	0.0039	-	0.10556	0.07577	0.00003
O	0.01814	0.1472	0.2905	0.21268	0.00201
Mineral Formulae	$(\text{Ca}_{0.47}\text{Mg}_{0.38}\text{Fe}_{0.49})\text{Fe}_{0.49}\text{Si}_{2.15}\text{O}_{10}(\text{OH})_8$	$\text{Fe}_{0.28}\text{O}_4$	$(\text{Ca}_{0.23}\text{Mg}_{0.19}\text{Fe}_{0.24})\text{Si}_{1.09}\text{O}_3$	$\text{Ca}_{0.62}\text{Na}_{0.25}\text{Al}_{1.02}\text{Si}_{2.85}\text{O}_8$	$\text{Si}_{0.042}\text{Al}_{0.05}\text{Fe}_{0.003}\text{Mg}_{0.019}\text{Ca}_{0.032}\text{Na}_{0.013}\text{K}_{0.008}\text{O}_{3.3}$
Mol. Fract.	0.03972	0.1575	0.4600	0.3387	0.00209

*Modal values given in volume percentages, - = very low values not considered in calculation.

**Figure 5a.** Time vs. saturation index (log Q/K) data plotted for plagioclase showing formation of calcite and aragonite.

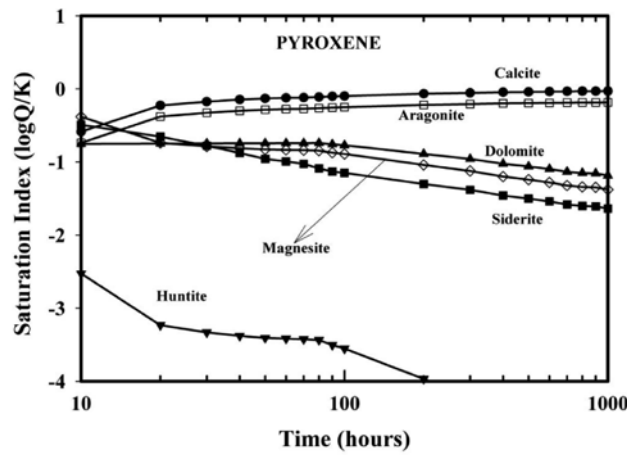


Figure 5b. Time vs. saturation index ($\log Q/K$) data plotted for pyroxene showing formation of calcite and aragonite, magnesite, siderite, dolomite and huntite

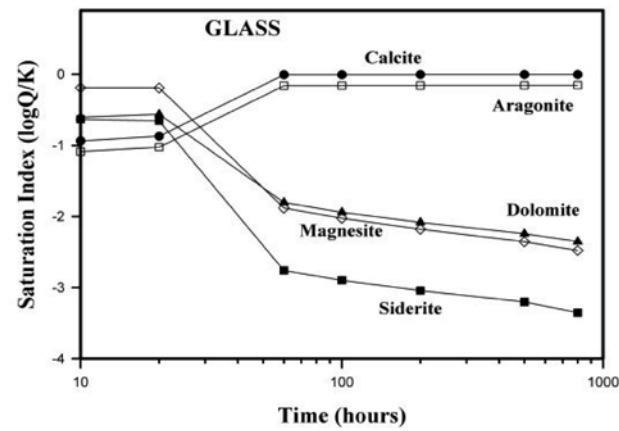


Figure 5c. Time vs. saturation index ($\log Q/K$) data plotted for glass showing formation of calcite and aragonite, magnesite, siderite and dolomite.

Table 5a Reaction progress for plagioclase (at 200°C temperature and 5 bar pressure)

Time (hours)	Aragonite ($\log Q/K$)	Calcite ($\log Q/K$)
10.00	-0.1567	-4.3500e-3
20.00	-0.1566	-4.2600e-3
30.00	-0.1565	-4.2000e-3
40.00	-0.1564	-4.0700e-3
50.00	-0.1563	-3.9500e-3
60.00	-0.1563	-3.9500e-3
70.00	-0.1562	-3.8700e-3
80.00	-0.1561	-3.8400e-3
100.00	-0.1560	-3.8000e-3
200.00	-0.1550	-3.5000e-3
300.00	-0.1545	-2.8400e-3
400.00	-0.1540	-2.5000e-3
500.00	-0.1530	-2.4000e-3
600.00	-0.1520	-1.4000e-3
700.00	-0.1488	-1.3000e-3
800.00	-0.1450	-1.2000e-3
900.00	-0.1410	-8.0000e-4

Table 5b. Reaction progress for pyroxene (at 200°C temperature and 5 bar pressure)

Time (hours)	Aragonite (log Q/K)	Calcite (log Q/K)	Dolomite (log Q/K)	Huntite (log Q/K)	Magnesite (log Q/K)	Siderite (log Q/K)
10.00	-0.7372	-0.5849	-0.7509	-2.5265	-0.3800	-0.4907
20.00	-0.3779	-0.2254	-0.7470	-3.2340	-0.7367	-0.6545
30.00	-0.3268	-0.1742	-0.7460	-3.3306	-0.7865	-0.7773
40.00	-0.2995	-0.1469	-0.7443	-3.3795	-0.8125	-0.8761
50.00	-0.2828	-0.1301	-0.7427	-3.4075	-0.8279	-0.9577
60.00	-0.2766	-0.1239	-0.7419	-3.4170	-0.8334	-0.9936
70.00	-0.2715	-0.1188	-0.7410	-3.4245	-0.8379	-1.0268
80.00	-0.2635	-0.1106	-0.7394	-3.4351	-0.8447	-1.0868
90.00	-0.2544	-0.1015	-0.7567	-3.5045	-0.8714	-1.1313
100.00	-0.2497	-0.0968	-0.7703	-3.5543	-0.8898	-1.1497
200.00	-0.2194	-0.0661	-0.8890	-3.9681	-1.0406	-1.3005
300.00	-0.2076	-0.0541	-0.9572	-4.1939	-1.1216	-1.3816
400.00	-0.1987	-0.0448	-1.0241	-4.4101	-1.1988	-1.4588
500.00	-0.1945	-0.0405	-1.0612	--	-1.2409	-1.5009
600.00	-0.1911	-0.0369	-1.0953	--	-1.2793	-1.5393
700.00	-0.1878	-0.0334	-1.1330	--	-1.3212	-1.5812
800.00	-0.1864	-0.0318	-1.1506	--	-1.3408	-1.6008
900.00	-0.1860	-0.0314	-1.1563	--	-1.3471	-1.6071
1000.00	-0.1843	-0.0296	-1.1838	--	-1.3774	-1.6374

- = data not reproducible

Table 5c. Reaction progress for glass (at 200°C temperature and 5 bar pressure).

Time (hours)	Aragonite (log Q/K)	Calcite (log Q/K)	Dolomite (log Q/K)	Magnesite (log Q/K)	Siderite (log Q/K)
10.00	-1.0879	-0.9356	-0.6042	-0.6317	-0.1868
20.00	-1.0223	-0.8700	-0.5614	-0.6545	-0.1914
60.00	-0.1589	-6.5300e-3	-1.8029	-2.7588	-1.8802
100.00	-0.1570	-4.6700e-3	-1.9399	-2.8974	-2.0254
200.00	-0.1557	-3.2800e-3	-2.0833	-3.0417	-2.1793
300.00	-0.1550	-3.2500e-3	-2.1500	-3.1000	-2.2512
400.00	-0.1548	-3.2418e-3	-2.2089	-3.1519	-2.3098
500.00	-0.1546	-2.2000e-3	-2.2425	-3.2013	-2.3541
800.00	-0.1541	-1.6600e-3	-2.3519	-3.3519	-2.4779

Table 6. Standard molar thermodynamic properties (Holland and Powell, 1998) for mineral and species considered in this study. $\Delta_f G$ is the Gibbs free energy of formation, $\Delta_f H$ enthalpy of formation, S is the entropy, V is the volume; a, b, c are the coefficients of the heat capacity.

Group	Mineral	$\Delta_f G$ KJ/mol	$\Delta_f H$ KJ/mol	S° J/mol/K	V° Cm ³ /mol	a J/mol/K	b J/mol/K	c J/mol/K
Carbonate	Calcite	-1128.81	-2161.51	-1027.74	36.89	140.9	0.5029	-9.50
	Dolomite	-1207.54	-2324.56	-1111.59	64.34	30.89	-0.4905	0.00
	Magnesite	92.50	156	65.10	28.03	186.4	0.00	0.00

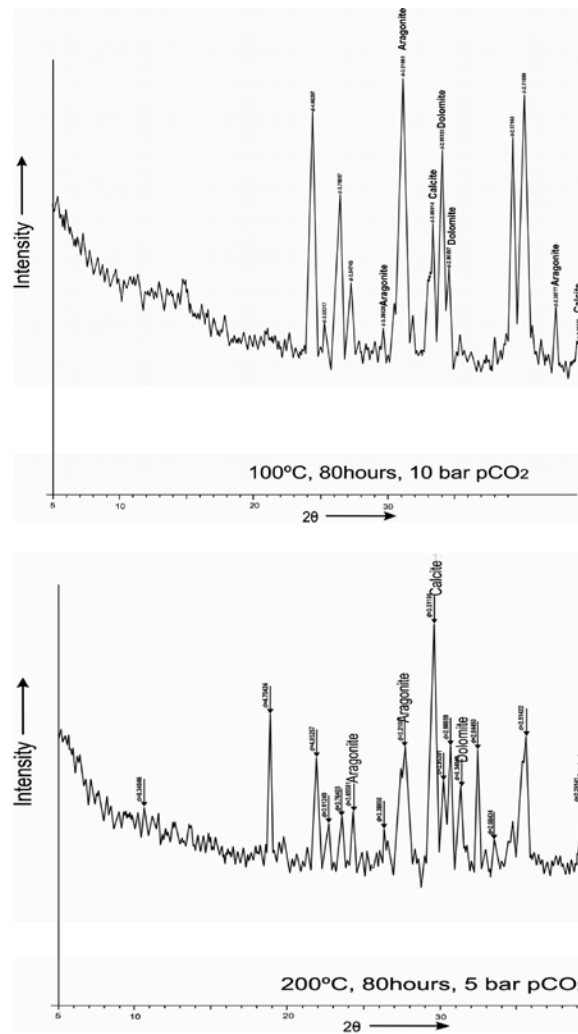


Figure 6. XRD patterns for basalt specimens treated at (a) 100°C temperature and 10 bar pressure and (b) 200°C temperature and 5 bar pressure showing appearance of calcite, aragonite and dolomite.

ultrasonic bath. The calculated (Oelkers and Gislason, 2001) geometric surface and measured BET surface area of a Deccan Traps basalt specimen is 250cm²/g and 23,000cm²/g, respectively. The sequestration experiments were carried out by putting 10 mg each of basalt in 100 ml of deionized water in a Parr Reactor for 50, 60, 70 and 80 hours at 100 and 200 °C temperatures, 5 and 10 bar pCO₂, respectively. The temperature was kept at constant at ± 100 °C and a stirring rate of 100 rpm was applied.

RESULTS AND DISCUSSION

After the experiments, chemical characterization and morphological studies on carbonated basalt specimens was carried out using X-ray Diffraction Analysis (XRD), Raman Spectroscopy and Fourier Transform Infrared Spectroscopy (FTIR) and Scanning Electron Microscope (SEM) mounted with EDS (for compositional determinations of secondary

carbonate minerals). The secondary mineralogy and basalt alteration were identified using X-ray powder diffractograms (XRD). The XRD patterns (Fig. 6) are dominated by the strong intensity peaks of calcite (90.6%), aragonite (75.9%), siderite (32.5%), and dolomite (42.4) at 100 °C and 200 °C for 60 and 80 hours. Calcite, magnesite, siderite, and aragonite have been observed at 100 °C and 5 bar pressure for 70 and 80 hours. Treated specimens were studied using Raman spectroscopic technique to ascertain presence of carbonate ions. It has been found that the characteristic peaks of carbonate ions are at 391.8, 509.7, 665.3 and 1009.6 (Fig. 7). Backscattered electron (BSE) images using SEM photomicrographs of the altered basalt acquired after sequestration experiments indicate that the role of time and temperature is significant. Secondary minerals were identified by SEM-EDS (Fig. 8) and their compositional data (Table 7a-b) obtained by EDS analysis. Based on these results mineral formulae were calculated, which show

Table 7a. Formation of carbonates with progression of time, SEM-EDS data (Specimen-1 treated for 80 hours at 100°C and 5 bars)

Oxides (wt. %)	
CO ₂	14.56
Na ₂ O	2.65
MgO	3.63
Al ₂ O ₃	9.80
SiO ₂	40.95
K ₂ O	0.76
CaO	2.20
TiO ₂	2.21
FeO	22.15
NiO	0.51

Mineral Formulae	
Calcite	Ca _{0.08} C _{0.69} O ₆
Aragonite	Ca _{0.43} C _{1.19} O ₉
Siderite	Fe _{0.64} C _{0.69} O ₆
Magnesite	Mg _{0.28} C _{0.69} O ₆

Table 7b. Formation of carbonates with progression of time, SEM-EDS data (Specimen-2 treated for 80 hours at 100°C and 5 bars)

Oxide (wt. %)	
CO ₂	17.5
Na ₂ O	2.31
MgO	2.39
Al ₂ O ₃	8.62
SiO ₂	43.38
K ₂ O	0.20
CaO	8.17
TiO ₂	1.15
FeO	15.29

Mineral Formulae	
Calcite	Ca _{0.29} C _{0.79} O ₆
Aragonite	Ca _{0.47} C _{1.19} O ₉
Siderite	Fe _{0.45} C _{0.79} O ₆
Magnesite	Mg _{0.11} C _{0.79} O ₆

presence of calcite (Ca_{0.29} C_{0.79} O₆), aragonite (Ca_{0.43} C_{1.19} O₉), siderite (Fe_{0.45} C_{0.79} O₆) and magnesite (Mg_{0.11} C_{0.79} O₆) in the treated samples, though, formation of secondary minerals consisting of clay minerals was not considered during mineral formulae calculations. Present experiments on basalt specimens under varied pressure, temperature and time limits have revealed formation of clay minerals. The clay minerals usually form in high proportions along with the formation of carbonates. Their proportion remains high in a few specimens, which

represent less quantity of carbonate minerals. Thus, it is possible that the clay minerals formed during carbonation process could have slowed down the process of carbonate formation as CO₃ ions trapped within the layered clay structures provide least opportunity for reaction with the freely available Ca, Mg or Fe cations. At 200 °C and 10 bar pressure experiments, the most common carbonate mineral formed is calcite, which suggests a fundamental change in the mobility of elements and secondary mineralogy with an increase of both temperature and pressure.

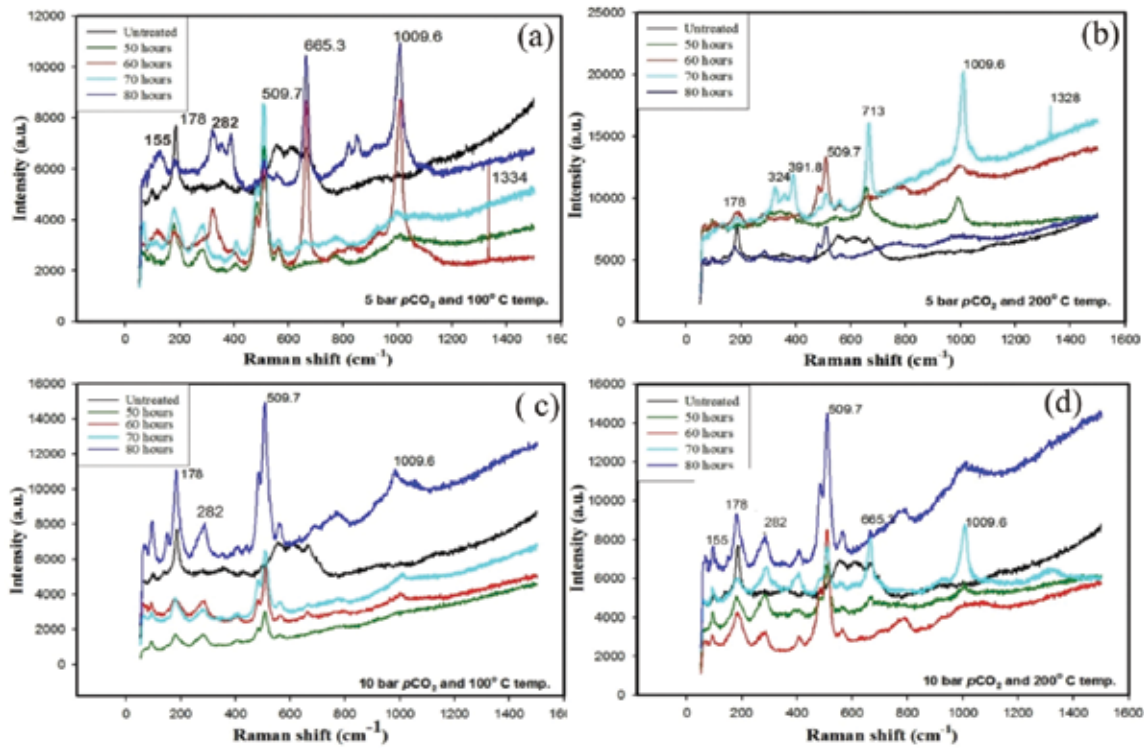


Figure 7. Raman spectra of basalt specimen treated at (a) 100°C temperature and 5 and 10 bar pressures, and (b) 200°C temperature and 5 and 10 bar pressures showing presence of carbonate peak at 391.8, 509.7, 665.3 and 1009.6 (cm⁻¹).

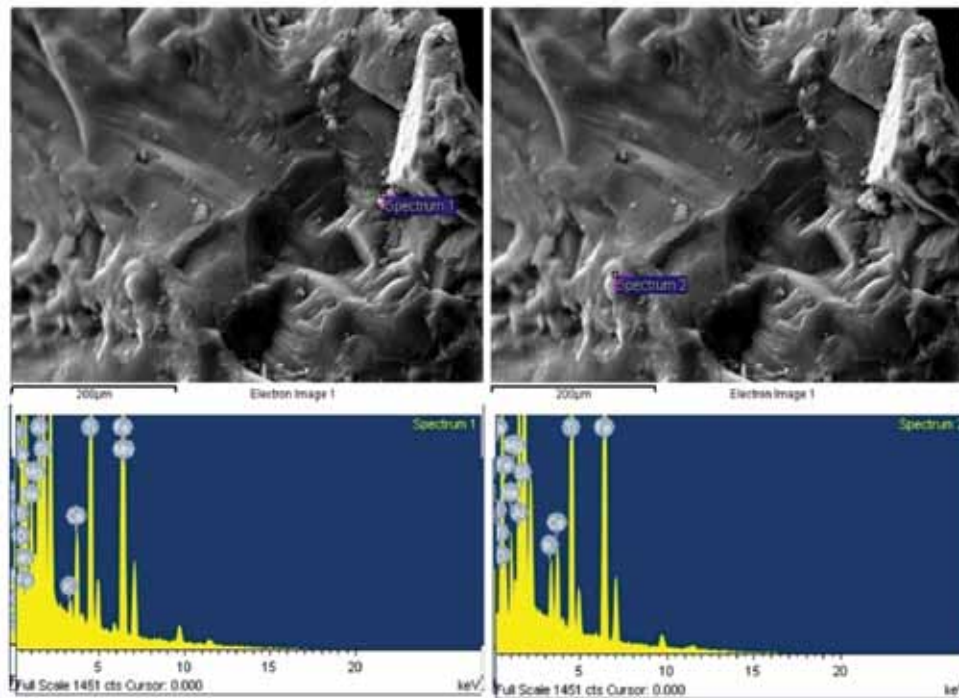


Figure 8. SEM-BSE images of the carbonate minerals formed after treatment of basaltic specimens treated at 100°C temperature and 10 bar pressure and (b) 200°C temperature and 5 bar pressure with their EDS spectra. Based on major oxide data (obtained after EDS analysis of the specimens) mineral formulae for carbonate minerals were calculated, showing presence of calcite, aragonite and dolomite.

CONCLUSION

The CO₂-water-basalt interaction is studied at 25 – 200 °C as a function of time to evaluate the effects of temperature and extent of reaction during basalt carbonation. Temperature and reaction time, affecting the reaction process is reflected in mineral assemblages, formed in compositional ranges of the basalt and secondary minerals. At 150 – 200 °C, calcite was the only carbonate formed and shows changes with temperature on elemental mobility and secondary mineralogy. Changes with temperature at elevated pCO₂ are reflected in dissolution rate and mechanism, at 200 °C. Dissolution rate is fast and primarily driven by secondary mineral replacement reactions and growth. Formation of the carbonate minerals - calcite, dolomite, aragonite, and siderite takes place largely due to the breakdown of pyroxene, feldspar and glass. However, the discrepancy is evident between thermodynamically calculated mineral species and the experimentally formed carbonate minerals.

ACKNOWLEDGEMENTS

For financial support towards this work, we acknowledge DST Project Grant [No. IS-STAC/CO₂-SR-79/10(G)]. Nishi Rani thanks CSIR for financial support in the form of Research Associateship.

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Monte – Carlo Simulations: Permeability Variation with Fractal and Pore Structural Parameters

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ABSTRACT

Often permeability is measured on rock samples in the laboratory or during well tests; however, such data is very sparse and takes a long time to generate. Permeability estimation from readily available porosity logs is an easier alternative approach. This paper proposes a new method of permeability estimation. The method relies on fractal theory and Monte – Carlo Simulations. Monte – Carlo simulations are performed to infer the dependence of permeability on pore size distribution. Unlike the empirical relations used to obtain permeability, present method generates random pore size distribution of the porous medium, which explains the fact that permeability depends on pore size distribution. Variation of permeability with pore-area fractal dimension, tortuosity fractal dimension and cluster mean size has been analyzed. The analysis infers that permeability is more sensitive to variations in cluster mean size compared to that of fractal dimensions. An attempt is made to define physical significance of fractal dimension of pore-area using fractal analysis of pore size distribution.

INTRODUCTION

Permeability is a key parameter for the production and management of hydrocarbon reservoirs as well as aquifers. Though the fluid content in the rock is proportional to porosity, the fluid flow depends on permeability, which is mostly predicted from effective porosity. However, even for a given effective porosity, permeability will be different for different rock types. This is because in addition to porosity other textural parameters like grain size, grain shape, sorting of grains also control permeability. These textural parameters affect tortuous nature of capillary pathways and arrangement of pores in the porous medium thereby making it complex to understand. In the past many authors related permeability to porosity in terms of empirical equations (Carman, 1956; Kozeny, 1927; Pape et al., 1999), which explain the results in terms of correlation coefficients. However, they do not explain the real physical situation existing in the porous medium. Fractal theory is applied to various complicated problems in earth sciences (Dimri, 2000b; Dimri, 2005; Dimri et al., 2005; Turcotte, 1997). Fractal theory is also found useful to explain complex nature of porous medium (Katz and Thompson, 1985; Dimri, 2000a; Dimri et al., 2012, Uma et al., 2014; Wheat craft and Tyler, 1988).

Monte – Carlo simulations for permeability of fractal porous media are formulated by Yu et al., 2005. According to them permeability is a function of cluster mean radius, relative particle size and fractal dimensions of pore area and tortuosity. In order to calculate permeability it is mandatory to select appropriate fractal dimensions of pore area and tortuous stream tubes and also cluster size, in the absence of measured information. In the present study permeability

variation with respect to these parameters is analyzed to understand its behavior with variation in these parameters.

THEORY

A porous medium consists of tortuous capillary pathways. The tortuous length of the capillary (L_t) in terms of pore diameter (λ) and the representative length (L_0) is given by Yu and Chen (2002).

$$L_t(\lambda) = \lambda^{1-D_T} L_0^{D_T} \dots\dots\dots (1)$$

Where, D_T is the tortuosity fractal dimension, with $1 < D_T < 2$, representing the extent of convolutedness of capillary pathways for fluid flow through a medium. Higher the value of D_T more will be the convolutedness or tortuosity. The limiting values of $D_T=1$ and $D_T=2$ correspond to a straight capillary and highly tortuous line that fills a plane (Wheatcraft and Tyler, 1988), respectively. The size distribution of pore diameters is another important property. Yu and Chen (2002) considered pores in a porous medium as analogous to the islands on earth or spots on engineering surfaces and obtained the following fractal scaling law.

$$N(L \geq \lambda) = \left(\frac{\lambda_{max}}{\lambda} \right)^{D_f} \dots\dots\dots (2)$$

where N is the number of pores with diameter (L) greater than or equal to λ , D_f the pore area dimension, with $1 < D_f < 2$, is the fractal dimension of the intersecting pore cross sections with a plane normal to the flow direction. Eq. (2) suggests that when λ corresponds to the maximum pore size, λ_{max} , the number of pores greater than or equal to it is one. Conversely, when λ corresponds to smallest

pore size, $\lambda_{\min}$, number of pores will be maximum and the scaling law becomes

$$N_t(L \geq \lambda_{\min}) = \left(\frac{\lambda_{\max}}{\lambda_{\min}}\right)^{D_f} \dots \dots \dots (3)$$

Where, N_t is the total number of pores.

Based on these fractal scaling laws Yu et al., (2005) obtained probability expressions for pore diameter (Eq. 4) and permeability (Eq. 5) and performed Monte - Carlo simulations to estimate permeability of bi-dispersed porous medium (Yu and Chen, 2002). Small particles form as clusters in bi-dispersed porous medium. The space between clusters get form macro pores and within the cluster micro pores exist.

$$\lambda_i = \left(\frac{\lambda_{\min}}{\lambda_{\max}}\right) \frac{\lambda_{\max}}{(1 - R_c)^{1/D_f}} \dots \dots \dots (4)$$

$$K = GA^{-\frac{(1+D_f)}{2}} \sum_{i=1}^J \lambda_i^{3+D_f} \dots \dots \dots (5)$$

Where

$$\left(\frac{\lambda_{\min}}{\lambda_{\max}}\right) = \frac{\sqrt{2}}{d^+} \sqrt{\frac{1 - \phi}{1 - \phi_c}} \dots \dots \dots (6)$$

$$\phi_c = 0.342\phi \dots \dots \dots (7)$$

$$\lambda_{\max} = \frac{\bar{R}_c}{2} \left[\sqrt{2 \left(\frac{1 - \phi_c}{1 - \phi} - 1 \right)} + \sqrt{\left(\frac{2\pi}{\sqrt{3}} \right) \left(\frac{1 - \phi_c}{1 - \phi} \right)} - 2 \right] \dots \dots \dots (8)$$

Where

The approximated pore area of the unit cell is given by

$$A_p = \sum_{i=1}^J \pi \lambda_i^2 / 4 \dots \dots \dots (9)$$

- $\lambda_{\min}$ - Minimum pore diameter
- $\lambda_{\max}$ - Maximum pore diameter
- ϕ - Effective Porosity
- ϕ_c - Micro porosity in the cluster
- R_c - Cluster mean radius
- d^+ - Ratio of cluster mean size and the minimum particle size
- λ_i - Diameter of the i^{th} capillary tube chosen by Monte - Carlo simulations
- A_p - Total pore area of a unit cell
- R - A random number that lies between 0 and 1
- A - Total cross sectional area of a unit cell

Since there are N_t pores in a unit cell, J in Eqs. (5) and (9) will be equal to N_t . In Eq. (3) $\lambda_{\max}/\lambda_{\min}$ is greater than one, thus total number of pores (N_t), increases with D_f . Readers may refer to Yu et al. (2005) and Uma et al. (2014) for detailed description of algorithm for determination of permeability using Monte - Carlo simulations.

RESULTS AND DISCUSSIONS

Permeability variations with tortuosity fractal dimension (D_T), pore area fractal dimension (D_f) and cluster mean radius (R_c)

Monte - Carlo simulations are run for different values of D_T by considering $R_c=0.3$ mm, $d^+=24$, $D_f=1.8$ (Yu et al., 2005). It is observed that permeability decreases with increase of D_T (Fig. 1(a)). As D_T increases from 1.1 to 1.3, the mean value of percentage reduction in permeability is 17. When all other parameters $R_c=0.3$ mm, $d^+=24$, $D_f=1.1$ are kept constant permeability decreases with increasing D_f (Fig. 1(b)). The mean value of percentage reduction for increase of D_f from 1.4 to 1.6 is 23. A small variation in Cluster mean radius (R_c) from 0.2 mm to 0.3 mm increases permeability by 53% (Fig. 1(c)). This indicates that permeability is more sensitive to Cluster mean radius (R_c) than to fractal dimensions.

DISCUSSIONS

The pore size distributions generated by Monte - Carlo simulations for a constant porosity with four different values of D_f are shown in Fig. 2(a) to 2(d). It is clear from these figures that with increase in D_f total number of pores are increased and their diameters are reduced. As we can see in Table1, when D_f increases the percentage number of pores having diameter less than 40 μm increases and those having diameter greater than 120 μm decreases. This analysis indicates that increased value of D_f increases the number of smaller scale/tiny pores, which increases flow resistance. The observation that permeability decreases with increase of D_f is supported by Pitchumani and Ramakrishnan (1999) and contradicts the results of Yu and Cheng (2002) in which an analytical expression is used to calculate permeability. Yu and Cheng (2002) state that permeability increases with increasing D_f because as D_f is increased, the total porosity or the pore area increases. For a given porosity with increase in D_f , increased number of pores leads to increase in simulated pore area (A_p) (Table 1). However, permeability decreases because the sum $\sum_{i=1}^J \lambda_i^{3+D_T}$ decreases. Smaller values of λ_i become negligible when they are raised to the power of $3+D_T$ and cannot contribute to the sum $\sum_{i=1}^J \lambda_i^{3+D_T}$. The variation of this sum is controlled by maximum value of simulated pore diameter, which again decreases with D_f (Table 1). The contribution of $\lambda_{\max}$ to total permeability ranges between 46% and 54% (Table 1). This in turn indicates that permeability is very sensitive to pore diameter. According to Yu and Cheng (2002), $\lambda_{\max}$ does not vary with D_f ; it is calculated from analytical formula (Eq. (8)) and remains constant for a given porosity and particle size. This is the difference between the approach based

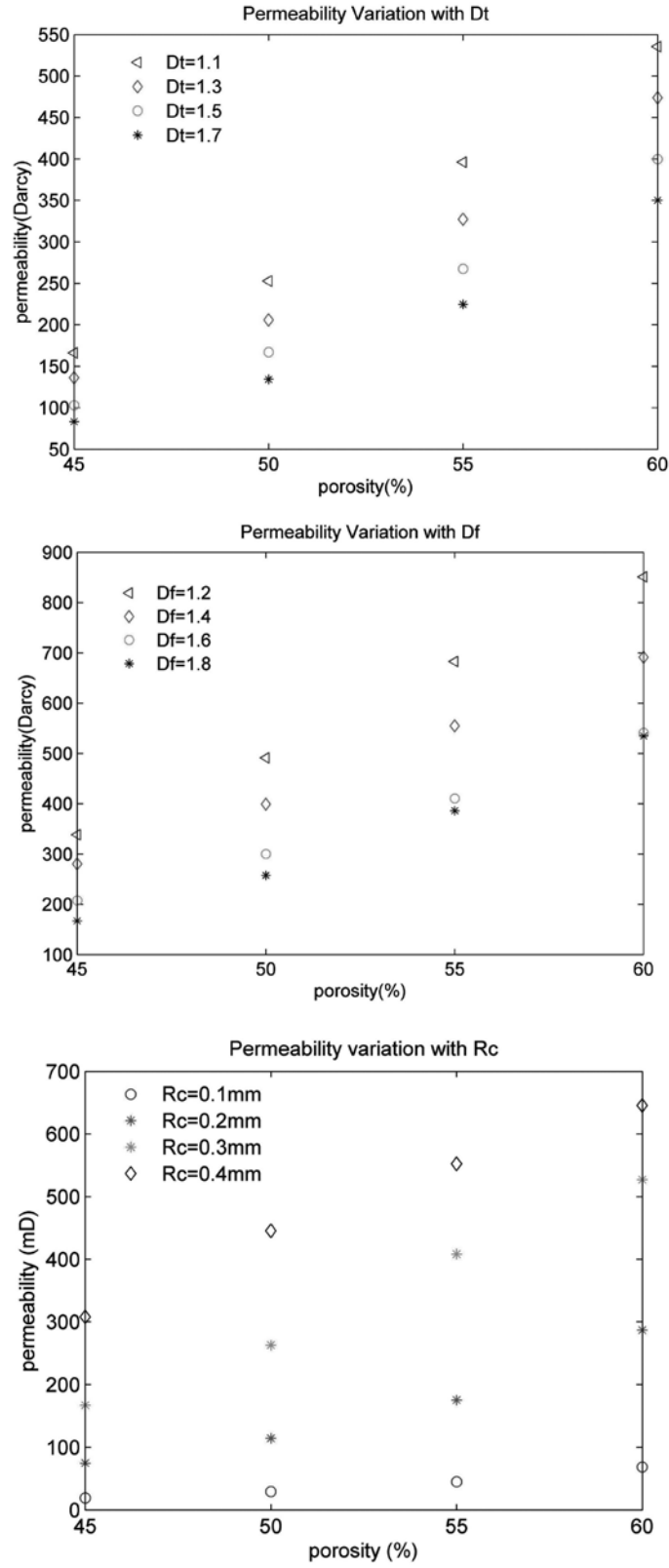


Figure 1. Permeability variations with tortuosity fractal dimension (D_T), pore area fractal dimension (D_f) and Cluster mean radius (R_c). Permeability decreases with increasing D_T , D_f and it increases with increase of R_c . The variation in permeability is more sensitive to changes in Cluster mean radius and it increases with porosity.

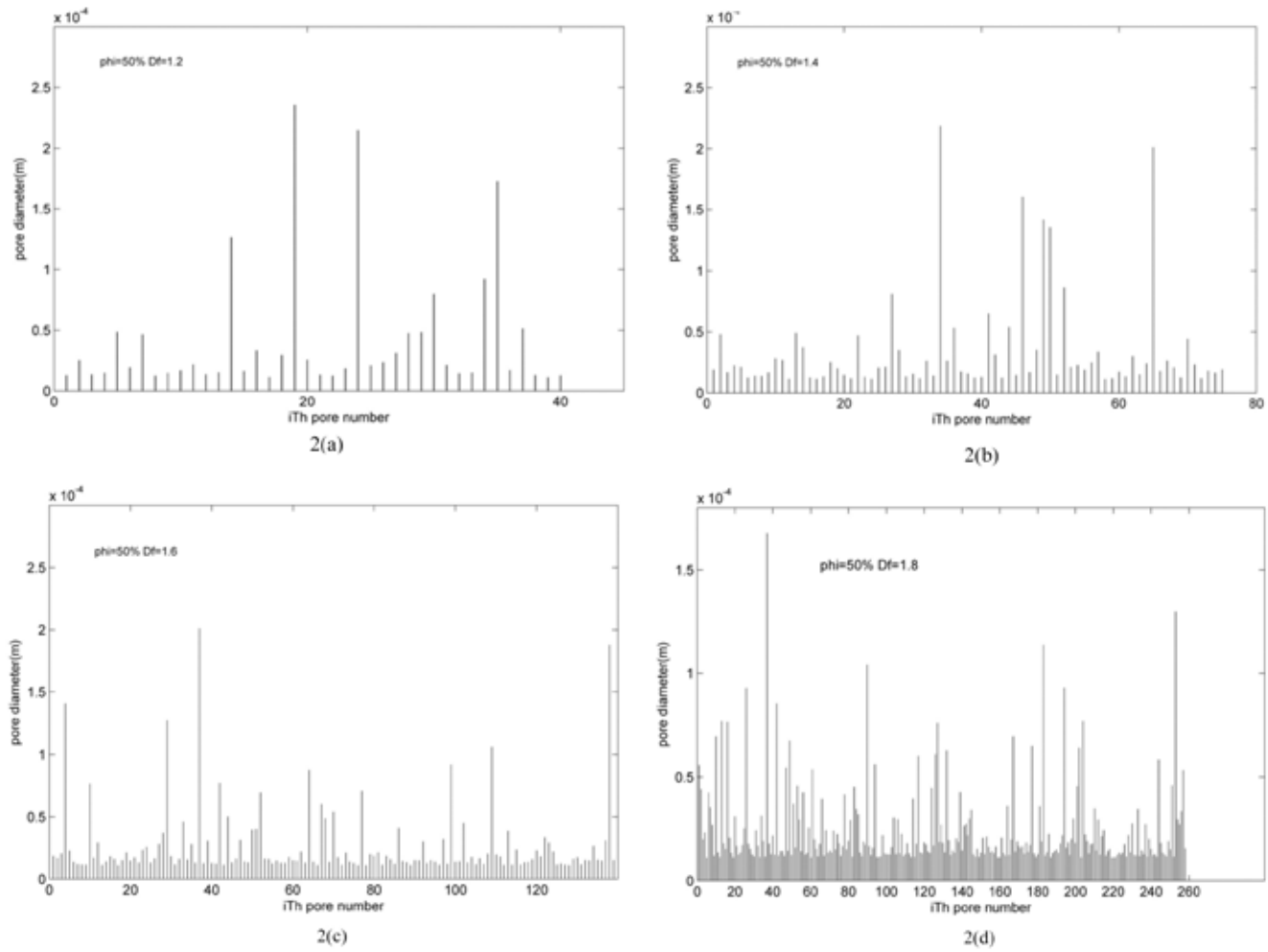


Figure 2. Fig. 2(a), 2(b), 2(c) and 2(d) illustrate the pore size distribution for a porosity of 50% and for $D_f=1.2$, $D_f=1.4$, $D_f=1.6$ and $D_f=1.8$, respectively. The random pore size distribution chosen by Monte – Carlo simulations replicates the irregular pore size distribution in porous media, provided porosity constant, with increase in D_f total number of pores are increased and their diameters are reduced.

on analytical formula (Yu and Cheng 2002) and Monte - Carlo simulations (Yu et al., 2005). The negligible effect of smaller values of λ_i which represent tiny pores, towards the sum $\sum_{i=1}^I \lambda_i^{3+D_T}$ can be explained in terms of real physical scenario. The power $3+D_T$ takes care of tortuosity. Smaller scale tortuous capillaries cannot transmit fluid and their contribution to permeability is negligible, however, they can contribute to pore area and increase it. The existing analysis of Yu and Lee (2000) also showed that the smaller scale/tiny pores in porous media have impact negligible effect on total permeability. The analysis concludes that $D_f=1$ corresponds to flow through the unit cell with a single capillary tube and $D_f=2$ corresponds to a compact cluster with infinite number of tiny pores, which cannot permeate fluid.

An increment in the value of D_T makes capillary pathways more twisted, thus increases resistance for flow and decreases permeability. As cluster mean radius

increases capillary pathways become less tortuous and wider, so permeability increases.

CONCLUSIONS

Permeability is very sensitive to variations in cluster mean size. With increase of pore area fractal dimension (D_f) permeability decreases because the number of smaller scale/tiny pores increase and thus flow resistance also increases. Consideration of this kind of complex real physical situation that exists in porous medium indicates superiority of the method over empirical relations. The observation that permeability decreases with D_f will help to select D_f value during permeability estimation. If the fractal dimensions and grain size of rock samples are known, permeability estimated by this method can be used as first-hand information even though well test data or laboratory measurements are not available.

Table 1. Variation in pore size distribution and permeability with increasing pore area fractal dimension (D_f) for a constant porosity of 50%, $D_T=1.1$ & $R_c=0.3$ mm.

D_f	Simulated $\lambda_{\max}$ (μm)	Simulated $\lambda_{\min}$ (μm)	% No. of Pores having $\lambda < 40 \mu\text{m}$	% No. of Pores having $\lambda > 120 \mu\text{m}$	$\sum_{i=1}^{N_i} \lambda_i^{3+D_f} * 10^{-16}$	Ap (m^2) * 10^{-6}	Permeability only due to $\lambda_{\max}$ (D)	Total Permeability (K) (D)	Percentage contribution of $\lambda_{\max}$ to total permeability (%)
1.2	218	11	82.5	12.5	21.5	0.270	228	490	46.5
1.4	209	11	84.0	6.7	17.7	0.282	196	404	48.5
1.6	191	11	86	0.7	17.7	0.300	158	290	54.4
1.8	177	11	88	0.38	11.2	0.376	124	256	48.4

ACKNOWLEDGEMENTS

The author would like to acknowledge the Royal Norwegian Embassy, New Delhi for financial support through the Indo-Norwegian research project. The author thanks Director, NGRI for his kind support to carry out this work. Author is thankful to Prof. V.P. Dimri, Dr. Ravi Prakash Srivastava and all other members of Indo-Norwegian research project for their useful suggestions and encouragement. Thanks are due to CSIR for providing research internship.

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Emerging Considerations about the Sustainability of CCS for EOR and other Downstream Applications: A Policy Perspective

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ABSTRACT

The paper briefly states important policy related aspects such as enabling access to CO₂ on a sustained basis supported with appropriate infrastructure. These aspects cannot be overlooked if the benefits of carbon capture, storage and utilization are to be up-scaled over time. This will also create an incentive for high emitters of CO₂ to enable mitigation, with cross – sectorial implications for integrated management of emissions.

INTRODUCTION

The importance of carbon capture and storage cannot be over emphasised, particularly as a mitigation measure with respect to climate change impact management. Production of power and use across several sectors entails the release of carbon-dioxide with significant impacts on climate systems. While the capture of CO₂ by itself has enormous value for mitigation, its subsequent use to enhance oil recovery from wells adds value to these mitigation measures. Recent papers by Hill et al., (2013) and Hidug (2012) highlight the twin advantages of use of captured CO₂ to enhance oil recovery and further explore the fate of CO₂ remaining within the reservoirs. The latter could generate additional mitigation benefits.

The present paper takes stock of some of the recent considerations in this context through a policy perspective especially at a juncture when India and countries with similar circumstances of growth may like to consider enhancing CO₂ capture options. It is well known that some portions of CO₂ injected in to the wells also return to the surface with oil production and can be separated and re-injected as part of a reuse strategy.

POLICY LINKAGES IN THE SEQUESTRATION AND USE INTERFACE

Several major policy highlights have been made by the Advanced Resources International, Inc. (2010) in the context of the above stated. It has been emphasized that the CO₂ storage in oil reservoirs and CO₂ enhanced oil recovery provide critical solutions for reducing GHG emissions. This is based on the experiences in commercial applications for nearly three decades: albeit, at a scale of some consequence predominantly in the United States. This has also helped establish extensive networks of CO₂ pipelines, duly integrated with anthropogenic sources of CO₂. This is however, constrained by not – so - easy availability of CO₂ of reliable quality and quantity. Enhancing access to CO₂ calls

for appropriate infrastructure with related incentives for capturing CO₂. Interestingly oil producers could be expected to pay for the CO₂ because it can be used to recover oil. Carbon capture and storage mechanisms therefore become important to ensure adequate supply of CO₂. These multiple benefits however call for well integrated regulatory fiscal and non-fiscal measures supported by institutional mechanisms.

A workshop organised jointly by the International Energy Agency and the Organisation of Petroleum Exporting Countries on CO₂-Enhanced Oil Recovery with Carbon Capture and Storage emphasised the importance of CO₂-EOR for the longevity of hydrocarbon assets. These could however be specific to regions and circumstances of extraction. Policy makers may like to take note of the fact that CO₂-EOR with carbon capture and storage is seen as an early stage mitigation option for duly capturing CO₂ from multiple and large point sources. It will therefore be useful to dovetail CO₂ capture and reuse with involvement of large scale CO₂ generators.

The workshop further discussed the dynamics of the behaviour of oil fields with respect to enhancing the storage of CO₂ vis-à-vis reservoir pressure, fracturing, etc. It is therefore important to understand the qualitative and quantitative aspects of storage and intended use before technological measures supported with policies are mainstreamed. Typically environmental management policies, programs and projects have to support well-designed monitoring, reporting and verification tools to ensure success. Appropriate capacity building of personnel to sustain transitions is equally crucial.

Godec (2012) at the above stated workshop deliberated on the occurrence and distribution of CO₂ during EOR with suggestions for increasing CO₂ storage. This is also aligned with a lesser CO₂ foot print of oil extraction processes. However, some of the significant challenges as highlighted by Godec pertain to balancing a requirement of CO₂ with steady supply and regulations for long term monitoring, and setting of pipelines. This particular aspect

is important especially for legislation on environmental impact, assessment and establishing trade offs for larger scale community benefits. This is further substantiated by the National Enhanced Oil Recovery Initiative (2012); that established a coalition of stakeholders from various sectors related to energy including oil and gas industries regulators and decision makers, signifying synergies across sectors. The initiative recommends incentives for capture and infrastructure projects including pipelines, probably with the assumption that this will promote mitigation benefits. This is further expected to enhance technological innovation for CO₂ capture and its distribution across large areas.

The initiative further draws attention to the possibility that anthropogenic sources of CO₂ may have to be given greater attention for capture compared to natural sources. The Petroleum Technology Transfer Council (2012) links the occurrence of mature wells that are not responsive any longer to primary and secondary methods of oil recovery as useful sites for EOR. The Council has highlighted the fact that the USDOE has transformed the concept of CCS to Carbon Capture, Utilization and Storage (CCUS) emphasising the utilisation and storage aspects. Eric (2012) reiterates the above stated and cites the case of the Texas Clean Energy Project that has delivered multiple benefits across sectors. This is particularly relevant for the power and oil sectors.

In a recent paper Eide et al., (2013) argues that adequate policy attention has not been conferred on CO₂ markets and therefore may affect the development of CCS on commercial scale. Policy makers are however urged to take note of the implications of utilisation and storage aspects before launching on integrated measures for mitigation and other benefits. This is further reinforced by the Centre for Environment Friendly Energy Research Newsletter (2012). This appears to have expanding recognition of CO₂ storage spaces with greater exploration potential in several parts of the world.

The special case of India on the prospects of carbon capture and storage technologies has been presented by the Wuppertal Institute for Climate, Environment and Energy with GIZ (2012). The report deals with the status of CCS in India, the potential storage in India, pathways for India's power sector and other industries to relate to CCS development, matching supply of CO₂ with the storage capacities, economic assessments of storage and lifecycle aspects of storage with respect to environmental impacts assessments for the coal sector in particular. This report was aligned with insights from China and South Africa; setting the context for a detailed policy framework.

Based on the above, it is interesting to note that CCS with special reference to utilisation for EOR is an evolving area with significant scope for embedding preventive approaches. Such approaches have to be specially aligned

with the structural integrity of reservoirs at their end use followed by detailed assessments of the sources of CO₂ and sustained access supported with adequate infrastructure.

CONCLUSION

India in particular has demonstrated its ability to be resilient even as significant chaos was seen across the globe with economic downturns. The opportunity to capture CO₂ and enhance oil recovery is indeed quite exciting because it can complement economic benefits with environmental protection gains. Policy measures that trigger and foster innovations in this dynamic area of technologies and tools including monitoring and verification, should also help leap frog into newer regimes of technologies supported by robust management information support systems. These systems should establish linkages between emissions and impacts avoided by gains that will be achieved through EOR; further quantifying environmental impacts that will be generated through the use of such recovered energy substrates. The period of time that could be allocated for extended energy use should also be spent on identifying and mainstreaming additional alternatives and enabling transitions to an environmentally and economically efficient energy future.

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